

Recent Advances in Bio-electrochemical System Analysis in Biorefineries

**Samarjeet Singh Siwal^{a,b}, Qibo Zhang^{a,c*}, Adesh Kumar Saini^d, Vijai Kumar Gupta^e,
Dave Roberts^f, Vipin Saini^g, Frederic Coulon^h, Bhawna Pareek^b and Vijay Kumar
Thakur^{e,i*}**

^a Key Laboratory of Ionic Liquids Metallurgy, Faculty of Metallurgical and Energy Engineering, Kunming University of Science and Technology, Kunming, 650093, P.R. China

^b Department of Chemistry, M.M. Engineering College, Maharishi Markandeshwar (Deemed to be University), Mullana-Ambala, Haryana, 133207, India

^c State Key Laboratory of Complex Nonferrous Metal Resources Cleaning Utilization in Yunnan Province, Kunming 650093, P.R. China

^d Department of Biotechnology, Maharishi Markandeshwar (Deemed to be University), Mullana-Ambala, Haryana, 133207, India

^e Biorefining and Advanced Materials Research Center, Scotland's Rural College (SRUC), 10 Kings Buildings, West Mains Road, Edinburgh EH9 3JG, UK

^f Agriculture and Business Management Department, SRUC Barony Campus Parkgate Dumfries DG1 3NE

^g Department of Pharmacy, Maharishi Markandeshwar University, Kumarhatti, Solan, Himachal Pradesh, 173229, India

^h School of Water, Energy and Environment, Cranfield University, Cranfield, MK430AL, United Kingdom

ⁱ Department of Mechanical Engineering, School of Engineering, Shiv Nadar University, Uttar Pradesh 201314, India

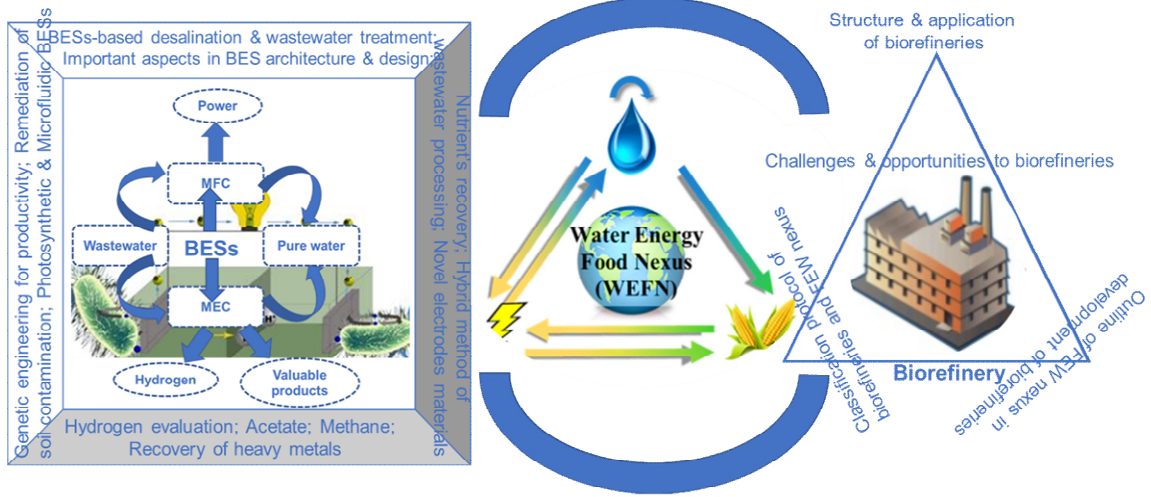
*Corresponding authors: Vijay.Thakur@sruc.ac.uk (Vijay Kumar Thakur), qibozhang@kust.edu.cn (Qibo Zhang)

Abstract:

Concerns around acquiring the appropriate resources toward a growing world population have emphasized the significance of crucial connections between food, energy, and water devices, as described within the food-energy-water nexus theory. Advanced biorefineries provide second-generation biofuels and added-value chemicals through food products have affected these nexus sources. We combine various conversion technologies and expected options to look further for cost-effective technologies that maximize the value of resource use and reuse and minimize the amount of resource needed and environmental impacts. In this review article, our central focus is on structure and application, the outline of food-energy-water (FEW) nexus in biorefineries and bio-electrochemical system (BES) and looking into the energy-efficient and value-added product recovery. In addition, based on BES analysis for energy efficiency and valuable product recoveries such as hydrogen evaluation, acetate, recovery of heavy metals, nutrient's recovery has been discussed under this article. Additionally, we focused on wastewater processing methods, novel electrode materials used in BES, BESs-based desalination and wastewater treatment, recent BES architecture and designs, genetic engineering for enhanced productivity, and valuable materials production surfactants and hydrogen peroxide. Finally, we concluded the topic by discussing the remediation of soil contamination, photosynthetic & microfluidic BES systems, possibilities of employing CO₂, including prospects and challenges.

Keywords: Bio-electrochemical system; biorefinery; microbial fuel cells; environmental impacts; value-added products.

Graphical abstract:



1. Introduction:

Bioelectrochemical systems (BES) have been developed as niche expertise toward converting scrap rivulets and algae within useful power. Microbial fuel cells (MFCs) and microbial electrolysis cells (MECs) are two kinds of BES that generate power and H₂ sequentially by utilizing the organic matter under aqueous rivulets [1]. A few biomass components that make it suitable raw material, from renewable (low-carbon) chemical and energy generation, are recyclable CO₂, garbage valorization, and its usage as an available renewable energy source [3-6]. The evolution of distinct approaches, which conferred similar biorefineries, has been done in biorefineries' initial configuration. An additional component causing the development in bio- and solar plants commonly used and valorizing the CO₂ generated within biorefineries is to combine sustainable energy [7]. Zhang *et al.* [8] proposed biorefineries-based photosynthesis to generate vast volumes of compounds through H₂/power and CO₂ for next-generation.

A biorefinery is the ease of solid fuel's maintainable transformation within combined, practical and straightforward processing in varied commodities (such as chemicals, feed, pet supplies and energy goods) [9,10]. According to the types of raw material used (i.e., algae, organic waste, and lignocellulose), the conversion technology (biological, thermochemical) can be grouped into biorefineries. Different stages of complexity in biorefineries are Class I (different raw material single output), Class II (solitary raw material several outputs) and Class III (various raw materials including different outcomes). The food-energy-water (FEW) nexus approaches have not yet recognized the biorefinery model's synergies during the assessment of biorefinery operation for the different classes. The advanced biorefineries (Classes II and III) have encouraged the possibilities to discuss connection problems with stakeholders producing food, energy, and clean water sustainably while protecting the climate and ecosystems [11-14].

The energy recovery from a biochemical conversion method to originating liquid fuels, for example, ethanol of lignocellulosic solid fuel under a standard procedure, is approximately 53.5% [15]. The biomass's residual energy is carried out towards the atmosphere within the class of small-scale garbage heat, finishing below a decay of about 46.5% from the received biomass energy. The standard systems receive energy from

remaining organics, including byproducts, through flaming under an evaporator or separate them by wastewater processing. The wastewater treatment plant can produce biogas applied to fuel towards a thermochemical oil generation component. Besides the energy waste, a notable quantity of water has been allowed through dissipation and chimney gas discharge facing the atmosphere. Current literature has shown the significance of water recycling within a standard biorefinery method, especially during biochemical conversion [17-19].

As we know, MFCs and MECs can be employed in polluted water while producing valuable energy products such as electricity or H₂ [1,22-25]. The employment of these machinery has studied biodiesel production and metropolitan waste-water processing within food production [26]. Common contaminants in the emerging biorefinery of the water system involve sugar- and lignin-degeneration goods, acetate, fermentation derivatives, and remaining carbohydrates. The contaminants' transformation includes furfural, hydroxymethyl furfural, acetate, hydroxybenzaldehyde, hydroxy acetophenone, and vanillic acid toward electricity using MFCs technique [27]. Renovation of different sugars and starches toward power generation has been described by employing MFCs [28,29]. This initiates opportunities to construct method choices applying MFCs to improve energy by remaining organics in the biorefinery by allowing water recycling.

The continuing estimation of water and energy synergy under the food operation began during the 1980s. In the previous years, the essential breakthroughs impacting water-related energy (WrE) estimate within the food policy [30].

The various developments in MECs and MFCs have collected together and shown for energy production in **table 1**.

Table 1. Different types of integration developments of microbial electrolysis cells (MEC) and different types of machinery for energy production.

S. No.	Integration type	Electrode materials (Anode & Cathode)	Applied voltage	Electrolyte/Substrate	Current/power density	References
1	MECs	Graphite brush & Carbon cloth (CC)	0.8 V	Waste activated sludge	-	[31]
2	Constructed wetland (CW)-MFCs	Stainless steel (SS) mesh & charcoal	0.44 V	Swine and synthetic wastewater	0.07 A/m ³	[32]
3	MEC with MFC	Carbon brush & Silicon nanowires	-	Growth media	0.68 A/m ²	[33]
4	CW-MFCs	Graphite granules & graphite plate	0.08 V	Synthetic wastewater	0.16 A/m ³	[34]
5	MECs	Pt-catalyst & CC	0.8 V	Domestic wastewater	189 A/m ³	[35]

6	MEC with dye-sensitized solar cell	Carbon felt (CF) & Plain graphite	0.7 V	Growth media	-	[36]
7	MECs	Carbon fibers & CC	0.8 V	Domestic wastewater	158 A/m ³	[21]
8	MEC with MFC	Carbon paper & Pt. coated cathode paper	0.8V	Growth media	0.25 mA	[37]
9	MEC with hydrogen bioreactor (HBR)	Graphite plate	0.2-1.0 V	HBR effluent	-	[38]
10	MECs	Graphite fiber brush & CC holding Pt catalyst	0.5 V	Wastewater	1.15 A/m ³	[39]
11	MEC with forward osmosis (FO)	Carbon brush & Pt. incorporated CC	0.6-1.0 V	Synthetic wastewater	4.5 mA	[40]

12	MEC with FO	CF fixed upon perforated SS plate using Ag paste & Perforated titanium plate	0.7 V	Synthetic wastewater	3.34 A/m ²	[41]
13	MEC with anaerobic digester (AD)	CF & SS	0.4 V	AD effluent	1.3 mA	[42]
14	MEC with AD	Carbon brush & SS mesh	0.8 V	Sludge fermentation liquid	9.6 mA	[43]
15	MEC with AD	Carbon brush & Ti/RuO ₂	0.8 V	Food and sewage sludge in changing ratios	Steady-state with applied voltage	[44]
16	MEC with AD	CF & SS mesh	0 mV	Raw pig slurry	2.01±0.63 A/m ²	[45]
17	MEC with simultaneous desalination	CC & Pt. coated CC	0.55 V	Phosphate buffer solution (PBS)	1.4 A/m ²	[46]

18	MECs	Carbon paper & Carbon paper/Pt	0.5 V	Domestic wastewater	0.6 A/m ²	[47]
19	Dark fermentor (DrF)-MEC MFC	Carbon brush & Pt incorporated CC	0.33 to 0.47 V	DrF effluent	52 A/m ³	[48]
20	MEC with DrF	Carbon fibres with SS mesh & SS mesh	-0.4 V	Sugar beet extract in the diverse substrate to inoculum proportions	3.6 A/m ²	[49]
21	MECs	Graphite fiber brush & CC/Pt	0.5 V	Swine wastewater	106 A/m ³	[50]
22	MEC with DSSC	CF & Pt.-coated titanium plate	0.7 V	Growth media	0.30 A/m ²	[51]
23	MEC with HBR with spent wash effluent	Graphite plate	0.2- 0.6 V	HBR effluent	-	[52]
24	MECs	Graphite fiber brush	0.7 V	Industrial and food processing wastewater	2.1 A/m ²	[53]

		& CC/Pt				
25	MEC with upflow anaerobic sludge bed (UASB) reactor	Ti mesh with Ir-MMO coating & SS mesh	140-260 mW/LR	UASB effluent	-	[54]
26	MEC (multi anode) with DrF	Graphite felt & CC	0.8 V	Dark fermentation effluent	298 A/m ³	[55]
27	MECs	Graphite fiber brush & CC/Pt	0.9 V	Potato wastewater	-	[56]
28	MEC with biomass & pyrolysis effluent	CF & Pt-incorporated CC	0.96 V	Bio-oil aqueous phase	202 A/m ³	[20]
29	MECs	Graphite felt & Ni-based gas diffusion	1.0 V	Synthetic fermentation effluent	206 A/m ³	[57]

30	MEC with electrical voltage from the thermoelectric micro converter	Plain CF & Carbon paper with Pt	0.17 to 0.83 V	Acetate	0.28 to 1.10 A/m ²	[58]
31	MECs	Graphite fiber brush & SS mesh	0.9 V	Winery wastewater	7.4 A/m ³	[59]
32	MECs	Graphite fiber brush & Graphite fiber cloth/Pt	0.8 V	Milk, glycerol, starch	150 A/m ³	[60]

Hence, there is an urgent need for upgrading the BES for energy efficiency and value-added product recovery [61]. Researchers worldwide are actively striving to optimize the method, and various substitute biological systems have been examined [62]. Various kinds of BES have been used as energy recovery or energy-efficient systems to use high-strength wastewater. It is challenging to use low-energy results, such as contaminated water and food/feed products. In this review article, our primary focus is on the principal technical and non-technical challenges and opportunities, classification protocol of biorefineries and bio-electrochemical systems (BES) and looking into the energy-efficient and value-added product recovery. This article describes novel cathode and anode materials and catalysts, recent BES architecture and designs, genetic engineering for enhanced productivity, production of valuable materials including surfactants and hydrogen peroxide, simultaneous desalination and wastewater treatment, remediation of soil contamination, photosynthetic BES systems, microfluidic BES, an alternate approach integrating BES within biorefinery designs to enhance energy effectiveness, nutrients recovery, wastewater processing, possibilities of employing CO₂ and prospects for technical assistance with the challenges, future prospects with concluding remarks.

2. Structure and applications, the outline of FEW nexus of biorefinery and bio-electrochemical system

2.1. Structure and applications of biorefineries

A ‘biorefinery’ is a theoretical design towards a prospective biofuel generation system comprising fuels and valuable substances. The biorefinery concept strives to use solid fuel transformation, which has been used to purify fuel. Bioindustries would instantaneously generate biofuels along with bio-based substances, heat, and energy. Biorefineries would perform a cost-effective alternative at which bio-based compounds are sided commodities concerning liquid combustible [63,64]. Imminent biorefineries would be suitable to imitate the energy performance of new oil purifying by massive heat mixture and side creation improvement. As per the National Renewable Energy Laboratory (NERL), a biorefinery combines biomass regeneration methods and devices to generate energy, substances, and fuels. The international energy agency (IEA) bioenergy Task 42

has described a biorefinery as “the sustainable processing of biomass within a spectrum of commercial commodities and energy”. Therefore, a biorefinery may be a convenient method for manufacturing or many equipment types to transform biomass [65].

2.1.1. Challenges and opportunities to biorefineries

There are several scientific and non-scientific holes in biorefinery that should be affected by industrialization. Current technological limitations, including energy yields, are connected to by-product, price, and issues during accumulation and stocking the substance produced. This is mainly for yearly or different crops that need to be accumulated quickly during the autumn. Several scientific non-technical challenges correlated with increasing energy crops involve nutrients plus the power of insects and sickness [66,67].

The main non-technical challenges are the applications upon the worth of soil (food, energy, home usage, business, classifications being fields of natural elegance, particular scientific concern), along with the ecological impacts of vast regions of monoculture [68,69]. On the other side, the extensive application of vegetable oils can create additional critical challenges and need in developing nations.

There are some scientific limitations in making biomass-ignited plants in a specific order. The volume and producing ability of biomass factories are significantly lower than the current natural-gas-fired turbine operations. On the residential scale, the output is about 50 MW; beyond that point, the accessibility and price of producing fuel become important [70]. The main non-practical barriers to advancing in more extensive operations are commercial, or few nations follow preparation circumstances and public view. A definite diagnosis cannot be executed among current, useful biomass energy plants and more traditional contaminating furnace projects[72,73]. From the commercial point of view, the application of crops without aid is too costly to provide, whichever bioethanol or biodiesel is at a cost contentious, including untaxed fuel. In contrast, techniques toward employing cost-effective lignocellulosic substances have not been completed [74-77].

2.1.2. Classification protocol of biorefineries and the FEW nexus

A classification approach toward biorefinery operations concentrating upon profitable ideas or might suit attention in the coming years to produce high volumes of transportable biofuels. The primary purpose of this classification method in biorefinery operation may be to classify the resulting four principal characteristics (arranged in order of significance): platforms, products, feedstock and processes. A biorefinery system is a transformation pathway of raw materials into outputs through platforms and processes. The platforms are intermediating via the final goods are obtained. They are the essential characteristic in defining the kind of biorefinery [78].

These also serve appropriate possibilities to promote complete operations, which convert nexus provocations within integrative possibilities. Such a complete operations strategy should include the FEW nexus as a structure for improving biomass utilization sustainability. The following points, such as biomass ingredients and extricate food segments within raw material, energy, and water, combine to develop environmental and utility chain stages. The remaining consumption, including wastewater processing by restoring nutrients, is restored toward food and biomass feed material improvement with water to reuse and recover and combined effect among FEW nexus elements with biomass utility chains [30]. Critical problems correlated by biorefineries, including the FEW nodes with the whole quantity chain, should be analyzed similarly [81].

2.2. Outline of food-energy-water (FEW) nexus in biorefineries

Kibler *et al.* [82] suggested the food trash impacts the FEW nexus through two consistent tools; mutually are operated by human performance and decision-creation. First, within both generation and waste administration aspects, food waste consequences toward the FEW nexus. Second, preferences constituted food acquiring, using, control, and joint determination forming on the societal level concerning food composition and food waste administration (such as landfilling *vs* AD) [83]. Entirely food delivered needs sources and energy, water manure, herbicides/pesticides, land, and employment, whether employed or misused. Water and energy are applied instantly into the food product when water and

energy are required to spraying water for ship food, or produce chemical manures and pesticides/herbicides. **Fig. 1** displays the design of the FEW nexus models [84].

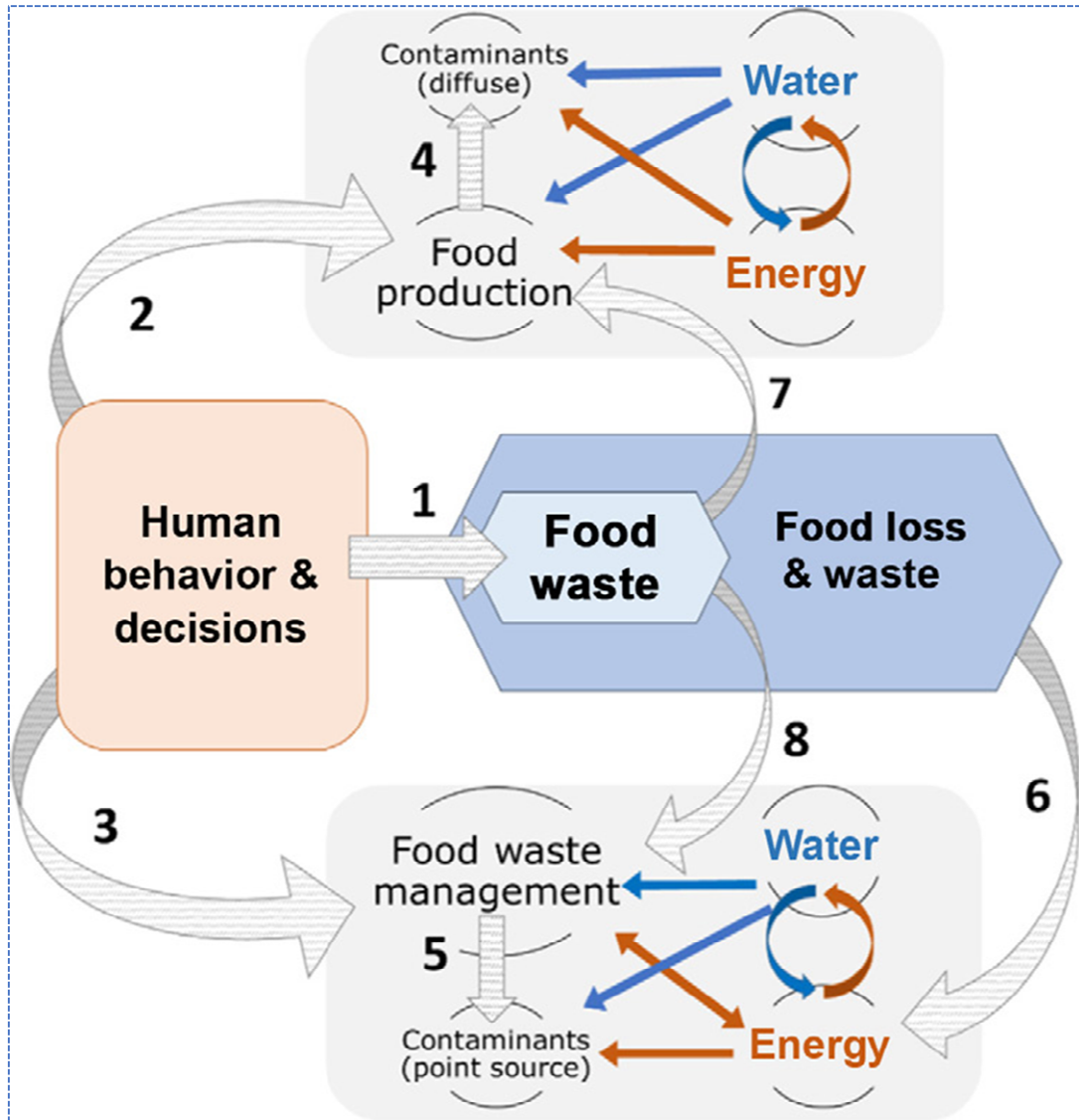


Fig. 1. FEW nexus in food construction and waste organization stages. Reprinted with permission from Ref. [84].

The FEW nexus of biorefineries shows the numerous designs contingent upon the biomass and methods. However, entirely biorefineries will affect water, energy sources and food stores. **Fig. 2** summarizes synergies of current biorefineries that primarily practice for food crops, such as corn, and wheat, to biofuels production [11].

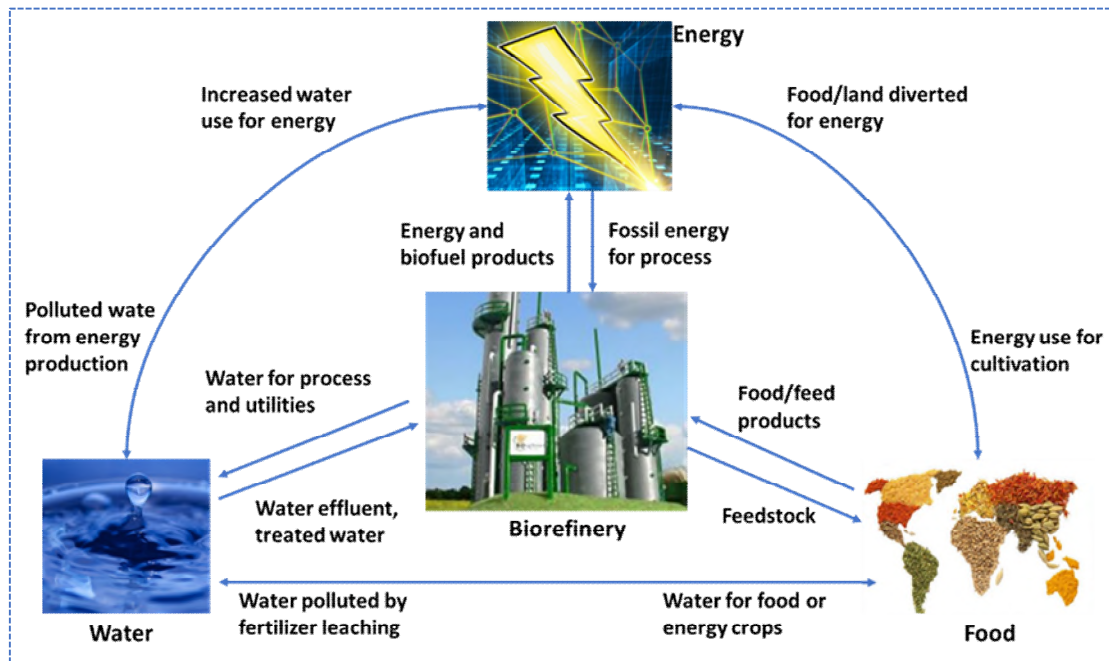


Fig. 2. FEW nexus relations about existing biorefineries that largely practice for food crops or first-generation raw materials. Reprinted with permission from Ref. [11].

Moreover, ineffective energy practice in biorefineries would drive improved water control to cooling services, building an energy-water node problem.

2.2.1. Prospects by FEW nexus in development of biorefinery

To understand the nexus hurdles, the current biorefineries should be developed in combined conveniences for allowing on-site energy production by sewage [88], crop utilization, alteration of supplies and food components at the outline step [89]. However, the demand for isolating biomass parts for effective adaptation within multiple outcomes generates sufficient complexity and range to mass and energy combination. Moreover, biorefineries may guide closing stock circles and annular economy [90,91] used for practical administration of consumption in combined effect including food, energy and water policies.

A further extensive review of the potential toward life cycle assessment (LCA), deep investigation and mathematical software design to fix nexus effects has been exhibited subsequently [94]. Several investigations have been done on the three elements of the FEW

nexus into an optimization structure [95]. The nexus has been overly stressed within analysis progress across the preceding decade as a device to drive organizations towards a green market and usually better source protection. Instead, the FEW nexus has been emphasized toward its potential to examine the interdependencies among sources at various ranges, given as an antidote to conventional “siloeed” techniques [96]. Hence, there is a requirement to produce devices, that may find innovative clarifications by approaching nexus problems, supply trade-offs, and analyzing nexus integration possibilities.

2.2.2. Biorefineries and various losses

An essential part of biomass power is wasted in heat through chimney gases, water vaporization, and atmospheric heat while the previous choice is practiced. The waste-water approach may produce biogas applied to fuel a reservoir or a component in thermochemical fuel generation. As well as the energy waste, a considerable water volume is also dropped through vaporization and chimney gas discharge to the atmosphere. Current literature has shown the significance of water reprocessing in a conventional biorefinery method, especially in the biochemical regeneration method [97]. Therefore, method options, including the potential to enhance the process's energy effectiveness and reuse water, are required. MFCs and MECs are possible technologies for employing polluted water while producing valuable energy into electricity or hydrogen sequentially [98]. Alteration of different sugars and carbohydrates toward power utilizing MFCs has been described [99,100]. It will unlock the opportunities for the growth of method alternatives employing MFC/MECs to improve energy by remaining organics into the biorefinery while allowing water recycling.

2.2.3. Connection between FEW nexus, biorefineries, BESs

Here, the question is raised: what will be a relation between FEW nexus, biorefineries, and BESs? Therefore, process options, including the possibilities to develop the method's energy performance and convert water, are required. The streams' capability to generate energy and H₂ described the synergy among biorefineries and BES. Combining BES technology in the biorefinery process may be notable for improving energy enhancement through biomass. It will drive advanced energy effectiveness, reduce surplus streams generated within the biorefinery, and reduce greenhouse gas ejections. These factors may

provide development into the sustainability of biorefinery methods. Several investigations have described the evaluation of sustainability parameters concerning BES technology.

In an introductory study of FEW synergies, biorefineries can provide first-generation bio-oils that can be applied as reservoirs. Bio oils give just 4% of worldwide carrying fuels. Hence, the production makes 3% of global water use. They practice land and products that could fill 30% of the community, allowing food scarcity. Using the constant Yellow River's yearly water exudation, 5-10% of developed land requires being distracted to give biorefinery raw material to reach China's bio-oil destination [85]. These facts show the influence of regularly examining the nexus synergies of bioindustries to circumvent negative results and scale supply trade-offs. The weak energy-food interfaces have been extensively discussed that present biorefineries participate obliquely towards increasing food with increasing animal supplies [86]. Synergies among the energy method derive using natural gas and power to drive biorefining methods, including reducing the life cycle energy releases carbon equilibrium [87]. This indicates an improvement to fossil sources that bioenergy, including bio-oils, to restore, an irony that needs to be evaded or reduced.

2.3. Bio-electrochemical system

BES is a suitable substitute approach associated with other traditional wastewater operation and bioenergy conversion or reproduction processes. In the last few years, BES prepares more fascination owing to the energy yield from wastewater with the metabolism of microbes. This method can work as a feasible approach toward clean energy production linked to more useful bioremediation. BES has been examined for its possible applications in several fields, particularly wastewater processing, including synchronous reproduction of bioenergy and biopolymers [101,102]. BES is an innovative method based on electrochemical alteration methods, able to transform the biological energy deposited in recyclable organic substances with the catalytic performance of microbes. BES involves oxidative and reductive half-cells to generate power or other chemically obtained outcomes (**Fig. 3a**) by combining the bio- and electrochemical methods.

The derivatives and sources such as heavy alloys, nutrients, ores, and transitional manufacturing chemicals may be retrieved through microorganisms catalyzed redox effects. A relatively massive quantity of analysis studies has been issued to discuss the

various features of BES. For example, electrode substances, materials employed for segments, supports, microbes, contaminant extraction, modelling perspectives, etc., to improve the activity of BES [103], though, some manuscripts (around 1%) addressed the extent of by-products improvement by wastewater in BES [104,105]. BES and its applications in energy generation (**Fig. 3b**) [106].

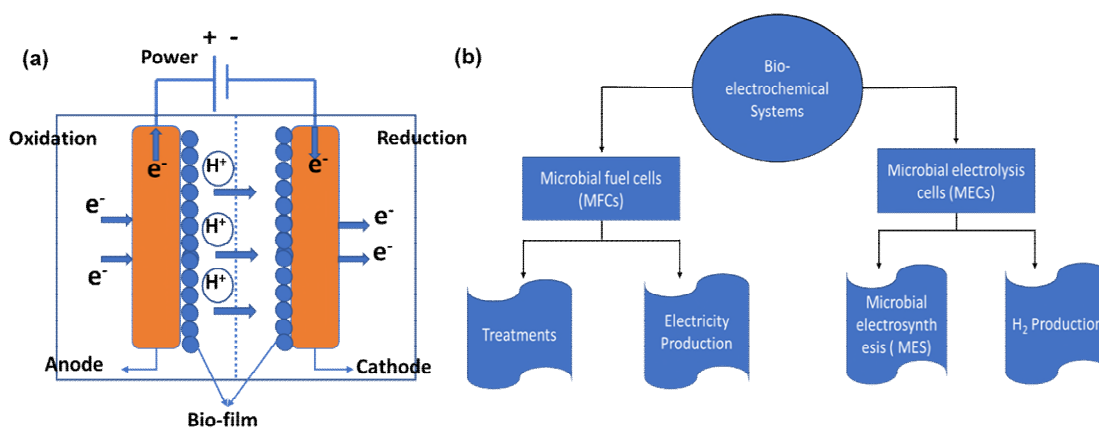


Fig. 3. (a) Schematic diagram of a bio-electrochemical system and (b) The BESs and used in energy generation[106].

The dual objective of BES for energy production and the bioremediation approach is more appealing for scientists. BES gives an encouraging and sustainable clarification towards restoring valuable synthetic derivatives in the presence (or absence) of a plan to produce renewable energy in the term of power. However, to deliver it more useful for real-world purpose and actual energy accumulation, the restoration of derivatives by wastewater should be concentrated.

2.3.1. Types of bio-electrochemical systems:

BES may be categorized reliant against the form of works: MFC, microbial electrosynthesis (MES), MEC, microbial desalination cell (MDC) and microbial solar cell (MSC). Generally, in each BES, the electroactive bio-sheet at anode oxidizes the organic material to produce energy. However, at the cathode, various products, for example, hydrogen generation and precious chemicals, may be obtained depending on the model of BES [107]. The appropriate kind of BES, including explicit classifications, are analyzed following in the next segments. The primary laws of MFC and MEC operations are

described in the following part. A description of cell studies is listed under the research examining MEC and MFC cells created, including an anode, a cathode, electrolyte, and an electrical loop[108].

2.3.1.1. Microbial fuel cells:

MFCs tackle power by the microbial oxidation for biological material employing a solid electrode as an electron receiver [109]. The cathode electrode's surface in MFCs promotes microbial adjunct, including oxidation of organics and producing electrons that assign together towards the cathode cell through an outer circuit comprising a surface charge. Electroneutrality is approved via ions transportation within an ion penetrable mechanism or a membrane if power is generated inside the system. M.C. Potter described the first of such practices in 1911, which attained the highest voltage of 0.3-0.5 V. Generation of bioelectricity in MFC is immediately reliant on the volume of exoelectrogens (microbes producing power) instant at the cathode side, which promote the transmission of electrons by the mitigate support toward the anode. Oxidation of complex organic material into wastewater needs distinct microbial inhabitants. Many bacterial varieties have been observed as capable of generating electrical current. The current progress in MFC as a biosensor has the potential for immediate and real-world intensive care of water conditions, making more substantial fascination. MFC improved with electrochemically effective microbes has been employed as biochemical oxygen demand (BOD) and toxic memory biosensor [110]. Here, the power generation was recognized to rise during the pollute range progress. Furthermore, [111] specified that MFC would be applied for detecting food scraps. This can be suitable for quick evidence and characteristics of microbes within food products.

2.3.1.2. Microbial electrolysis cells:

MECs is an encouraging process concerned with biorefinery for trash to outcome regeneration [112]. Apparent power used at the power circuit of BES makes electrons as positive to the anode electrode also promotes the H₂ production on the cathode surface [113]. Opposite from the MFCs, the cathode of MECs works below anaerobic

circumstances that promote H₂ generation. However, the anoxic conditions in the MECs and the high concentrations of H₂ composition may also increase CH₄ generation once CO₂ and methanogens are possible. Some of the approaches to alleviate CH₄ generation involve:

- the aeration of the cathode assembly among groups
- decreasing the pH
- performance on low detention times
- providing a heat stupor upon the inoculum
- combining chemicals that hinder the extension of methanogens.

Hydrogen generation under MECs, according to their arrangements and the supports employed. Advanced electric currents are naturally perceived into MECs while associated with MFCs, owing to the different employed potential supporting the cathode restrictions [114]. Another separate MFC can give the energy needed for the MEC process as a power reservoir [48]. The high hydraulic detention time was agreeable for CH₄ generation within MEC and cost-effective.

MECs are innovative technologies that may produce H₂ gas by low-grade supports, for example, organic trash. A standard MEC is comprised of an anode and a cathode divided through an ion-exchange membrane, including the broad scope of purposes of MECs (**Fig. 4**) [115,116]. The electrochemically activated microorganisms oxidize organic material and alteration electrons toward the anode terminal to extracellular inhalation through membrane-connected cytochromes, shuttle particles and/or electroactive pili [117]. MECs, as an anaerobic method, generates slight scrap residue and are repellent to atmosphere changes than the standard activated residue method, highlighting the sustainability of H₂ generation and wastewater processing. During the prior several years intensively investigated on lab-scale MECs [116,118], industrial-scale MECs have displayed the following: the probability of H₂ generation by residential and metropolitan wastewater [119].

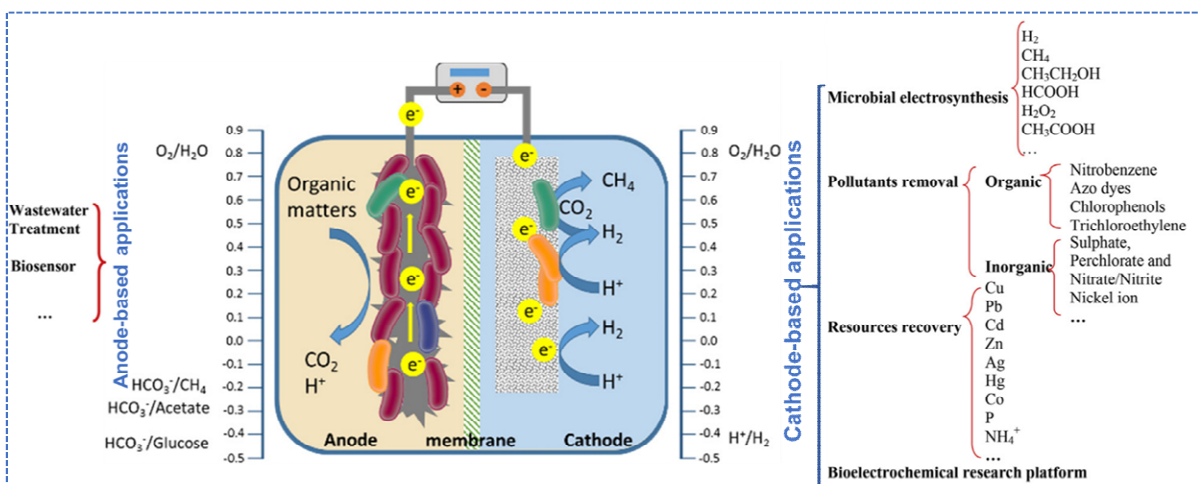


Fig. 4. Representation of a MEC, including an anode and cathode-based applications. The redox potentials are shown as V vs. NHE on pH 7. Reprinted with permission from Ref. [115,116].

3. BES analysis for energy efficiency and value-added product recovery

3.1. Hydrogen evaluations

In addition, the energy effectiveness increases the potential by applying a low-temperature MEC result; the production of H_2 can also increase the energy products by the method. The H_2 can be employed to produce high-octane fluid oils or the generation of more significant valuable compounds. The quantity of H_2 which a MEC focus can provide to several constituents with the chemical oxygen mandate of the method rivulet, flow rate, biodegradability about the biological material, and competence of catalytic H_2 generation reactions [120]. A protocol for ascertaining rates of possible H_2 generation with used-water rivulets was published recently. It can be applied to evaluate H_2 generation by several stillage rivulets, with those cellulosic and ethanol manufacturing methods [122].

A lignocellulosic biorefinery employing 2000 tons of maize Stover/day as supplies have a technical latent to produce $7200 \text{ m}^3/\text{hr}$ of H_2 , depends upon the opinion of 50% regeneration performance and a biodegradability aspect of about 0.5. The biodegradability aspect relates to the highest organic matter segment, which may be diminished with bacteria. The assessment involves H_2 , which produced with the sugar-degradation goods, e.g., furfural, HMF, besides the lignin-degradation goods, for example, phenolic aldehydes, ketones, alcohols, plus acids, as well as acetate and remaining sugars. The technique to

obtain energy by these particles within the electricity mode through an MFC was described latterly [123,124]. The anode microbial reactant acts as the primary function of material division, electron and proton generation interfaced among the H₂ generating electrochemical reactant at the cathode surface and the required overpotential (i.e., a MEC). So, H₂ generation by these streams applying a MEC is achievable [125].

Wastewaters by industries, farms, and municipal roots include large quantities of decayed organic material that are an inherent source for chemical and fuel generation [126,127]. Biological processing techniques are recognized for the generation of energy by used water. Methanogenic anaerobic absorption has previously been applied globally to the generation of biogas. H₂ generation by wastewater with acetogenic agitation is a different process of utilizing wastewater [128,129]. Though lowering products and thermodynamic constraints into microbial metabolism are under the agitation method. BESs suggest a substitute technology to improve the H₂ product at a comparatively low electric energy supply compared to traditional water electro-oxidation (**Fig. 5**). This technology is also called bioelectrohydrogenesis [130].

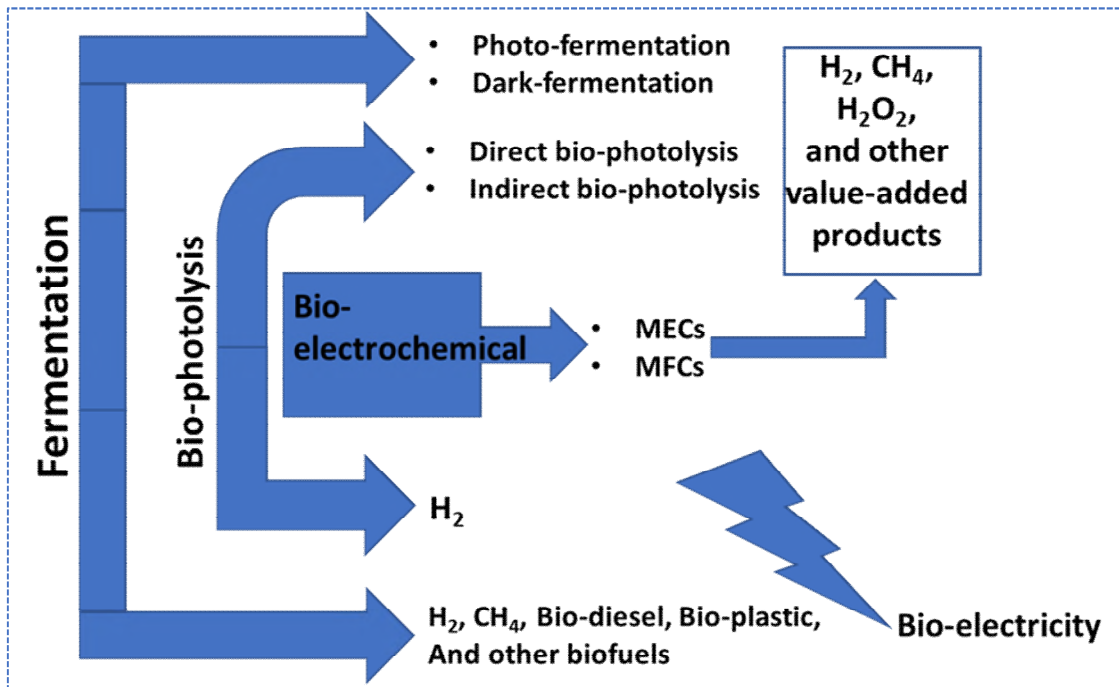
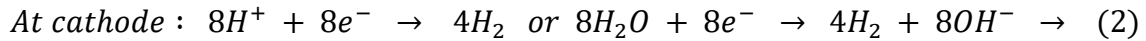
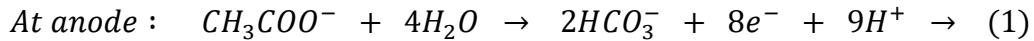


Fig. 5. Representation of the biological paths toward clean and sustainable H₂ generation. Reprinted with permission from Ref. [129].

Hydrogen is widely accepted equally for chemical and fuel, into different industrial methods, for example, updating fossil fuels including immersing oils, so H₂ has anomalous related to CH₄. H₂ generation by wastewater is seven times higher than CH₄ generated by the equivalent quantity of wastewater toward the COD source. H₂ generation during MEC can be prepared by several organic origins, including waste substances in addition to non-fermentable substrates [1]. Microbes oxidize organic composites naturally, acetate and produce CO₂, electrons, and protons on a MEC cathodic probe. Besides an outside potential, the electrons move toward the anodic probe at which it joins with protons to produce H₂ gas (**Eqs. (1) and (2)**) [2]. Toward charge impartiality, the protons transfer from cathodic to anodic electrode into the suspension. In tradition, a potential of >0.2 V is needed for H₂ generation in MEC, which is smaller than the standard potential (>1.6 V) expected toward water electro-oxidation. A membrane divides the cathodic and anodic probe assembly to evade the complexity of the support and output. The reactions which arise within a MEC by acetate as support is as below.



H₂ production on the negative probe happens after overwhelming the endothermic limit of 0.414 V versus SHE using a low exterior potential variation of 0.14 V toward the MEC. The oxidation reaction brought out with the cathodic probe germs may provide the residual over potential (0.279 V) [132]. Simultaneously wastewater processing and H₂ generation in MEC is an effective way to produce reliable energy.

There are numerous benefits of composing H₂ vs CH₄ with the organics already within the biorefinery method water rivulet. These involve a great interest due to the high cost of H₂, the capability to apply instantly for upgrading biorefinery outcomes, such as F-T liquids, possible to work as oil into fuel cells, and as an essential substance toward the regeneration of different biorefinery intermediates to valuable commodities [133]. Updating F-T liquids toward significant octane or big decane oils will be a crucial move to generate renewable shipping fuels. While the application of CH₄ to improving F-T liquids is probable [134], the further moves into such a method than that of MECs can drive to more expensive and

economical overall energy competence. Application of H₂ being a fuel under PEM also different fuel cells can similarly occur toward higher energy regeneration competencies related to the practice of CH₄ under elevated heat fuel cells or into turbines, etc. [135].

Several methods into biorefineries may help from onsite H₂ accessibility. The three fractions-cellulose, hemicelluloses, and lignin of each stock tree may apply H₂ generation of other higher-value results [125]. Sorbitol is an initial element of various goods, with vitamin C, alkyd resins, and surfactants [138]. The hydrogenation procedure from d-glucose into d-sorbitol. *Fructose* is a different output produced by glucose through glucose, applying high pressure plus temperature circumstances under the appearance of H₂. Its application into food and pharmaceutical products may be transformed within hydroxymethyl furfural (HMF) plus levulinic acid, which are valuable initial substances toward several manufacturing chemicals. H₂ is utilized in many of these methods for turning HMF plus levulinic acid within furan and levulinate byproducts. The reproduction of succinic acid by glucose is at its approach toward speculation. Subjunction of the support glucose, including H₂, has been given to improve succinic acid produces suggestively [139].

3.2. Acetate

MES used a different way to set CO₂ within multi-carbon aggregates, for example, liquid oils or chemicals with current-ambitious reduction reactions. This treatment is recognized as a technology toward the warehouse of electrical energy into the C-C bond of valuable compounds [140]. Additionally, it showed the first evidence of MES applying the acetogen *Sporomusa ovata*, which could allow atoms immediately received by a graphite cathode toward the mitigation of CO₂ to generate acetate and low amounts of 2-oxobutyrate. Electron curing in certain goods was additional substantial than 85% of the electrons carried on the cathodes. The subsequent investigations showed the method of CO₂ mitigation employing MES through a more comprehensive array of microorganisms [141]. The product yield (or yield performance) of inorganic nutrients and organic substances are arranged within **Table 2** [103].

Table 2: Relative yield also energy productivity of biochemical complexes. Reprinted with permission from Ref. [103].

Elements	Reactor/technique	Subtraction productivity	Energy/current detention efficacy
CH ₄	CO ₂ - CH ₄ into BES (at -0.65V E _c vs. SHE)	0.005±0.002 m mol/day	
	MEC with electromethanogenesis (at -0.7V E _c vs. Ag/Agcl)	80%	90%
	MEC (on -0.55V E _c vs. SHE)	0.006 m ³ /(m ³ /day)	51.3%
H ₂	Bio-catalyzed electrolysis (used potential: 0.5 V)	0.02 m ³ / (m ³ day)	53 ± 3.5%
	Bio-catalyzed electrolysis (on -0.7 V EC vs. SHE)	0.63 m ³ /(m ³ day)	49%
H ₂ O ₂	BES (used potential: 0.5 V)	1.9 ± 0.2 kg/(m ³ day)	83.1 ± 4.8%
Ethanol	BES (on -0.7 V EC vs. SHE)	13.5 ± 0.7 mM	74.6 ± 6%
Triglycerides	Smooth dish electrochemical cell (on -0.9 V EC vs. SHE)	Caproate: 739 mg/L	26%
		Caprylate: 36 mg/L	

The MES method's circumstances need to be designed to provide the ideal metabolism of biocatalyst already on the biocathode. A proper electrode electron receiver could be possible toward the cathodic reaction. A strong cathodic reaction employed the cathodic or used voltage, which discloses the thermodynamic boundary of a biological response within MES. Homo-acetogenic microorganisms may effectively change CO₂ into acetate, a primary central particle to the generation of biochemicals [142]. Thermodynamically, the reduction of CO₂ into acetate needs 280 mV vs SHE cathodic voltage. Following applied circumstances, a significantly below voltage is needed to succeed the latent wounds due to the microbial energy acceptance, including mass and charge transmission resistances associated with a bio-electrochemical operation. Aside from the primary determinants, many other fundamental issues, such as probe substances, reactor configuration, and media in electron transfer, control the complete process performance [143].

Nevin *et al.* [140] described the probability of converting CO₂ to acetate through acetogenic microorganism *Sporomusa ovata* electrons carried instantly of a graphite probe. It was observed that *S. ovata* biofilms at the graphite cathode facet absorbed electrons of the terminal and turned CO₂ into acetate and little volumes of 2-oxobutyrate. These outcomes apprehended 85 % of provided electrons. An engaging and unique path introduced the idea of MEC, which might transform solar energy into precious organic commodities more efficiently compared to conventional methods. The relevant tools, base opinion, and method perception are still on the route as entirely different processing. Rabaey and Rozendal [144] explained the policies, hurdles and possibilities of MECs, provided an essential detail of illustration on this exciting and unique system on the nexus of microbiology and electrochemistry. Currently, the MES technology is opening; hence, modern research focuses on building new ideas.

3.3. Methane

A membrane-less MEC would generate pure energy, including methane (CH₄) as the final major commodity. CH₄ is usually identified within the MECs at the time of H₂ generation due to methanogens' extension. The CH₄ generation through MECs is different from inoculum, support and container arrangement [145]. The presence of methanogens is stunning within H₂ generating MECs, as it reduces the H₂ product. Various approaches

have been utilized to hinder the completion of methanogens during MECs. Though the maximum of the techniques is inefficient or energy accelerated. Somewhat of restraint of methanogens, the direct product of CH₄ within MECs includes numerous benefits than that of conventional anaerobic digestion (AD) techniques.

Methane production by AD of organic supports had primarily been used, which is acknowledged as a renewable method. However, to develop the CH₄ production method [146], reprocessing the CO₂ produced by different biological methods may generate higher CH₄ by MES. Microbial electrosynthesis of CH₄, also identified as electro-methanogenesis, may move at lower heat compared to AD. The importance of electro-methanogenesis by MEC is that the organic material existing within wastewater may be employed and at a similar time, the CH₄ product is obtained. This method gives more benefits than conventional methanogenesis regarding greater CH₄ yield and application of the wastes emitted by AD methods [50,31].

Wang *et al.* [147] examined the CH₄ and H₂ generation during MEC; H₂ gas was recognized in the last of the first cycle. The reactors were later run across various batch cycles covering the next month to secure durable and steady performance before conducting H₂ generation experiments (**Fig. 6(a)**). In contradiction to BES, lumazine did not affect CH₄ generation. The lumazine direct CH₄ generation into mixed-culture MECs on an input of 0-1144 μM (**Fig. 6(b)**). **Fig. 6(c)** confers a standard CH₄ product into the anode and cathode assembly below anodic methanogens restrained circumstances following a BES dose. Acetate was not identified within the cathode, and a BES dose in the cathode assembly did not decrease the level of CH₄ generation. It shows that CH₄ composition under the cathode is essentially a hydrogenotrophic methanogen-linked method, including acetoclastic methanogens' minimum participation. Remarkably, CH₄ was yet composed on the cathode. In contrast, the cathode chamber was autoclaved to reduce the biogenic movement near 120 °C concerning 20 min as an abiotic controller experiment. However, its capacity was comparatively small associated with the unautoclaved cathode (**Fig. 6(d)**). Therefore, a different pathway for cathodic CH₄ product is also assumed [145].

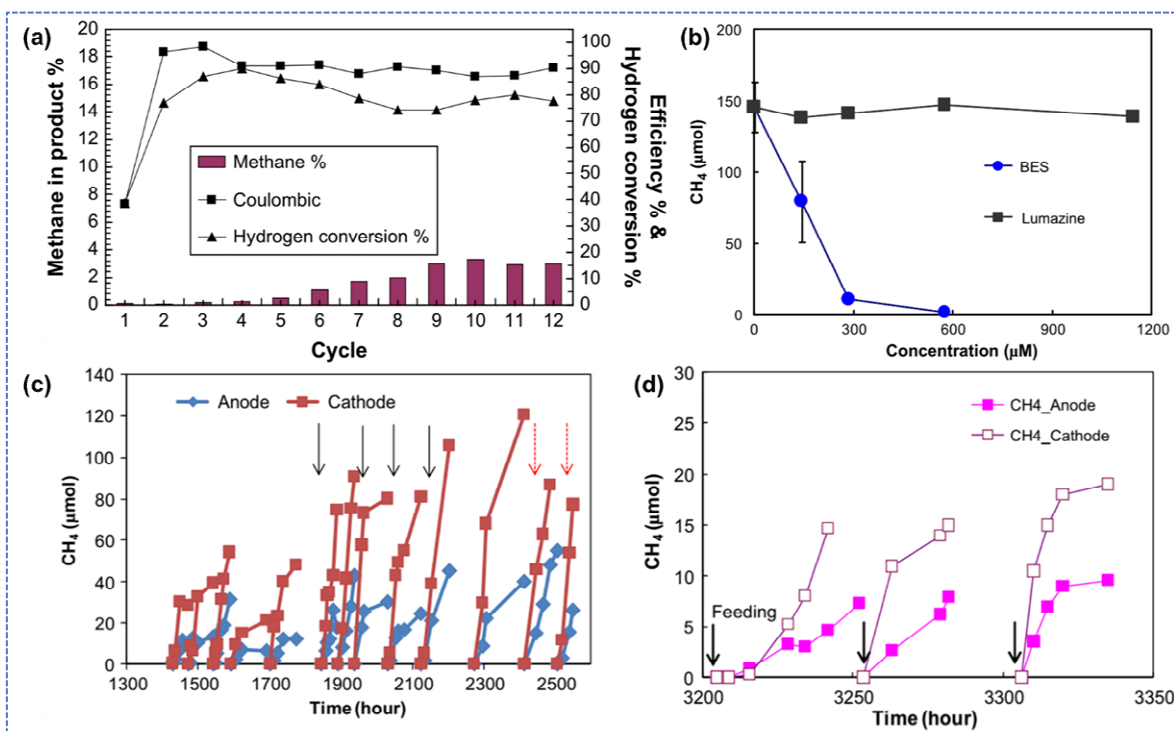


Fig. 6(a) CH₄ generation and H₂ conversion performance into a separate container MEC on 0.7 V. Reprinted with permission from Ref. [147]. (b) CH₄ generation as a function of methanogen inhibitor combinations (about 2.86 μM acetate at 0.5 V). (c) CH₄ composition under the anode and cathode while methanogen-contained process including the BES dose within anode (about 2.86 μM acetate). The arrows show chemical doses in the cathode: solid for BES of 286 μM; dashed toward lumazine about 286 μM. (d) Abiotic CH₄ composition, including an autoclaved cathode chamber, happened as the anodic methanogen-contained process. Reprinted with permission from Ref. [145].

Further, the bioelectrochemical CH₄ products applying a different container shape into MEC mode [148], employing conventional graphite as an electrode. Processing of AD effluents within a separate container MEC and CH₄ generation has prospected as a viable technology toward effective scrap processing. Sasaki et al. [149] concentrated on applying a membrane-free form in a practical voltage 0.395 (vs SHE) in neutral pH and showed effective CH₄ generation. It was mentioned that the controlling methanogens at the anode probe were different from those developed under command reactors without electrochemical reactions. Villano *et al.* [150] practiced different methanogenic cultures as cathodic biocatalysts into a dual compartment arrangement CH₄ by CO₂ on different employed cathode voltage -0.65 V to -1.0 V (vs SHE).

During the scalability comparisons toward MEC, including multi-probe arrangements toward H₂ generation employing acetate as support [151], published that the methanogenesis rates improved continuously over time. Furthermore, after 16 days of running, the method presented a small H₂ product rather than a consequent CH₄ generation, including better substrate reduction performance, revealing that electro-methanogenesis became the practical method on the biocathode while working with wastes shady agitation and AD methods.

3.4. Recovery of heavy metals

BESs are rising as a new technology floor toward removing and recovering metal ions by metallurgical scraps, method rivulets, and wastewaters. Biodegradation of organic material through electroactive biofilms on the anode has been favourably linked to the cathodic reduction concerning metal ions. Heavy metals existing within industrial effluents model a severe obstacle to the climate and public fitness as several of these minerals are poisonous on lower densities. Physical, chemical, and biochemical types of machinery towards heavy metals extraction from drainage water has been formed and works assigned for its potential restoration. Such processing methods can be offered economical and renewable by promoting metal restoration utilizing BES [152]. Furthermore, BES can extract and collect metals when being on actual low frequency in an aqueous means, delivering a cost-effective feasible opportunity compared to other substitute methods. Some investigations exhibited retrieval of metals through lower redox potential, for example, Ni²⁺(E^{0'}=0.250 V), Pb²⁺(E^{0'}=0.130 V), Cd²⁺(E^{0'}=0.400 V) and Zn²⁺(E^{0'}=0.762 V), via applying biotic anode and abiotic cathode below the impact of outer energy origin to drive the migration of atoms by anode to the cathode front. Furthermore, the more leading reduction of chromium and different transition elements may be accomplished following microbial metabolism on bio-cathode, including the absence of poised voltage [153].

Metals existing into trash brooks should be collected to reduce their warning to the atmosphere and reuse limited sources. For example, numerous metal cations, such as Cu²⁺ and Hg²⁺, have been examined as electron gainers on the cathode of MFCs [154,155].

These alloys may be retrieved into MFCs including, further power generation, as they usually have excellent conversion potential (for example, 0.34 V toward Cu^{2+}) [156]. Unfortunately, though, to these metal ions, including lower mitigation voltages, restoration may not be accomplished during MFCs in the absence of additional energy accumulation. Nevertheless, considering MECs possess numerous benefits, such as low energy necessary than that of the standard electrochemical method, this could be a substitute for energy-effective metal restoration.

The bioelectrochemical recovery of heavy metals (such as; Cu, Pb, Cd, and Zn) of a blended solution into 2 M HCl was provided out under four test periods for recovering the elements in the direction of reducing conversion voltage. At the cathode surface, the metal ion arrays are displayed within **Fig. 7(a)**. Through periods 1, 2, 3 and 4, Cu^{2+} , Pb^{2+} , Cd^{2+} , and Zn^{2+} were dismissed of the catholyte. The input potential needed to collect the metals into the bioelectrochemical reactor enhanced, including reducing cathode voltage (negative cell potential indicates electrical power input under **Fig. 7(b)**). The nutrient mechanism TOC absorption and pH declined at the anode surface due to acetate oxidation and charge immigration (**Fig. 7(c)**). During period 3, about 230 h, the test was discontinued for four days. The new nutrient medium supplied while again started over the anode, which describes the rise in TOC absorption and pH recognized. While periods 1-3, the anode voltage was constant among -0.25 and -0.17 V (**Fig. 7(d)**) [156].

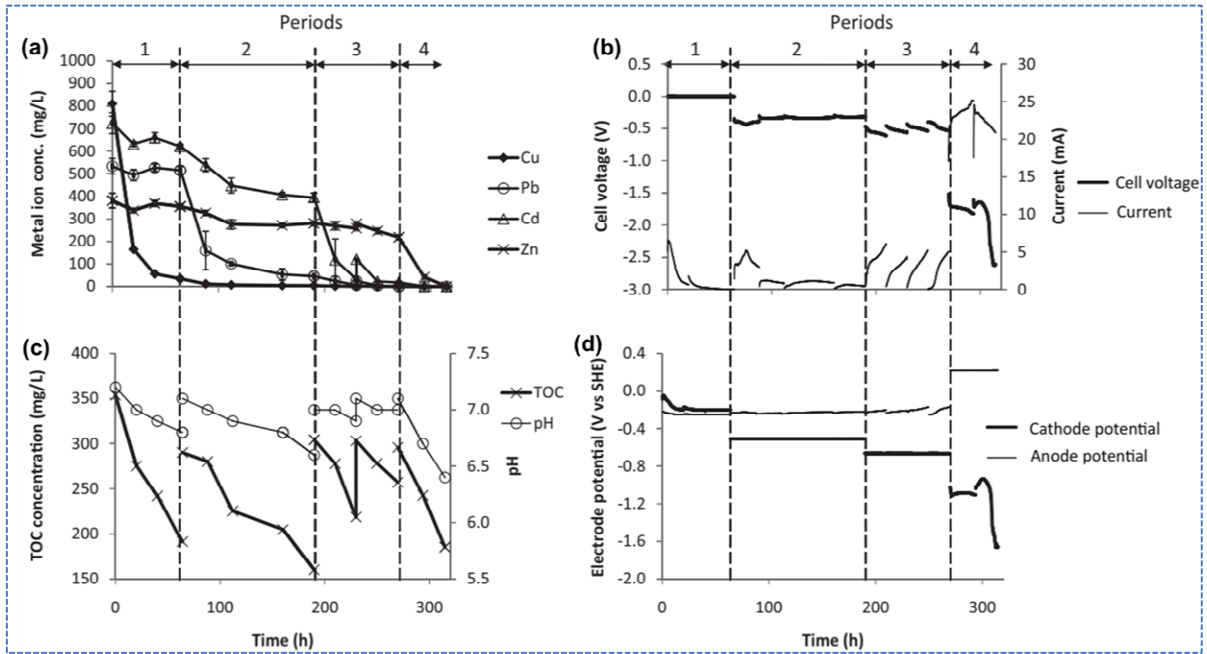


Fig. 7. (a) Metal ions concentrations near the catholyte, (b) cell potential and current, (c) TOC absorption and pH within the anolyte, and (d) cathode and anode voltage while the examination toward the restoration of metals by a various solution catholyte. Reprinted with permission from Ref. [156].

3.5. Nutrient's recovery

Nutrient's administration is an essential responsibility of wastewater processing, and the elimination of nutrients is normally compared among significant energy/financial prices. The nutrient extraction and rehabilitation by urine and pig drainage water may be brought out with choosing biological methods under BES [157-159]. Urine may be recognized as an absolute fuel (origin of C, P, N, and K) toward power generation within MFC. Therefore, the potential energy and NH_3 retrieval by urine applying a BES was investigated [160,161]. NH_4 revival in MFC was achieved by managing urine near the anodic container. The following consumption attended movement of NH_4^+ by dispersion to the cathodic container. The higher current density of 3.6 A/m^2 including the pointedly advanced improvement of NH_4^+ near 61% in MFC, was summarized into a microbial electrochemical cell below power-ambitious immigration [162]. The modern MFC promoted ammonium improvement theory is presented in **Fig. 8(a)**, and **Fig. 8(b)** displays the test structure's design.

Blázquez *et al.* [163] published a unique method for handling high potency sulphate wastewater with bio-electrochemical methods. The advancement of bio-cathode including autotrophic sulphate reduction bacteria (SRB) and sulphide oxidizing bacteria (SOB), drives metal sulphur restoration through the infectious metabolic method. By employing H_2 as an electron contributor, the conversion of sulphate with SRB was perceived inside the cathodic container assembling H_2S [164], completing sulphate conversion rate of about 388 mg. $S-SO_4^{2-}/(l. \text{ day})$. Additionally, H_2O was scattered in the anodic container to provide O_2 , which moved to the cathodic cell by anion transfer film and gave attention to SOB to catalyze the H_2S oxidation to generate elemental sulphur collected at the cathode exterior.

3.6. Hybrid method of wastewater processing for effective co-product retrieval

By improving BES investigation, numerous readings appeared by combining wastewater processing, including BES, to improve the effluent feature and retrieve the most accessible energy instant toward the wastewater. The outcomes, for example, bio- H_2 , bio- CH_4 , power, bio-fertilizer, caustic soda, H_2O_2 , etc. were developed through the combined performance of composite BES. H_2 and CH_4 generation when such integrated process executes the composite practice more engaging and effective [165]. **Fig. 8. (c & d)** A sediment microbial fuel cell (SMFC) including microbial anode and cathode gives energy toward the power control operation [166] and representing the sensor device for estimating subsurface microbial movement through the current generation by subsurface probes [167], correspondingly.

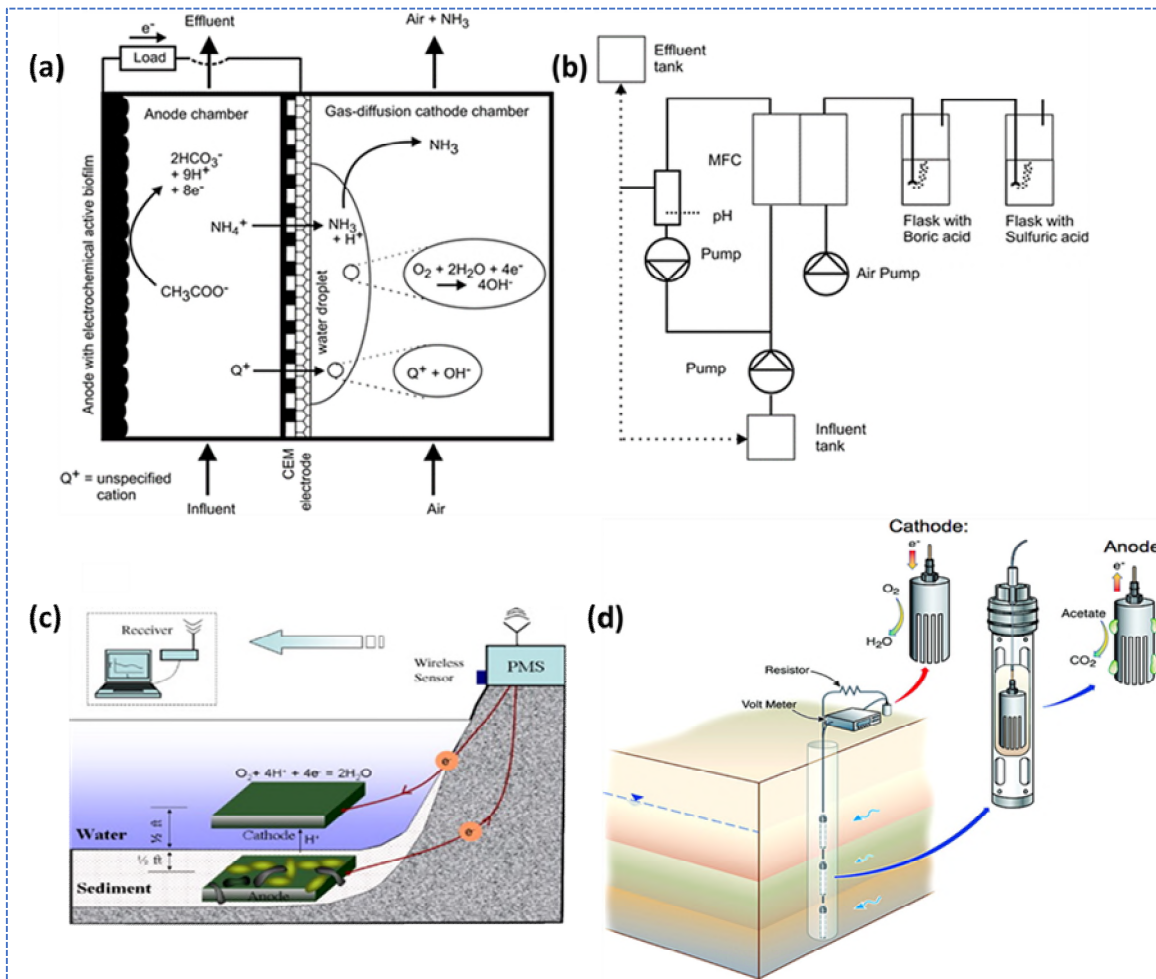


Fig. 8. (a) Schematic illustration of the methods included within the NH_4 retrieval employing an MFC; (b) Schematic illustration of the laboratory structure for NH_4 retrieval utilizing an MFC. Reprinted with permission from Ref. [160]. (c) An SMFC, including microbial anode and cathode, gives energy toward the power management system (PMS). Reprinted with permission from Ref. [166]. (d) The sensor device for estimating subsurface microbial performance through the current product from subsurface probes. Reprinted with permission from Ref. [167].

BES process reflected efficient wastewater processing separate from power production. Improved substrate (COD) decrease was recognized near-contemporary appearance and TDS extraction. COD elimination response ranged within 33.33% and 72.84% while the MFC process. According to energy production, the carbon part of wastewater worked as an electron contributor during the metabolic method occurring in substrate degeneration. A good decrease in COD identifies the valuable niche of diverse microflora in managing still wastewater.

Around 29.5% increase in substrate reduction was seen owing to the combination of probes in the dark-fermentation method (MFC operation; producing bioelectricity) related toward the dark-fermentation method (biofilm reactor without probe arrangement; producing biohydrogen) [168]. Distillery wastewater revealed higher substrate reduction performance related to composing synthetic and chemical wastewater assessed below comparable working situations in the same MFC arrangement. Improvement in substrate degeneration was published under MFC (dual container) while composite chemical sewer water processing on the developed substrate storing position as opposed to the traditional anaerobic processing method [169]. The perceived advancement in substrate degeneration through MFC performance may be associated with the bio-electrochemical catalyzed aspects as described: (a) direct anodic oxidation (DAO) wherever the contaminants are adsorbed at the anode exterior and become degenerated through the anodic electron substitution reactions [170]; (b) Indirect oxidation (IO) negotiated with the oxidants similar to chlorine dioxide, hypochlorite, hydroxyl fanatics, ozone, and H₂O₂ produced electrochemically at the anode exterior oxidizes the organic material.

3.7. The novel cathode and anode materials used in BES

An electrode is a compact electric material that provides the electric flow within BESs. The electrodes in BESs are classified in anode and cathode, including particular attention presented when selecting a substance as electrodes toward BESs. A few of these constituents such as good conductivity, improved mass transference, and high surface area. The activity of BESs depends upon the electrode substance. Metals prototype toward the anode comprises carbon and its components [172].

Substances fit toward the cathode comprise Pt black and graphite [173]. Pt covered electrodes usually are used owing to their effectiveness and perfection during the production of electricity. Notwithstanding the high price of Pt materials, this is the best-known reactant toward oxygen reduction reactions (ORR). However, inadequate oxygen conversion happens among Pt-free cathodes. Therefore, various electron acceptors rather than O₂ were employed within BESs to overwhelm the weak ORR, for example, potassium ferricyanide [174], permanganate [175], and sodium hypochlorite [176]. Biocathode is a

different approach that utilises the aerobic biofilm on the cathode exterior as an electron gainer [177]. During MEC, Pt is also the most suitable material for hydrogen evolution reactions (HER). However, owing to its vast price and easy poisoning, its use in these operations is insufficient. Distinct Pt-free synergists were studied in MEC toward HER, for example, stainless steel, nickel compounds, including tungsten carbide [178,179]. Likewise, biocathode is recommended to be utilized within MEC toward HER. Though, the biocathode influences the cleanliness of H₂ generated owing to the increasing methanogenesis bacteria [180].

3.8. BESs based desalination and wastewater treatment

Based upon the prior primary composition of the BESs depended on desalination, different arrangements, while depends upon the MFC or MEC, will be included within the next part.

3.8.1. MFC based desalination cells (MDCs)

A microbial desalination cell (MDC) is a BES that yields comparable characteristics by an MFC [181]. These Desalination cells based upon the electric driving power produced from the microbial metabolism within the anode assembly, which animates ion departure by the desalination container to both electrodes [182]. Several investigations have been brought out on the MDC utilized acetate, including phosphate buffer as an anolyte. As a result, the MDC gained higher activity, greater power density, higher COD discharge, and higher coulombic efficiency. This notable development within the cell performance was correlated toward the advancement of the conduction of the sewer water through the ions, stabilization of the pH on the anode, and the difference of the anode microbial arrangement [183,184].

3.8.2. MEC based desalination cells (MEDCs)

MEC is an example of the BESs employed towards new wastewater processing and biohydrogen generation implementing a minute voltage between the anode and the cathode

assembly. The microbial electrolysis desalination cell (MEDC) was a newly expanded technology by merging the MDC and MEC rather than the MFC arrangement as the desalination performance of the MDC is restricted with the varied voltages generated through the exoelectrogenic microbes [25]. Microbial electrochemical technologies (MET) dependent upon MFC or MEC (**Fig. 9(a)**) allow a chance to treat wastewater by the concomitant regeneration of power biologically. Within MFCs or MECs, microorganisms, including extracellular electron transfer (EET) ability, transport the electrons produced through the oxidation of organics during wastewater towards the anode [185].

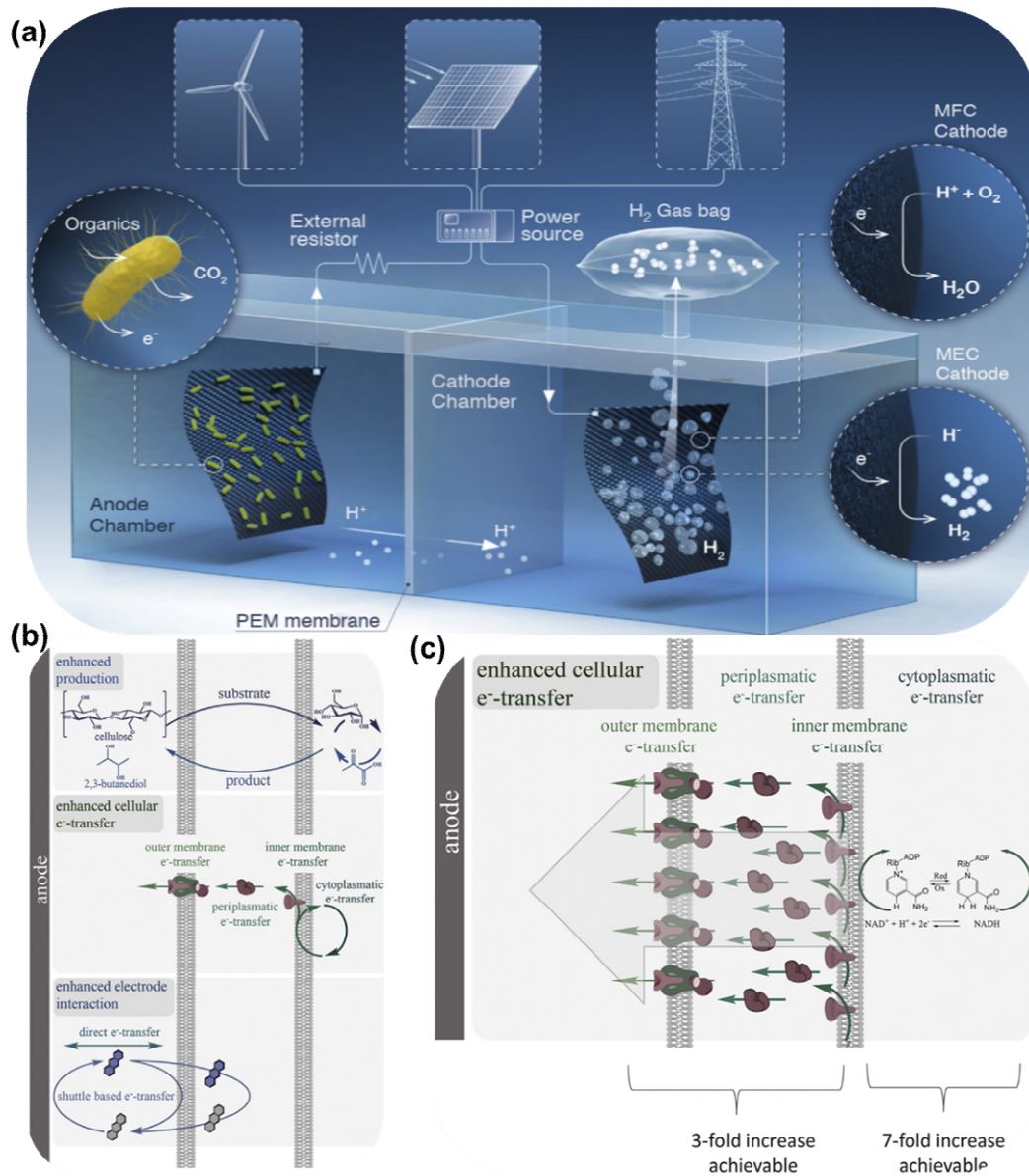


Fig. 9. (a) Schematic illustration of the various arrangements of METs toward wastewater procedure, including the restoration of energy. Reprinted with permission from Ref. [185]. (b) Various policies to improve the applications of microorganisms within BES. (c) Design the central way toward EET, including a tailored cytochrome channel and manufacturing the intracellular particle equipment. Recent research linking the design policies would drive over a 10-times increment in current value. Reprinted with permission from Ref. [186].

3.9. Important aspects in BES architecture and designs

Despite the layout and structure of the MFCs, the specific ingredients, e.g., separator, electrode substances, inoculum, and wastewater, are the final part of the production. Therefore, during the previous years, an extensive investigation has surfaced dealing with the growth of the peculiar parts to obtain this technology fit toward employment within the area. There is also an inadequate expanse to get a few novel progressions into the membrane and electrode substances these days. The membrane separators, or ion exchange layers, are a crucial ingredient of MFCs, which should have different vital resources, e.g., high ion exchangeability, durability into an intense atmosphere, and impermeability of the mass flux of oxygen and basic materials. Membranes are also the initial condition of biofouling improvement upon the cathode exterior. Hence, antifouling layers are further fit within MFCs [187].

Likewise, electrodes would maintain a large surface area, durability, biocompatibility, and electric performance. Most importantly, both electrode and separators would be economical to guarantee their practical purpose within industrial MFCs. Therefore, the maximum of the novel study turns throughout the construction of cost-effective substances.

3.10. Genetic engineering for enhanced productivity in BES

As we know, BESs may be utilized to produce valuable compounds, bioelectricity, wastewater processing, and biosensing, among others. Transforming a substrate toward the required output through generating current within a BES can be separated within various modules. To enhance the complete performance of the biocatalyst, the generative architecture of the smallest one of these modules should be examined. Within **Fig. 9(b)**, the four central modules are displayed. A facile-utilized path to improve extracellular inhalation is to concentrate upon intracellular destinations that interfere with electron alteration (i) by the cytoplasm toward the menaquinone equipment, (ii) by the menaquinone pool toward the periplasm, (iii) via the periplasm and (iv) via the exterior layer.

Other strategies try to develop potency by optimizing microorganism-electrode synergy. It may either be transferred with enhancing straight or shuttle-based electron transfer.

Immediate electron alteration can be enhanced by optimizing the organisms' biofilm production capabilities. In contrast, the shuttle-based particle transference is usually enhanced with growing the number of convenient shuttle composites. While in another side, through manufacturing, the intracellular electron supplies a seven-time increment within the current value that appears to be possible. While there seem to be no apparent causes of why the two consequences would not be combinable, the purpose of the central cytochrome way toward EET and the progress in intracellular releasable particles may start to a 10-times advancement within the current value (**Fig. 9 (c)**) [186].

3.11. Production of valuable materials in BESs

Hydrogen peroxide (H_2O_2) is a multifaceted, eco-friendly, oxidizing solid compound, including various modern uses. This is affirmed that BES is a promising technology for H_2O_2 biosynthesis, such as MFC and MEC usually. Currently, the BESs are developed of pure electricity production toward the reductive composition of an energy-rich substance, for example, H_2O_2 within MECs. The idea of MECs was introduced in 2005 by the essential characteristic of applying an outer voltage over the MFC potential to produce H_2 gas on the cathode by the proton conversion. A comparable policy, two-electron ORR toward H_2O_2 production, was primarily identified in 2009 within the cathodic container [188].

Electrodes that may generate H_2O_2 are supposed flawed for MFC uses because of not maximum power production. Although, it should provide an industrial compound into a green integration, especially during joined electrochemical methods [189]. It extensively knows that the anode voltage based upon the respiratory enzymes' potential to produce power toward the cell from the oxidation of the organic material. Generally, MECs need the potential produced from support oxidation on the anode to be supported, including an external power amount; to succeed the thermodynamic limitation and/or force the cathodic H_2O_2 production, extraordinary speeds initiated the creation of anolyte distinct from MFCs [190].

The current development within bioelectrochemistry of BESs and electro-biotechnology revealed a necessary combination of composites that significantly provide performance and capability: surface-active tools and surfactants. Lately, crucial scientific observation has been adjusted to replace the chemically manufactured surfactants with biosurfactants that microorganisms may present within various bioreactors. However, biosurfactant functionalities in existing systems have yet not been completely known. Models of the essential functions of biosurfactants cover: improving the exterior area and bioavailability of hydrophobic supports, adjusting the addition, the impartiality of microorganisms to the facades, joining within *quorum sensing* devices, merging of heavy elements, and antimicrobial performance. The functions discussed up can consequently have a significant influence on the BES execution [191].

3.12. Remediation of soil contamination by BESs

Effective non-invasive methods are required to observe the remedy manner of polluted clays. Hydrocarbon poisoning owing to unexpected spills and vessel leakages describes a warning over the atmosphere. Several design explications have been used to improve the remedy technique among the natural attenuation of the contaminants. Lately, BESs were produced and examined towards the remedy of polluted hydrocarbon clays. While the remediation method, the BES presents an atmosphere toward electrochemically active bacteria (EAB) for catalyzing the oxidation of natural electron contributors, such as hydrocarbon poisoning, and convey the electrons on the electrode [192].

In this regard, Ren *et al.* [193] composed a life-size execution of column-kind BESs as in situ clay bioremediation could base upon the length by the anode where biodegradation remains to be improved, that is the radius of impact (ROI). **Fig. 10 (a)** shows the image of column-kind BES into a polluted soil remediation analysis. Two column-kind BESs were assembled as exhibited within **Fig. 10 (b)**, including one BES comprising a graphite granule anode (GGA) and the additional a biochar anode (BCA). The clay pH raised to 0.13 in the initial 5 days on a circular length of 1 cm by the anodes of every BES (**Fig. 10 (c)**). It is associated with the adsorption of ions and H⁺ via the anode bodies, accorded by a 17-37% reduction in electrical conductivity (EC). On days 5 to 25, the pH slowly reduced

near 0.22 on each sampling period, probably due to organic acid production by hydrocarbon metabolism. However, the organic acids could be quickly absorbed, following into pH rebound. It was recognized under the research that a pH rebound around 0.13 at day 45. As a result, the soil EC improved around the anodes of all BES but reduced by length. Ions immigration inspired by water destruction on the cathodes to electrode modules may cause this conductivity pattern [194].

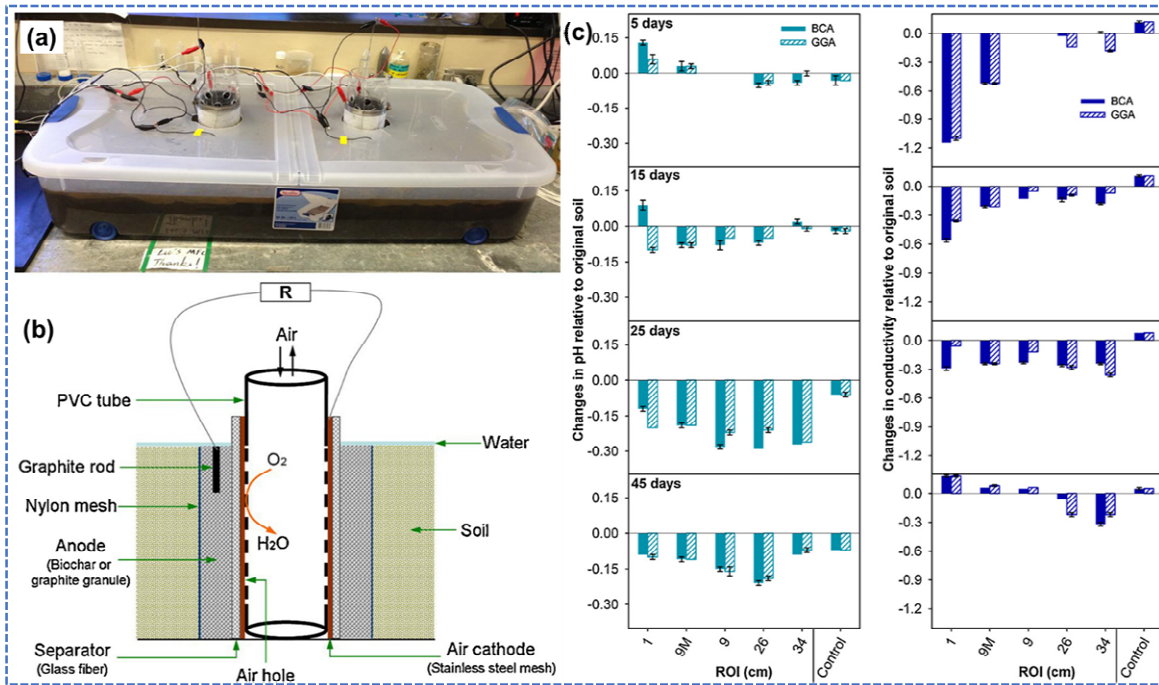


Fig. 10. (a) The image of column-kind BES within polluted soil remediation analysis, and (b) diagram of the column-kind BES configuration. (c) Variations into soil pH and EC during the BES- and control reactors. Reprinted with permission from Ref. [193].

3.13. Photosynthetic BESs

The scope of BES investigation covers a broad area of developing technologies that employ bacteria to catalyze anodic and/or cathodic responses in a fuel cell structure and has developed dramatically. Although most BESs use organic supports as a fuel reservoir (*such as* MFCs), numerous declarations are fueled with light steam. BESs applying different photosynthetic biocatalysts develop a comparatively latest method that is inadequately represented within the classification. Earlier, it may be separated within photosynthetic

microbial fuel cells (PMFCs; **Fig. 11 (a)**) and bio photovoltaic cells (BPVs; **Fig. 11 (b)**). Each kind of tool utilize an electron-generating phototroph on the anode; though, the final root of those electrons is separate. PMFCs are colonized with anaerobic chemoautotrophic bacteria that uses biochemical fuel being the initial electron origin and produce electric flows into a light-dependent style [195]. Different from traditional MFCs, P-MFCs may produce power regularly during plant development and may be utilized within any field used toward crop products without injuring the plants. However, the purpose of these regularities upon an extensive range has displayed significant hurdles up to now. It is initially owing to commercialization; further, low energy ability while associated with other operations [197].

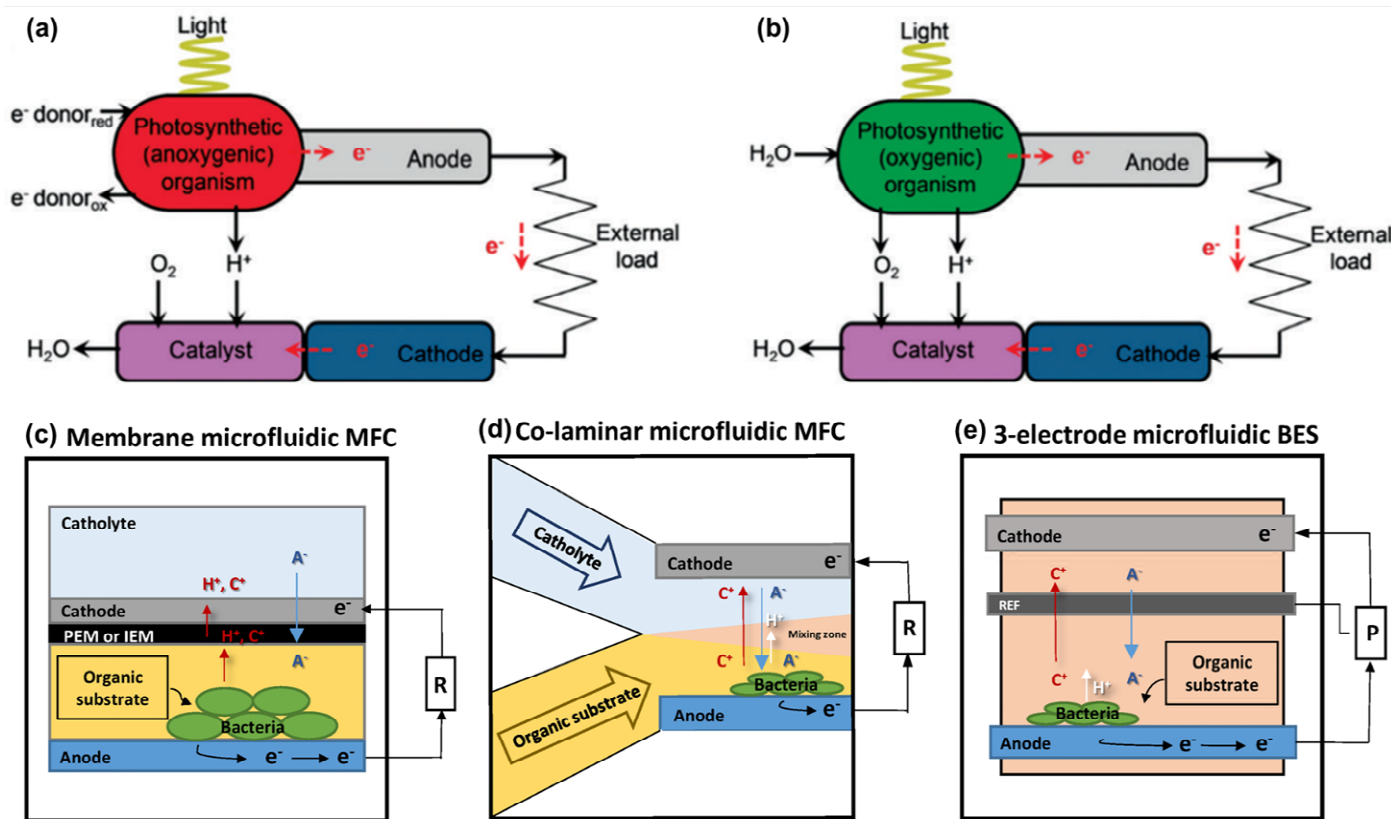


Fig. 11. Graphic illustrations of (a) P-MFCs and (b) cellular bio photovoltaic cells. Reprinted with permission from Ref. [195]. Exhibition of the three kinds of microfluidic BES. (c) low-range membrane MFC, (d) co-laminar electrode microfluidic, (e) BES is incorporating an added reference electrode. Reprinted with permission from Ref. [198].

3.14. Microfluidic BES

Since the last previous years, microfluidic devices have significantly improved, explain the primary area of bacterial EET and introduce innovative optimization approaches for improving and designing the latest BES systems. Microfluidics will undoubtedly remain to give to various interesting decisive fields of analysis. While studying microfluidics to promote the research and development of the BES area, this is essential to concentrate upon the powers that microfluidics systems maintain.

Within a standard MFC, electroactive microbes (**Fig. 11 (c)**) treat the anode. The microbes oxidize natural substrates into the electrolyte and contribute electrons on the anode exterior immediately or lengthily. The electrons are transported toward the cathode via an outer shield; whither an abiotically catalyzed reaction electrochemically diminishes oxidants. A proton exchange membrane (PEM) or, more widely, an ionic exchange membrane (IEM) divides the two assemblies for preventing the mixing of their solution, during ions with protons may yet be switched within them. The arrangement is identical during a membrane microfluidic MFC; besides this, the electrodes are usually provided immediately, including microfabrication systems, to meet the microreactor system's range. As two identical channels could provide the cathodic and the anodic electrolytes to be introduced individually during a co-laminar microfluidic MFC, the fluid flow dynamics could limit the two electrolytes by mixing (**Fig. 11 (d)**). The appearance of a reference electrode determines the three-electrode microfluidic BES apparatus to normalize the effects noted by the working electrode (**Fig. 11 (e)**). Such a practice provides secure data to be received and electrochemical reaction kinetics reported separately at anodes and cathodes. Up to now, just rare classes of three-electrode arrangements have been developed within microfluidic BESs [198].

4. Possibilities for employing CO₂ in bio-factories

There are several different opportunities (as explained earlier) to usage CO₂ in biofactories to produce added-value compounds [199]. Mainly, the utilization of CO₂ is not triggered with the object toward reducing the greenhouse gas (GHG) emissions but through the opportunity to have substitute methods, outcomes by distinct features, or to utilize carbon causes on economic value. Generally, the purpose of these methods is not to combine sustainable energy inside the biobased product series. The product of succinic acid by glucose plus CO₂ with biotechnological methods [200]; are operating at the pilot/demo range composition of bio-succinic acid (C₄H₆O₄). C₄H₆O₄ may be hydrogenated into butanediol/tetrahydrofuran. The applicability of succinic acid is supposed to improve shortly, including plasticizers, polyurethanes, bioplastics, and chemical intermediates.

The generation concerning acetone by CO₂ plus H₂ with enzymatic ways (Evonik method); acetone, mutually including NH₃, CH₄, and CH₃OH are feedstocks to the Evonik method for generating methyl methacrylate (MMA) by acetone cyanohydrin [201]. So, in this section, the generic approach to producing 100 % fossil-fuel-free MMA has been discussed. Unlike C₄H₆O₄, the reaction and the whole generation of “renewable” MMA are yet far away from applicability. The composition of valuable compounds by the reaction of CO₂ [202] and a few of the biorefineries' derivatives are explained by Alper and Yuksel [203]. An enormous number of inorganics, organics, and metal reactants have been produced to different chemical transformations of CO₂. The accomplishments above the preceding five years are especially remarkable. The construction of cyclic carbonates (CCs) or polycarbonates, epoxides and carboxylation effects by CO₂, mitigation of CO₂, including another exciting reaction, have also been produced and examined broadly (**Fig. 12**) [204].

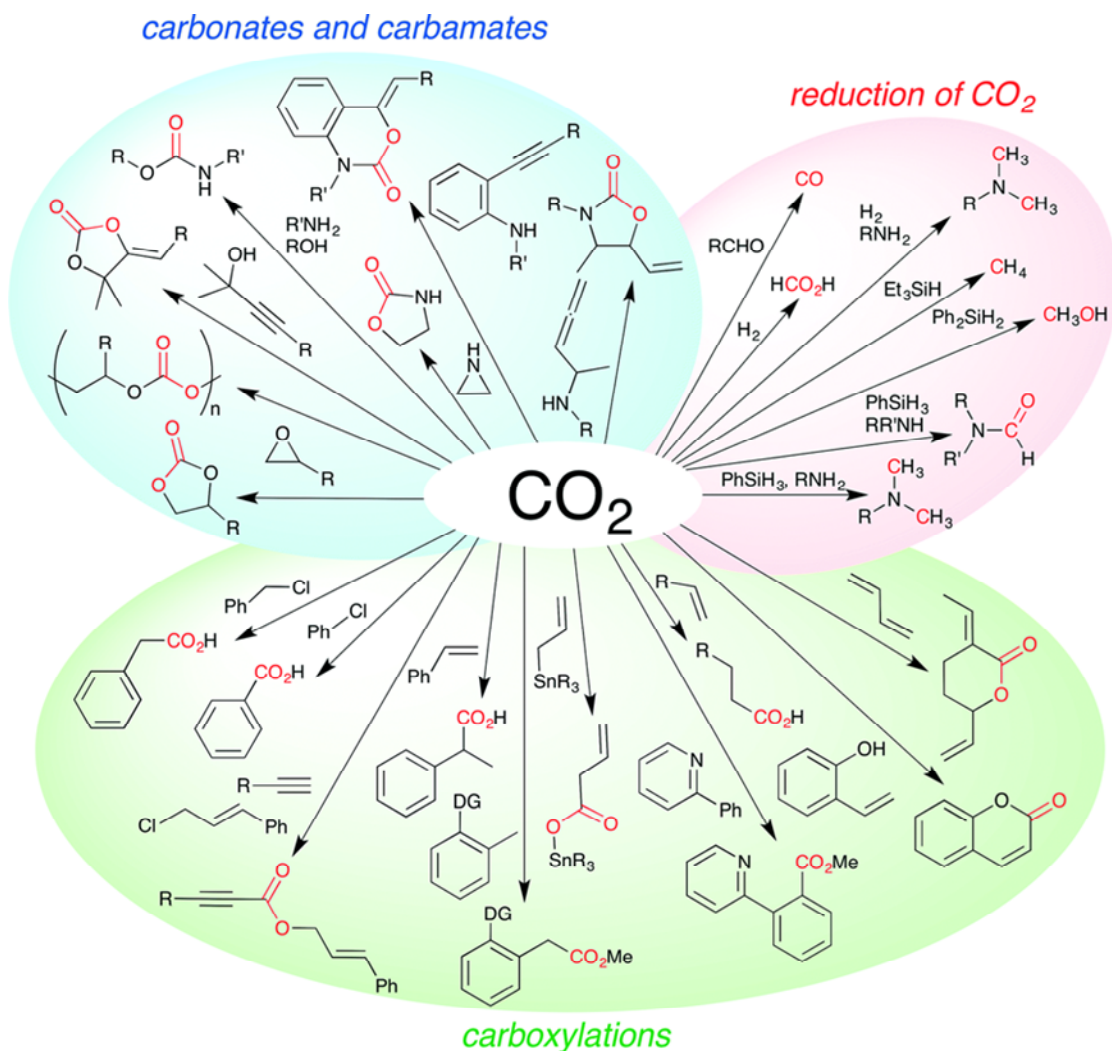


Fig. 12. Illustrations of the chemical fascination of CO₂. Reprinted with permission from Ref. [204].

Though the carboxylation reaction's application to make a broad class of compounds is identified, less data is possible for the uninterrupted usage of derivatives by biobased products originating from activated materials.

4.1. The amalgamation of cyclical carbonates by CO₂ and epoxides

CCs have been extensively utilized essentially (i) feedstocks toward polycarbonates, (ii) electrolytes into Li-ion dependent batteries, (iii) aprotic polar solutions, and (iv) fuel preservatives. While CCs have been integrated with the phosgene process, this approach has disadvantages, such as applying highly poisonous gas (phosgene), the production of

HCl (derivative) and wastewater comprising dichloromethane. While the phosgene process may create economical CCs at a wide range, the construction of environmentally favorable practices is needed, for example, the catalytic approach employing CO₂ and epoxides below moderate forms (**Fig. 13a**). **Fig. 13(b)** shows the metalloporphyrin networks of active reactants. A co-catalyst can be essential toward the nucleophilic ring-opening concerning the epoxide. Following the pioneering investigations by numerous metalloporphyrin compounds, for example, Co [207-209], or Sn [210] as a metal focus, have been established.

4.2. Carboxylation by CO₂

Besides the ring-opening effects of epoxides, including CO₂, considerable work has been done to promote catalytic transformations of CO₂ within other valuable aggregates. Tsuji *et al.* [211] produced the nickel-catalyzed carboxylation of aryl and alkenyl chlorides, including CO₂ and reaction tool at ambient condition (**Fig. 13 (c & d)**). This unique catalytic method is beneficial for aryl bromides, tosylates, including triflates. Mn⁰ degrades the Ni (II) compound toward Ni (0) kinds against proceeding. The following oxidative summation of aryl chloride provides a Ni (II) transitional. Ni (II) is later degraded with the Mn (0)/Et₄NI method to provide a Ni (I) standard. This electron donor Ni (I) sorts reacts by CO₂ to provide the carboxylatonickel transitional, which is demoted by Mn (0) to provide the similar Mn carboxylate, including the restoration of the Ni (0) synergist.

4.3. Reduction of CO₂

The conversion of CO₂ may generate CO, HCOOH, CH₃OH, CH₄, and so on. Newly, catalytic conversions of CO₂ have been investigated vigorously. Zhang and Gu obtained the N-heterocyclic carbene (NHC) conversion of CO₂ to CO utilizing aromatic aldehydes as reductants (**Fig. 13(e)**) [212]. It has presumed that carbonate and/or anhydride could be recovered by the reaction among cinnamaldehyde and imidazolium-carboxylate (NHC-CO₂), and then indeed, CO₂ acted being an oxidant. An active rotation path is given in **figure 13(f)**. (i) NHC responds by CO₂ for producing NHC-CO₂. (ii) The carboxylate strikes the formyl assembly, creating a potential transitional. (iii) Deprotonation effects within the NHC-CO network and a carboxylate salt through the C-O chain breaking. (iv) The NHC-CO compound discharges CO by the restoration of the NHC.

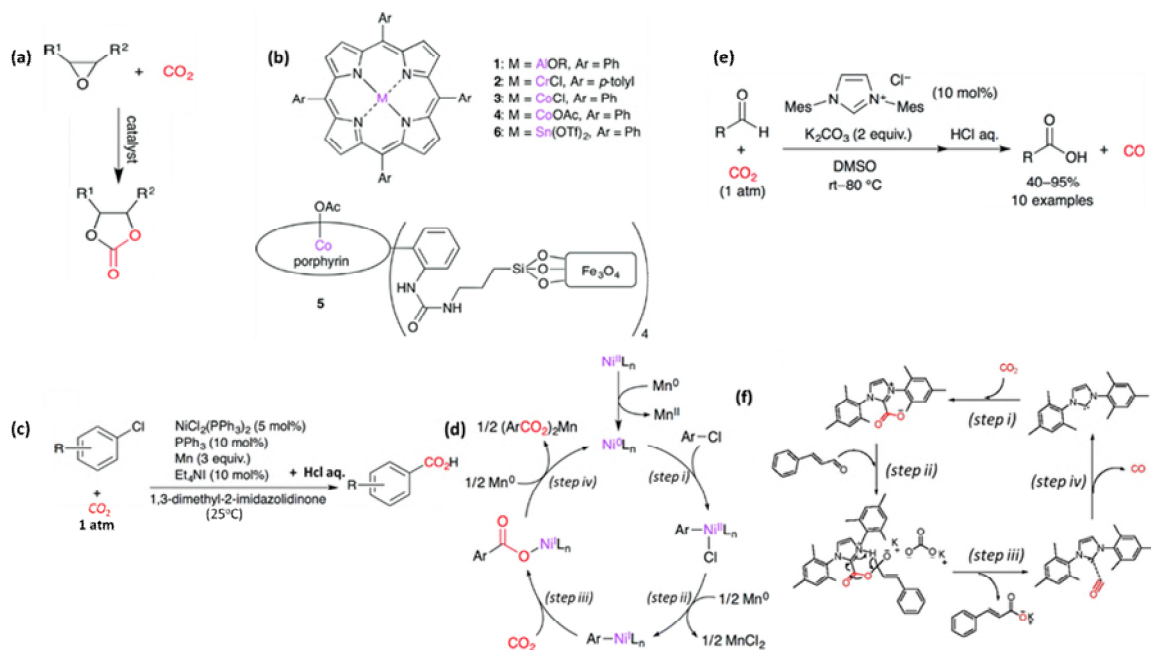


Fig. 13. (a) The combination of CCs by epoxides and CO_2 . (b) Metalloporphyrin catalysts 1-6. Reprinted with permission from Ref. [204]. (c) Nickel-catalyzed carboxylation and (d) reasonable reaction pathways. Reprinted with permission from Ref. [211]. (e) The NHC-catalyzed and (f) reaction mechanism reduction of CO_2 . Reprinted with permission from Ref. [212].

These cases provide a sight of the various opportunities possible to combine the consumption of CO_2 in biofactories, including the option that depends upon the particular business objectives to the biofactory.

5. Challenges and future prospects

Though several efforts have been made, MEC has not reached practical purpose, essentially examining the lab range. Furthermore, few other different technologies of this hybrid method (e.g., phosphate extraction based-high scale-initiated residue method) are also not implemented in the real world. Therefore, comparing up this combination way would be necessary. It needs to enhance this hybrid system's capacity and processing capability to a full-scale level and allowing proper energy restoration. Consequently, the most critical hurdle towards an experimental condition is how to optimize the practice size and concurrently support the fulfilment of every individual. MECs that can perform

essential roles within this hybrid method encounter some particular restrictions in full-scale utilization:

1. A long-distance within electrodes and pH inclination could drive to high intrinsic resistances during a large-range reactor.
2. High initial prices of charge electrodes and membrane substances also restrict its use.
3. There is still a complete understanding of microbiology into MECs, including fermentative bacteria, exoelectrogens and biocathode microorganisms.

With additional scheming MEC reactors, producing new substances, profoundly generous and appropriating command above microbiology is required to discuss these difficulties.

Since the previous decade, the investigation and advancement within ecological electrochemistry, the functionality for microbial electrochemical skills has grown intensely by adapted and enhanced production. Also, BES confers the ability to retrieve additional essential outcomes (H₂, power, etc.), for example, heavy metals, nutrients, manufactured substances, and volatile fuels. As explained in this review article, BES offers attractive possibilities for resource improvement; however, few biotechnological issues and commercial difficulties are still resolved. It may be executed on a wide range and acknowledged as an appealing option for the currently used water processing approaches.

The undesirable energy scale of existing used water processing operations may be modified if BES is employed to improve the energy by organic carbon, including phototrophic techniques, are executed toward nutrients and additional derivatives recovery [213]. Depending on the LCA research [214,215], MFC's performance must be developed with higher energy build as a commercially practical and competing solution similar to presenting anaerobic processing techniques. This mainly needs consideration in producing the cost-effective and active electrode materials to stimulate the O₂ conversion reaction at the cathode toward overcoming the cathodic overpotential to sustain advanced current densities.

After including the current scenario into biorefineries and possibly current patterns (e.g., olefin and standard/functional compound composition), the opportunities to combine CO₂ utilization in the bioeconomy were introduced to mark the options, also by evidence concerning the cost-effectiveness of a solar energy alliance. Lastly, the possibilities for applying CO₂ in biofactories plus the openings for technical cooperation were discussed. There is such a growing prospect for biobased generation, by innovative possibilities allowed through the combined utilization of CO₂ and renewable energy to drive sustainable, including economical carbon bioeconomy. Some machinery must be extended to facilitate the prospect and authoritative guidance, and purposes should understand the opportunities given. However, this is a clear way to support replacing fossil fuels to generate a low-carbon community.

6. Conclusion

BESs have been extensively studied towards electrical energy regeneration, bioremediation, the conventional compound composition, including newly, to valuable resource improvement. However, the recognition of the multi-disciplinary features associated with MES techniques yet on its start. Sustainability problems of biorefineries employing first-generation raw material observed the point of synergy among other regularities (for example, food and water composition). This review focused on BES analysis for energy efficiency and valuable product recoveries such as hydrogen evaluation, acetate, recovery of heavy metals, and nutrient recovery. Further, wastewater processing method, novel electrodes materials used in BES, BESs-based desalination and wastewater treatment, recent BES architecture and designs, genetic engineering for enhanced productivity, production of valuable materials have been discussed. Additionally, we have focused on surfactant and hydrogen peroxide, remediation of soil contamination and photosynthetic & microfluidic BES systems. Some other factors to mitigate the CO₂ towards the value-added products also discussed the advantages of H₂ compared to the other derivatives.

Acknowledgement: The authors gratefully acknowledge the financial support of the National Natural Science Foundation of China (21962008, 51464028), Yunnan Province

Excellent Youth Fund Project (202001AW070005), Candidate Talents Training Fund of Yunnan Province (2017PY269SQ, 2018HB007), and Yunnan Ten Thousand Talents Plan Young & Elite Talents Project (YNWR-QNBJ-2018-346) Further, we have also acknowledged the support from Department of Chemistry, Maharishi Markandeshwar (Deemed to be University), Mullana, Ambala, Haryana, India.

References:

1. Logan, B.E., Call, D., Cheng, S., Hamelers, H.V.M., Sleutels, T.H.J.A., Jeremiasse, A.W., Rozendal, R.A.: Microbial Electrolysis Cells for High Yield Hydrogen Gas Production from Organic Matter. *Environmental Science & Technology* 42(23) (2008) 8630-8640, doi:10.1021/es801553z.
2. Logan, B.E., Hamelers, B., Rozendal, R., Schröder, U., Keller, J., Freguia, S., Aelterman, P., Verstraete, W., Rabaey, K.: Microbial Fuel Cells: Methodology and Technology. *Environmental Science & Technology* 40(17) (2006) 5181-5192, doi:10.1021/es0605016.
3. Lanzafame, P., Centi, G., Perathoner, S.: Catalysis for biomass and CO₂ use through solar energy: opening new scenarios for a sustainable and low-carbon chemical production. *Chemical Society Reviews* 43(22) (2014) 7562-7580, doi:10.1039/C3CS60396B.
4. Bakshi, B.R.: The path to a sustainable chemical industry: progress and problems. *Current Opinion in Chemical Engineering* 1(1) (2011) 64-68, doi:<https://doi.org/10.1016/j.coche.2011.07.004>.
5. Singh Siwal, S., Zhang, Q., Sun, C., Thakur, S., Kumar Gupta, V., Kumar Thakur, V.: Energy production from steam gasification processes and parameters that contemplate in biomass gasifier – A review. *Bioresource Technology* 297 (2020) 122481, doi:<https://doi.org/10.1016/j.biortech.2019.122481>.
6. Ebikade, E., Athaley, A., Fisher, B., Yang, K., Wu, C., Ierapetritou, M.G., Vlachos, D.G.: The Future is Garbage: Repurposing of Food Waste to an Integrated Biorefinery. *ACS Sustainable Chemistry & Engineering* 8(22) (2020) 8124-8136, doi:10.1021/acssuschemeng.9b07479.
7. Aristizábal-Marulanda, V., Cardona A, C.A.: Experimental production of ethanol, electricity, and furfural under the biorefinery concept. *Chemical Engineering Science* 229 (2021) 116047, doi:<https://doi.org/10.1016/j.ces.2020.116047>.
8. Zhang, Y.H.P.: Production of biocommodities and bioelectricity by cell-free synthetic enzymatic pathway biotransformations: Challenges and opportunities. *Biotechnology and Bioengineering* 105(4) (2010) 663-677, doi:10.1002/bit.22630.
9. Chen, C.-F., Feng, K.-L., Ma, H.-w.: Uncover the interdependent environmental impacts associated with the water-energy-food nexus under resource management strategies. *Resources, Conservation and Recycling* 160 (2020) 104909, doi:<https://doi.org/10.1016/j.resconrec.2020.104909>.
10. Siwal, S.S., Thakur, S., Zhang, Q.B., Thakur, V.K.: Electrocatalysts for electrooxidation of direct alcohol fuel cell: chemistry and applications. *Materials Today Chemistry* 14 (2019) 100182, doi:<https://doi.org/10.1016/j.mtchem.2019.06.004>.
11. Martinez-Hernandez, E., Samsatli, S.: Biorefineries and the food, energy, water nexus—towards a whole systems approach to design and planning. *Current Opinion in Chemical Engineering* 18 (2017) 16-22, doi:<https://doi.org/10.1016/j.coche.2017.08.003>.
12. Zhang, Y.H.P.: Next generation biorefineries will solve the food, biofuels, and environmental trilemma in the energy–food–water nexus. *Energy Science & Engineering* 1(1) (2013) 27-41, doi:10.1002/ese3.2.
13. Yuan, M.-H., Chiueh, P.-T., Lo, S.-L.: Measuring urban food-energy-water nexus sustainability: Finding solutions for cities. *Science of The Total Environment* 752 (2021) 141954, doi:<https://doi.org/10.1016/j.scitotenv.2020.141954>.
14. Liu, Z., Huang, Q., He, C., Wang, C., Wang, Y., Li, K.: Water-energy nexus within urban agglomeration: An assessment framework combining the multiregional input-output

- model, virtual water, and embodied energy. *Resources, Conservation and Recycling* 164 (2021) 105113, doi:<https://doi.org/10.1016/j.resconrec.2020.105113>.
15. Chitra Devi, V., Mothil, S., Sathish Raam, R., Senthilkumar, K.: Thermochemical Conversion and Valorization of Woody Lignocellulosic Biomass in Hydrothermal Media. In: Praveen Kumar, R., Bharathiraja, B., Katak, R., Moholkar, V.S. (eds.) *Biomass Valorization to Bioenergy*. pp. 45-63. Springer Singapore, Singapore (2020)
 16. Aden A, R.M., Ibsen K, Jechura J, Neeves K, Sheehan J et al.: Lignocellulosic biomass ethanol process design and economics utilizing co-current dilute acid prehydrolysis and enzymatic hydrolysis for corn stover. In. Golden, Colo: National Renewable Energy Laboratory, (2002)
 17. Lewis, A.J., Borole, A.P.: Chapter 6.5 - Bio-Electro-Refinery: Conversion, Sustainability, and Policy. In: Mohan, S.V., Varjani, S., Pandey, A. (eds.) *Microbial Electrochemical Technology*. pp. 1059-1085. Elsevier, (2019)
 18. de Farias Silva, C.E., Barbera, E., Bertuccio, A.: Chapter 17 - Biorefinery as a Promising Approach to Promote Ethanol Industry From Microalgae and Cyanobacteria. In: Ray, R.C., Ramachandran, S. (eds.) *Bioethanol Production from Food Crops*. pp. 343-359. Academic Press, (2019)
 19. Siwal, S.S., Yang, W., Zhang, Q.: Recent progress of precious-metal-free electrocatalysts for efficient water oxidation in acidic media. *Journal of Energy Chemistry* 51 (2020) 113-133, doi:<https://doi.org/10.1016/j.jechem.2020.03.079>.
 20. Lewis, A.J., Ren, S., Ye, X., Kim, P., Labbe, N., Borole, A.P.: Hydrogen production from switchgrass via an integrated pyrolysis–microbial electrolysis process. *Bioresource Technology* 195 (2015) 231-241, doi:<https://doi.org/10.1016/j.biortech.2015.06.085>.
 21. Lu, L., Ren, N., Xing, D., Logan, B.E.: Hydrogen production with effluent from an ethanol–H₂-coproducing fermentation reactor using a single-chamber microbial electrolysis cell. *Biosensors and Bioelectronics* 24(10) (2009) 3055-3060, doi:<https://doi.org/10.1016/j.bios.2009.03.024>.
 22. Yu, Z., Leng, X., Zhao, S., Ji, J., Zhou, T., Khan, A., Kakde, A., Liu, P., Li, X.: A review on the applications of microbial electrolysis cells in anaerobic digestion. *Bioresource Technology* 255 (2018) 340-348, doi:<https://doi.org/10.1016/j.biortech.2018.02.003>.
 23. Rabaey, K., Verstraete, W.: Microbial fuel cells: novel biotechnology for energy generation. *Trends in Biotechnology* 23(6) (2005) 291-298, doi:<https://doi.org/10.1016/j.tibtech.2005.04.008>.
 24. Wang, B., Zhang, H., Yang, Y., Xu, M.: Diffusion and filamentous bacteria jointly govern the spatiotemporal process of sulfide removal in sediment microbial fuel cells. *Chemical Engineering Journal* 405 (2021) 126680, doi:<https://doi.org/10.1016/j.cej.2020.126680>.
 25. Kadir, A., Jain, P., Lai, B., Kalil, M.S., Kondaveeti, S., Alabbosh, K.F.S., Abu-Reesh, I.M., Mohanakrishna, G.: Biorefinery perspectives of microbial electrolysis cells (MECs) for hydrogen and valuable chemicals production through wastewater treatment. *Biofuel Research Journal* 7(1) (2020) 1128-1142, doi:10.18331/BRJ2020.7.1.5.
 26. Pant, D., Van Bogaert, G., Diels, L., Vanbroekhoven, K.: A review of the substrates used in microbial fuel cells (MFCs) for sustainable energy production. *Bioresource Technology* 101(6) (2010) 1533-1543, doi:<https://doi.org/10.1016/j.biortech.2009.10.017>.
 27. Borole, A.P., Mielenz, J.R., Vishnivetskaya, T.A., Hamilton, C.Y.: Controlling accumulation of fermentation inhibitors in biorefinery recycle water using microbial fuel cells. *Biotechnology for Biofuels* 2(1) (2009) 7, doi:10.1186/1754-6834-2-7.
 28. Catal, T., Li, K., Bermek, H., Liu, H.: Electricity production from twelve monosaccharides using microbial fuel cells. *Journal of Power Sources* 175(1) (2008) 196-200, doi:<https://doi.org/10.1016/j.jpowsour.2007.09.083>.

29. Niessen, J., Schröder, U., Scholz, F.: Exploiting complex carbohydrates for microbial electricity generation – a bacterial fuel cell operating on starch. *Electrochemistry Communications* 6(9) (2004) 955-958, doi:<https://doi.org/10.1016/j.elecom.2004.07.010>.
30. Islam, K.M.N., Kenway, S.J., Renouf, M.A., Lam, K.L., Wiedmann, T.: A review of the water-related energy consumption of the food system in nexus studies. *Journal of Cleaner Production* 279 (2021) 123414, doi:<https://doi.org/10.1016/j.jclepro.2020.123414>.
31. Liu, W., Huang, S., Zhou, A., Zhou, G., Ren, N., Wang, A., Zhuang, G.: Hydrogen generation in microbial electrolysis cell feeding with fermentation liquid of waste activated sludge. *International Journal of Hydrogen Energy* 37(18) (2012) 13859-13864, doi:<https://doi.org/10.1016/j.ijhydene.2012.04.090>.
32. Ren, B., Wang, T., Zhao, Y.: Two-stage hybrid constructed wetland-microbial fuel cells for swine wastewater treatment and bioenergy generation. *Chemosphere* 268 (2021) 128803, doi:<https://doi.org/10.1016/j.chemosphere.2020.128803>.
33. Wan, L.-L., Li, X.-J., Zang, G.-L., Wang, X., Zhang, Y.-Y., Zhou, Q.-X.: A solar assisted microbial electrolysis cell for hydrogen production driven by a microbial fuel cell. *RSC Advances* 5(100) (2015) 82276-82281, doi:10.1039/C5RA16919D.
34. Srivastava, P., Yadav, A.K., Garaniya, V., Lewis, T., Abbassi, R., Khan, S.J.: Electrode dependent anaerobic ammonium oxidation in microbial fuel cell integrated hybrid constructed wetlands: A new process. *Science of The Total Environment* 698 (2020) 134248, doi:<https://doi.org/10.1016/j.scitotenv.2019.134248>.
35. Lu, L., Ren, N., Zhao, X., Wang, H., Wu, D., Xing, D.: Hydrogen production, methanogen inhibition and microbial community structures in psychrophilic single-chamber microbial electrolysis cells. *Energy & Environmental Science* 4(4) (2011) 1329-1336, doi:10.1039/C0EE00588F.
36. Chae, K.-J., Choi, M.-J., Kim, K.-Y., Ajayi, F.F., Chang, I.-S., Kim, I.S.: A Solar-Powered Microbial Electrolysis Cell with a Platinum Catalyst-Free Cathode To Produce Hydrogen. *Environmental Science & Technology* 43(24) (2009) 9525-9530, doi:10.1021/es9022317.
37. Sun, M., Sheng, G.-P., Zhang, L., Xia, C.-R., Mu, Z.-X., Liu, X.-W., Wang, H.-L., Yu, H.-Q., Qi, R., Yu, T., Yang, M.: An MEC-MFC-Coupled System for Biohydrogen Production from Acetate. *Environmental Science & Technology* 42(21) (2008) 8095-8100, doi:10.1021/es801513c.
38. Lenin Babu, M., Venkata Subhash, G., Sarma, P.N., Venkata Mohan, S.: Bio-electrolytic conversion of acidogenic effluents to biohydrogen: An integration strategy for higher substrate conversion and product recovery. *Bioresource Technology* 133 (2013) 322-331, doi:<https://doi.org/10.1016/j.biortech.2013.01.029>.
39. Lalaurette, E., Thammannagowda, S., Mohagheghi, A., Maness, P.-C., Logan, B.E.: Hydrogen production from cellulose in a two-stage process combining fermentation and electrohydrogenesis. *International Journal of Hydrogen Energy* 34(15) (2009) 6201-6210, doi:<https://doi.org/10.1016/j.ijhydene.2009.05.112>.
40. Yuan, H., Lu, Y., Abu-Reesh, I.M., He, Z.: Bioelectrochemical production of hydrogen in an innovative pressure-retarded osmosis/microbial electrolysis cell system: experiments and modeling. *Biotechnology for Biofuels* 8(1) (2015) 116, doi:10.1186/s13068-015-0305-0.
41. Lee, M.-Y., Kim, K.-Y., Yang, E., Kim, I.S.: Evaluation of hydrogen production and internal resistance in forward osmosis membrane integrated microbial electrolysis cells. *Bioresource Technology* 187 (2015) 106-112, doi:<https://doi.org/10.1016/j.biortech.2015.03.079>.
42. Bo, T., Zhu, X., Zhang, L., Tao, Y., He, X., Li, D., Yan, Z.: A new upgraded biogas production process: Coupling microbial electrolysis cell and anaerobic digestion in single-chamber,

- barrel-shape stainless steel reactor. *Electrochemistry Communications* 45 (2014) 67-70, doi:<https://doi.org/10.1016/j.elecom.2014.05.026>.
43. Cai, W., Han, T., Guo, Z., Varrone, C., Wang, A., Liu, W.: Methane production enhancement by an independent cathode in integrated anaerobic reactor with microbial electrolysis. *Bioresource Technology* 208 (2016) 13-18, doi:<https://doi.org/10.1016/j.biortech.2016.02.028>.
 44. Zhi, Z., Pan, Y., Lu, X., Zhen, G., Zhao, Y., Zhu, X., Xiong, J., Zhao, T.: Electrically regulating co-fermentation of sewage sludge and food waste towards promoting biomethane production and mass reduction. *Bioresource Technology* 279 (2019) 218-227, doi:<https://doi.org/10.1016/j.biortech.2019.01.142>.
 45. Cerrillo, M., Viñas, M., Bonmatí, A.: Overcoming organic and nitrogen overload in thermophilic anaerobic digestion of pig slurry by coupling a microbial electrolysis cell. *Bioresource Technology* 216 (2016) 362-372, doi:<https://doi.org/10.1016/j.biortech.2016.05.085>.
 46. Mehanna, M., Kiely, P.D., Call, D.F., Logan, B.E.: Microbial Electrodialysis Cell for Simultaneous Water Desalination and Hydrogen Gas Production. *Environmental Science & Technology* 44(24) (2010) 9578-9583, doi:10.1021/es1025646.
 47. Ditzig, J., Liu, H., Logan, B.E.: Production of hydrogen from domestic wastewater using a bioelectrochemically assisted microbial reactor (BEAMR). *International Journal of Hydrogen Energy* 32(13) (2007) 2296-2304, doi:<https://doi.org/10.1016/j.ijhydene.2007.02.035>.
 48. Wang, A., Sun, D., Cao, G., Wang, H., Ren, N., Wu, W.-M., Logan, B.E.: Integrated hydrogen production process from cellulose by combining dark fermentation, microbial fuel cells, and a microbial electrolysis cell. *Bioresource Technology* 102(5) (2011) 4137-4143, doi:<https://doi.org/10.1016/j.biortech.2010.10.137>.
 49. Dhar, B.R., Elbeshbishy, E., Hafez, H., Lee, H.-S.: Hydrogen production from sugar beet juice using an integrated biohydrogen process of dark fermentation and microbial electrolysis cell. *Bioresource Technology* 198 (2015) 223-230, doi:<https://doi.org/10.1016/j.biortech.2015.08.048>.
 50. Wagner, R.C., Regan, J.M., Oh, S.-E., Zuo, Y., Logan, B.E.: Hydrogen and methane production from swine wastewater using microbial electrolysis cells. *Water Research* 43(5) (2009) 1480-1488, doi:<https://doi.org/10.1016/j.watres.2008.12.037>.
 51. Ajayi, F.F., Kim, K.-Y., Chae, K.-J., Choi, M.-J., Chang, I.S., Kim, I.S.: Optimization studies of biohydrogen production in a coupled microbial electrolysis–dye sensitized solar cell system. *Photochemical & Photobiological Sciences* 9(3) (2010) 349-356, doi:10.1039/B9PP00097F.
 52. Modestra, J.A., Babu, M.L., Mohan, S.V.: Electro-fermentation of real-field acidogenic spent wash effluents for additional biohydrogen production with simultaneous treatment in a microbial electrolysis cell. *Separation and Purification Technology* 150 (2015) 308-315, doi:<https://doi.org/10.1016/j.seppur.2015.05.043>.
 53. Tenca, A., Cusick, R.D., Schievano, A., Oberti, R., Logan, B.E.: Evaluation of low cost cathode materials for treatment of industrial and food processing wastewater using microbial electrolysis cells. *International Journal of Hydrogen Energy* 38(4) (2013) 1859-1865, doi:<https://doi.org/10.1016/j.ijhydene.2012.11.103>.
 54. Tartakovsky, B., Mehta, P., Bourque, J.S., Guiot, S.R.: Electrolysis-enhanced anaerobic digestion of wastewater. *Bioresource Technology* 102(10) (2011) 5685-5691, doi:<https://doi.org/10.1016/j.biortech.2011.02.097>.
 55. Li, X.-H., Liang, D.-W., Bai, Y.-X., Fan, Y.-T., Hou, H.-W.: Enhanced H₂ production from corn stalk by integrating dark fermentation and single chamber microbial electrolysis cells with

- double anode arrangement. *International Journal of Hydrogen Energy* 39(17) (2014) 8977-8982, doi:<https://doi.org/10.1016/j.ijhydene.2014.03.065>.
56. Kiely, P.D., Cusick, R., Call, D.F., Selembo, P.A., Regan, J.M., Logan, B.E.: Anode microbial communities produced by changing from microbial fuel cell to microbial electrolysis cell operation using two different wastewaters. *Bioresource Technology* 102(1) (2011) 388-394, doi:<https://doi.org/10.1016/j.biortech.2010.05.019>.
 57. Escapa, A., Lobato, A., García, D.M., Morán, A.: Hydrogen production and COD elimination rate in a continuous microbial electrolysis cell: The influence of hydraulic retention time and applied voltage. *Environmental Progress & Sustainable Energy* 32(2) (2013) 263-268, doi:<https://doi.org/10.1002/ep.11619>.
 58. Chen, Y., Chen, M., Shen, N., Zeng, R.J.: H₂ production by the thermoelectric microconverter coupled with microbial electrolysis cell. *International Journal of Hydrogen Energy* 41(48) (2016) 22760-22768, doi:<https://doi.org/10.1016/j.ijhydene.2016.09.209>.
 59. Cusick, R.D., Bryan, B., Parker, D.S., Merrill, M.D., Mehanna, M., Kiely, P.D., Liu, G., Logan, B.E.: Performance of a pilot-scale continuous flow microbial electrolysis cell fed winery wastewater. *Applied Microbiology and Biotechnology* 89(6) (2011) 2053-2063, doi:10.1007/s00253-011-3130-9.
 60. Montpart, N., Rago, L., Baeza, J.A., Guisasola, A.: Hydrogen production in single chamber microbial electrolysis cells with different complex substrates. *Water Research* 68 (2015) 601-615, doi:<https://doi.org/10.1016/j.watres.2014.10.026>.
 61. Das, S., Das, S., Das, I., Ghangrekar, M.M.: Application of bioelectrochemical systems for carbon dioxide sequestration and concomitant valuable recovery: A review. *Materials Science for Energy Technologies* 2(3) (2019) 687-696, doi:<https://doi.org/10.1016/j.mset.2019.08.003>.
 62. Sayed, E.T., Shehata, N., Abdelkareem, M.A., Atieh, M.A.: Recent progress in environmentally friendly bio-electrochemical devices for simultaneous water desalination and wastewater treatment. *Science of The Total Environment* 748 (2020) 141046, doi:<https://doi.org/10.1016/j.scitotenv.2020.141046>.
 63. Almeida, J.R.M., Fávoro, L.C.L., Quirino, B.F.: Biodiesel biorefinery: opportunities and challenges for microbial production of fuels and chemicals from glycerol waste. *Biotechnology for Biofuels* 5(1) (2012) 48, doi:10.1186/1754-6834-5-48.
 64. Lopez-Hidalgo, A.M., Magaña, G., Rodriguez, F., De Leon-Rodriguez, A., Sanchez, A.: Co-production of ethanol-hydrogen by genetically engineered *Escherichia coli* in sustainable biorefineries for lignocellulosic ethanol production. *Chemical Engineering Journal* 406 (2021) 126829, doi:<https://doi.org/10.1016/j.cej.2020.126829>.
 65. Takkellapati, S., Li, T., Gonzalez, M.A.: An overview of biorefinery-derived platform chemicals from a cellulose and hemicellulose biorefinery. *Clean Technologies and Environmental Policy* 20(7) (2018) 1615-1630, doi:10.1007/s10098-018-1568-5.
 66. Dawson, I.K., Park, S.E., Attwood, S.J., Jamnadass, R., Powell, W., Sunderland, T., Carsan, S.: Contributions of biodiversity to the sustainable intensification of food production. *Global Food Security* 21 (2019) 23-37, doi:<https://doi.org/10.1016/j.gfs.2019.07.002>.
 67. Demuner, I.F., Colodette, J.L., Demuner, A.J., Jardim, C.M.: Biorefinery Review: Wide-Reaching Products Through Kraft Lignin. *BioResources*; Vol 14, No 3 (2019) (2019),
 68. Demirbas, A.: Inexpensive oil and fats feedstocks for production of biodiesel. *Energy Education Science and Technology Part A: Energy Science and Research* 23 (2009) 1-13, doi:10.1007/978-1-84882-511-6_1.
 69. Kohli, K., Prajapati, R., Sharma, B.K.: Bio-Based Chemicals from Renewable Biomass for Integrated Biorefineries. *Energies* 12(2) (2019), doi:10.3390/en12020233.

70. Sikarwar, V.S., Zhao, M., Fennell, P.S., Shah, N., Anthony, E.J.: Progress in biofuel production from gasification. *Progress in Energy and Combustion Science* 61 (2017) 189-248, doi:<https://doi.org/10.1016/j.pecs.2017.04.001>.
71. Rodriguez Correa, C., Kruse, A.: Supercritical water gasification of biomass for hydrogen production – Review. *The Journal of Supercritical Fluids* 133 (2018) 573-590, doi:<https://doi.org/10.1016/j.supflu.2017.09.019>.
72. Olabi, A.G., Wilberforce, T., Abdelkareem, M.A.: Fuel cell application in the automotive industry and future perspective. *Energy* 214 (2021) 118955, doi:<https://doi.org/10.1016/j.energy.2020.118955>.
73. Lin, C.-Y., Lu, C.: Development perspectives of promising lignocellulose feedstocks for production of advanced generation biofuels: A review. *Renewable and Sustainable Energy Reviews* 136 (2021) 110445, doi:<https://doi.org/10.1016/j.rser.2020.110445>.
74. Araújo, K., Mahajan, D., Kerr, R., Silva, M.d.: Global Biofuels at the Crossroads: An Overview of Technical, Policy, and Investment Complexities in the Sustainability of Biofuel Development. *Agriculture* 7(4) (2017), doi:10.3390/agriculture7040032.
75. Ghatak, H.R.: Biorefineries from the perspective of sustainability: Feedstocks, products, and processes. *Renewable and Sustainable Energy Reviews* 15(8) (2011) 4042-4052, doi:<https://doi.org/10.1016/j.rser.2011.07.034>.
76. Nouri, M.: Potentials and challenges of date pits as alternative environmental clean-up ingredients. *Biomass Conversion and Biorefinery* (2021), doi:10.1007/s13399-020-01215-w.
77. Paul, V., Rai, S., Tripathi, A.D., Rai, D.C., Agarwal, A.: Impact of Fermentation Types on Enzymes Used for Biofuels Production. In: Srivastava, N., Srivastava, M., Mishra, P.K., Gupta, V.K. (eds.) *Bioprocessing for Biofuel Production: Strategies to Improve Process Parameters*. pp. 1-27. Springer Singapore, Singapore (2021)
78. Cherubini, F., Jungmeier, G., Wellisch, M., Willke, T., Skiadas, I., Van Ree, R., de Jong, E.: Toward a common classification approach for biorefinery systems. *Biofuels, Bioproducts and Biorefining* 3(5) (2009) 534-546, doi:<https://doi.org/10.1002/bbb.172>.
79. Sadhukhan, J., Ng, K.S., Martinez, E.: *Biorefineries and Chemical Processes: Design, Integration and Sustainability Analysis*. (2014)
80. Azapagic, A.: Sustainability considerations for integrated biorefineries. *Trends in biotechnology* 32 (2014) 1-4, doi:10.1016/j.tibtech.2013.10.009.
81. Zhang, P., Zhang, L., Chang, Y., Xu, M., Hao, Y., Liang, S., Liu, G., Yang, Z., Wang, C.: Food-energy-water (FEW) nexus for urban sustainability: A comprehensive review. *Resources, Conservation and Recycling* 142 (2019) 215-224, doi:<https://doi.org/10.1016/j.resconrec.2018.11.018>.
82. Kibler, K.M., Reinhart, D., Hawkins, C., Motlagh, A.M., Wright, J.: Food waste and the food-energy-water nexus: A review of food waste management alternatives. *Waste Management* 74 (2018) 52-62, doi:<https://doi.org/10.1016/j.wasman.2018.01.014>.
83. Loh, S.K., James, S., Ngatiman, M., Cheong, K.Y., Choo, Y.M., Lim, W.S.: Enhancement of palm oil refinery waste – Spent bleaching earth (SBE) into bio organic fertilizer and their effects on crop biomass growth. *Industrial Crops and Products* 49 (2013) 775-781, doi:<https://doi.org/10.1016/j.indcrop.2013.06.016>.
84. Barik, D.: Chapter 3 - Energy Extraction From Toxic Waste Originating From Food Processing Industries. In: Barik, D. (ed.) *Energy from Toxic Organic Waste for Heat and Power Generation*. pp. 17-42. Woodhead Publishing, (2019)

85. Yang, H., Zhou, Y., Liu, J.: Land and water requirements of biofuel and implications for food supply and the environment in China. *Energy Policy* 37(5) (2009) 1876-1885, doi:<https://doi.org/10.1016/j.enpol.2009.01.035>.
86. Dale, B.E., Bals, B.D., Kim, S., Franki, P.: Biofuels Done Right: Land Efficient Animal Feeds Enable Large Environmental and Energy Benefits. *Environmental Science & Technology* 44(22) (2010) 8385-8389, doi:10.1021/es101864b.
87. Cherubini, F., Bird, N.D., Cowie, A., Jungmeier, G., Schlamadinger, B., Woess-Gallasch, S.: Energy- and greenhouse gas-based LCA of biofuel and bioenergy systems: Key issues, ranges and recommendations. *Resources, Conservation and Recycling* 53(8) (2009) 434-447, doi:<https://doi.org/10.1016/j.resconrec.2009.03.013>.
88. Martinez-Hernandez, E., Ibrahim, M.H., Leach, M., Sinclair, P., Campbell, G.M., Sadhukhan, J.: Environmental sustainability analysis of UK whole-wheat bioethanol and CHP systems. *Biomass and Bioenergy* 50 (2013) 52-64, doi:<https://doi.org/10.1016/j.biombioe.2013.01.001>.
89. Martinez-Hernandez, E., Sadhukhan, J., Campbell, G.M.: Integration of bioethanol as an in-process material in biorefineries using mass pinch analysis. *Applied Energy* 104 (2013) 517-526, doi:<https://doi.org/10.1016/j.apenergy.2012.11.054>.
90. Satchatippavarn, S., Martinez-Hernandez, E., Leung Pah Hang, M.Y., Leach, M., Yang, A.: Urban biorefinery for waste processing. *Chemical Engineering Research and Design* 107 (2016) 81-90, doi:<https://doi.org/10.1016/j.cherd.2015.09.022>.
91. Sadhukhan, J., Ng, K.S., Martinez-Hernandez, E.: Novel integrated mechanical biological chemical treatment (MBCT) systems for the production of levulinic acid from fraction of municipal solid waste: A comprehensive techno-economic analysis. *Bioresource Technology* 215 (2016) 131-143, doi:<https://doi.org/10.1016/j.biortech.2016.04.030>.
92. Yue, D., Pandya, S., You, F.: Integrating Hybrid Life Cycle Assessment with Multiobjective Optimization: A Modeling Framework. *Environmental Science & Technology* 50(3) (2016) 1501-1509, doi:10.1021/acs.est.5b04279.
93. You, F., Tao, L., Graziano, D.J., Snyder, S.W.: Optimal design of sustainable cellulosic biofuel supply chains: Multiobjective optimization coupled with life cycle assessment and input-output analysis. *AIChE Journal* 58(4) (2012) 1157-1180, doi:10.1002/aic.12637.
94. Garcia, D.J., You, F.: The water-energy-food nexus and process systems engineering: A new focus. *Computers & Chemical Engineering* 91 (2016) 49-67, doi:<https://doi.org/10.1016/j.compchemeng.2016.03.003>.
95. Leung Pah Hang, M.Y., Martinez-Hernandez, E., Leach, M., Yang, A.: Designing integrated local production systems: A study on the food-energy-water nexus. *Journal of Cleaner Production* 135 (2016) 1065-1084, doi:<https://doi.org/10.1016/j.jclepro.2016.06.194>.
96. Proctor, K., Tabatabaie, S.M.H., Murthy, G.S.: Gateway to the perspectives of the Food-Energy-Water nexus. *Science of The Total Environment* (2020) 142852, doi:<https://doi.org/10.1016/j.scitotenv.2020.142852>.
97. Hameed, Z., Aslam, M., Khan, Z., Maqsood, K., Atabani, A.E., Ghauri, M., Khurram, M.S., Rehan, M., Nizami, A.-S.: Gasification of municipal solid waste blends with biomass for energy production and resources recovery: Current status, hybrid technologies and innovative prospects. *Renewable and Sustainable Energy Reviews* 136 (2021) 110375, doi:<https://doi.org/10.1016/j.rser.2020.110375>.
98. Kargbo, H., Harris, J.S., Phan, A.N.: "Drop-in" fuel production from biomass: Critical review on techno-economic feasibility and sustainability. *Renewable and Sustainable Energy Reviews* 135 (2021) 110168, doi:<https://doi.org/10.1016/j.rser.2020.110168>.

99. Yaqoob, A.A., Mohamad Ibrahim, M.N., Umar, K., Bhawani, S.A., Khan, A., Asiri, A.M., Khan, M.R., Azam, M., AlAmmari, A.M.: Cellulose Derived Graphene/Polyaniline Nanocomposite Anode for Energy Generation and Bioremediation of Toxic Metals via Benthic Microbial Fuel Cells. *Polymers* 13(1) (2021), doi:10.3390/polym13010135.
100. Chen, W., Liu, Z., Li, Y., Xing, X., Liao, Q., Zhu, X.: Improved electricity generation, coulombic efficiency and microbial community structure of microbial fuel cells using sodium citrate as an effective additive. *Journal of Power Sources* 482 (2021) 228947, doi:<https://doi.org/10.1016/j.jpowsour.2020.228947>.
101. Banu, J.R., Kumar, M.D., Gunasekaran, M., Kumar, G.: Biopolymer production in bio electrochemical system: Literature survey. *Bioresource Technology Reports* 7 (2019) 100283, doi:<https://doi.org/10.1016/j.biteb.2019.100283>.
102. Zhang, J., Liu, Y., Sun, Y., Wang, H., Cao, X., Li, X.: Effect of soil type on heavy metals removal in bioelectrochemical system. *Bioelectrochemistry* 136 (2020) 107596, doi:<https://doi.org/10.1016/j.bioelechem.2020.107596>.
103. Jadhav, D.A., Ghosh Ray, S., Ghangrekar, M.M.: Third generation in bio-electrochemical system research – A systematic review on mechanisms for recovery of valuable by-products from wastewater. *Renewable and Sustainable Energy Reviews* 76 (2017) 1022-1031, doi:<https://doi.org/10.1016/j.rser.2017.03.096>.
104. Wang, H., Ren, Z.J.: Bioelectrochemical metal recovery from wastewater: A review. *Water Research* 66 (2014) 219-232, doi:<https://doi.org/10.1016/j.watres.2014.08.013>.
105. Nancharaiyah, Y.V., Venkata Mohan, S., Lens, P.N.L.: Metals removal and recovery in bioelectrochemical systems: A review. *Bioresource Technology* 195 (2015) 102-114, doi:<https://doi.org/10.1016/j.biortech.2015.06.058>.
106. Scopus data. www.scopus.com.
107. Wang, H., Ren, Z.J.: A comprehensive review of microbial electrochemical systems as a platform technology. *Biotechnology Advances* 31(8) (2013) 1796-1807, doi:<https://doi.org/10.1016/j.biotechadv.2013.10.001>.
108. Yasri, N., Roberts, E.P.L., Gunasekaran, S.: The electrochemical perspective of bioelectrocatalytic activities in microbial electrolysis and microbial fuel cells. *Energy Reports* 5 (2019) 1116-1136, doi:<https://doi.org/10.1016/j.egy.2019.08.007>.
109. Rabaey, K., Girguis, P., Nielsen, L.K.: Metabolic and practical considerations on microbial electrosynthesis. *Current Opinion in Biotechnology* 22(3) (2011) 371-377, doi:<https://doi.org/10.1016/j.copbio.2011.01.010>.
110. Chandrasekhar, K., Kadier, A., Kumar, G., Nastro, R.A., Jeevitha, V.: Challenges in Microbial Fuel Cell and Future Scope. In: Das, D. (ed.) *Microbial Fuel Cell: A Bioelectrochemical System that Converts Waste to Watts*. pp. 483-499. Springer International Publishing, Cham (2018)
111. Chandrasekhar, K., Amulya, K., Venkata Mohan, S.: Solid phase bio-electrofermentation of food waste to harvest value-added products associated with waste remediation. *Waste Management* 45 (2015) 57-65, doi:<https://doi.org/10.1016/j.wasman.2015.06.001>.
112. Xia, C., Zhang, D., Pedrycz, W., Zhu, Y., Guo, Y.: Models for Microbial Fuel Cells: A critical review. *Journal of Power Sources* 373 (2018) 119-131, doi:<https://doi.org/10.1016/j.jpowsour.2017.11.001>.
113. Jeremiasse, A.W., Hamelers, H.V.M., Buisman, C.J.N.: Microbial electrolysis cell with a microbial biocathode. *Bioelectrochemistry* 78(1) (2010) 39-43, doi:<https://doi.org/10.1016/j.bioelechem.2009.05.005>.

114. Sleutels, T.H.J.A., Ter Heijne, A., Buisman, C.J.N., Hamelers, H.V.M.: Bioelectrochemical Systems: An Outlook for Practical Applications. *ChemSusChem* 5(6) (2012) 1012-1019, doi:10.1002/cssc.201100732.
115. Yuan, H., He, Z.: Platinum Group Metal-free Catalysts for Hydrogen Evolution Reaction in Microbial Electrolysis Cells. *The Chemical Record* 17(7) (2017) 641-652, doi:10.1002/tcr.201700007.
116. Zhang, Y., Angelidaki, I.: Microbial electrolysis cells turning to be versatile technology: Recent advances and future challenges. *Water Research* 56 (2014) 11-25, doi:<https://doi.org/10.1016/j.watres.2014.02.031>.
117. Lovley, D.R.: Electromicrobiology. *Annual Review of Microbiology* 66(1) (2012) 391-409, doi:10.1146/annurev-micro-092611-150104.
118. Yossan, S., Xiao, L., Prasertsan, P., He, Z.: Hydrogen production in microbial electrolysis cells: Choice of catholyte. *International Journal of Hydrogen Energy* 38(23) (2013) 9619-9624, doi:<https://doi.org/10.1016/j.ijhydene.2013.05.094>.
119. Heidrich, E.S., Edwards, S.R., Dolfing, J., Cotterill, S.E., Curtis, T.P.: Performance of a pilot scale microbial electrolysis cell fed on domestic wastewater at ambient temperatures for a 12month period. *Bioresource Technology* 173 (2014) 87-95, doi:<https://doi.org/10.1016/j.biortech.2014.09.083>.
120. Awasthi, M.K., Sarsaiya, S., Wainaina, S., Rajendran, K., Awasthi, S.K., Liu, T., Duan, Y., Jain, A., Sindhu, R., Binod, P., Pandey, A., Zhang, Z., Taherzadeh, M.J.: Techno-economics and life-cycle assessment of biological and thermochemical treatment of bio-waste. *Renewable and Sustainable Energy Reviews* 144 (2021) 110837, doi:<https://doi.org/10.1016/j.rser.2021.110837>.
121. Kasprzyk-Hordern, B., Ziótek, M., Nawrocki, J.: Catalytic ozonation and methods of enhancing molecular ozone reactions in water treatment. *Applied Catalysis B: Environmental* 46(4) (2003) 639-669, doi:[https://doi.org/10.1016/S0926-3373\(03\)00326-6](https://doi.org/10.1016/S0926-3373(03)00326-6).
122. Fu, Q., Wang, D., Li, X., Yang, Q., Xu, Q., Ni, B.-J., Wang, Q., Liu, X.: Towards hydrogen production from waste activated sludge: Principles, challenges and perspectives. *Renewable and Sustainable Energy Reviews* 135 (2021) 110283, doi:<https://doi.org/10.1016/j.rser.2020.110283>.
123. Borole, A.P., Hamilton, C.Y., Vishnivetskaya, T., Leak, D., Andras, C.: Improving power production in acetate-fed microbial fuel cells via enrichment of exoelectrogenic organisms in flow-through systems. *Biochemical Engineering Journal* 48(1) (2009) 71-80, doi:<https://doi.org/10.1016/j.bej.2009.08.008>.
124. Borole, A.P., Hamilton, C.Y., Vishnivetskaya, T.A., Leak, D., Andras, C., Morrell-Falvey, J., Keller, M., Davison, B.: Integrating engineering design improvements with exoelectrogen enrichment process to increase power output from microbial fuel cells. *Journal of Power Sources* 191(2) (2009) 520-527, doi:<https://doi.org/10.1016/j.jpowsour.2009.02.006>.
125. Borole, A.P.: Improving energy efficiency and enabling water recycling in biorefineries using bioelectrochemical systems†. *Biofuels, Bioproducts and Biorefining* 5(1) (2011) 28-36, doi:10.1002/bbb.265.
126. Rozendal, R.A., Hamelers, H.V.M., Molenkamp, R.J., Buisman, C.J.N.: Performance of single chamber biocatalyzed electrolysis with different types of ion exchange membranes. *Water Research* 41(9) (2007) 1984-1994, doi:<https://doi.org/10.1016/j.watres.2007.01.019>.
127. Cheng, S., Logan, B.E.: High hydrogen production rate of microbial electrolysis cell (MEC) with reduced electrode spacing. *Bioresource Technology* 102(3) (2011) 3571-3574, doi:<https://doi.org/10.1016/j.biortech.2010.10.025>.

128. Venkata Mohan, S., Lenin Babu, M., Venkateswar Reddy, M., Mohanakrishna, G., Sarma, P.N.: Harnessing of biohydrogen by acidogenic fermentation of Citrus limetta peelings: Effect of extraction procedure and pretreatment of biocatalyst. *International Journal of Hydrogen Energy* 34(15) (2009) 6149-6156, doi:<https://doi.org/10.1016/j.ijhydene.2009.05.056>.
129. Kumar, G., Sivagurunathan, P., Pugazhendhi, A., Thi, N.B.D., Zhen, G., Chandrasekhar, K., Kadier, A.: A comprehensive overview on light independent fermentative hydrogen production from wastewater feedstock and possible integrative options. *Energy Conversion and Management* 141 (2017) 390-402, doi:<https://doi.org/10.1016/j.enconman.2016.09.087>.
130. Pasupuleti, S.B., Srikanth, S., Venkata Mohan, S., Pant, D.: Development of exoelectrogenic bioanode and study on feasibility of hydrogen production using abiotic VITO-CoRE™ and VITO-CASE™ electrodes in a single chamber microbial electrolysis cell (MEC) at low current densities. *Bioresource Technology* 195 (2015) 131-138, doi:<https://doi.org/10.1016/j.biortech.2015.06.145>.
131. Kim, B.H., Chang, I.S., Cheol Gil, G., Park, H.S., Kim, H.J.: Novel BOD (biological oxygen demand) sensor using mediator-less microbial fuel cell. *Biotechnology Letters* 25(7) (2003) 541-545, doi:10.1023/A:1022891231369.
132. Rozendal, R.A., Hamelers, H.V.M., Euverink, G.J.W., Metz, S.J., Buisman, C.J.N.: Principle and perspectives of hydrogen production through biocatalyzed electrolysis. *International Journal of Hydrogen Energy* 31(12) (2006) 1632-1640, doi:<https://doi.org/10.1016/j.ijhydene.2005.12.006>.
133. Rajalakshmi, N., Balaji, R., Ramakrishnan, S.: 14 - Recent developments in hydrogen fuel cells: Strengths and weaknesses. In: Dutta, S., Mustansar Hussain, C. (eds.) *Sustainable Fuel Technologies Handbook*. pp. 431-456. Academic Press, (2021)
134. Laser, M., Jin, H., Jayawardhana, K., Dale, B.E., Lynd, L.R.: Projected mature technology scenarios for conversion of cellulosic biomass to ethanol with coproduction thermochemical fuels, power, and/or animal feed protein. *Biofuels, Bioproducts and Biorefining* 3(2) (2009) 231-246, doi:10.1002/bbb.131.
135. Piroonlerkgul, P., Wiyaratn, W., Soottitantawat, A., Kiatkittipong, W., Arpornwichanop, A., Laosiripojana, N., Assabumrungrat, S.: Operation viability and performance of solid oxide fuel cell fuelled by different feeds. *Chemical Engineering Journal* 155(1) (2009) 411-418, doi:<https://doi.org/10.1016/j.cej.2009.08.001>.
136. Kusserow, B., Schimpf, S., Claus, P.: Hydrogenation of Glucose to Sorbitol over Nickel and Ruthenium Catalysts. *Advanced Synthesis & Catalysis* 345(1 - 2) (2003) 289-299, doi:10.1002/adsc.200390024.
137. Hoffer, B.W., Crezee, E., Devred, F., Mooijman, P.R.M., Sloof, W.G., Kooyman, P.J., van Langeveld, A.D., Kapteijn, F., Moulijn, J.A.: The role of the active phase of Raney-type Ni catalysts in the selective hydrogenation of d-glucose to d-sorbitol. *Applied Catalysis A: General* 253(2) (2003) 437-452, doi:[https://doi.org/10.1016/S0926-860X\(03\)00553-2](https://doi.org/10.1016/S0926-860X(03)00553-2).
138. Revuelta, J.L., Buey, R.M., Ledesma-Amaro, R., Vandamme, E.J.: Microbial biotechnology for the synthesis of (pro)vitamins, biopigments and antioxidants: challenges and opportunities. *Microbial Biotechnology* 9(5) (2016) 564-567, doi:10.1111/1751-7915.12379.
139. McKinlay, J.B., Vieille, C.: ¹³C-metabolic flux analysis of *Actinobacillus succinogenes* fermentative metabolism at different NaHCO₃ and H₂ concentrations. *Metabolic Engineering* 10(1) (2008) 55-68, doi:<https://doi.org/10.1016/j.ymben.2007.08.004>.

140. Nevin, K.P., Woodard, T.L., Franks, A.E., Summers, Z.M., Lovley, D.R.: Microbial Electrosynthesis: Feeding Microbes Electricity To Convert Carbon Dioxide and Water to Multicarbon Extracellular Organic Compounds. *mBio* 1(2) (2010) e00103-00110, doi:10.1128/mBio.00103-10.
141. Nevin, K.P., Hensley, S.A., Franks, A.E., Summers, Z.M., Ou, J., Woodard, T.L., Snoeyenbos-West, O.L., Lovley, D.R.: Electrosynthesis of Organic Compounds from Carbon Dioxide Is Catalyzed by a Diversity of Acetogenic Microorganisms. *Applied and Environmental Microbiology* 77(9) (2011) 2882, doi:10.1128/AEM.02642-10.
142. Batlle-Vilanova, P., Puig, S., Gonzalez-Olmos, R., Balaguer, M.D., Colprim, J.: Continuous acetate production through microbial electrosynthesis from CO₂ with microbial mixed culture. *Journal of Chemical Technology & Biotechnology* 91(4) (2016) 921-927, doi:10.1002/jctb.4657.
143. Desloover, J., Arends, Jan B.A., Hennebel, T., Rabaey, K.: Operational and technical considerations for microbial electrosynthesis. *Biochemical Society Transactions* 40(6) (2012) 1233-1238, doi:10.1042/BST20120111.
144. Rabaey, K., Rozendal, R.A.: Microbial electrosynthesis — revisiting the electrical route for microbial production. *Nature Reviews Microbiology* 8(10) (2010) 706-716, doi:10.1038/nrmicro2422.
145. Chae, K.-J., Choi, M.-J., Kim, K.-Y., Ajayi, F.F., Chang, I.-S., Kim, I.S.: Selective inhibition of methanogens for the improvement of biohydrogen production in microbial electrolysis cells. *International Journal of Hydrogen Energy* 35(24) (2010) 13379-13386, doi:<https://doi.org/10.1016/j.ijhydene.2009.11.114>.
146. Guo, Y., Chen, Z., Zhang, S.: Methane-fueled microbial fuel cells with the formate-acclimating electroactive culture. *Science of The Total Environment* 754 (2021) 142186, doi:<https://doi.org/10.1016/j.scitotenv.2020.142186>.
147. Wang, A., Liu, W., Cheng, S., Xing, D., Zhou, J., Logan, B.E.: Source of methane and methods to control its formation in single chamber microbial electrolysis cells. *International Journal of Hydrogen Energy* 34(9) (2009) 3653-3658, doi:<https://doi.org/10.1016/j.ijhydene.2009.03.005>.
148. Clauwaert, P., Verstraete, W.: Methanogenesis in membraneless microbial electrolysis cells. *Applied Microbiology and Biotechnology* 82(5) (2009) 829-836, doi:10.1007/s00253-008-1796-4.
149. Sasaki, K., Hirano, S.-i., Morita, M., Sasaki, D., Matsumoto, N., Ohmura, N., Igarashi, Y.: Bioelectrochemical system accelerates microbial growth and degradation of filter paper. *Applied Microbiology and Biotechnology* 89(2) (2011) 449-455, doi:10.1007/s00253-010-2972-x.
150. Villano, M., Aulenta, F., Ciucci, C., Ferri, T., Giuliano, A., Majone, M.: Bioelectrochemical reduction of CO₂ to CH₄ via direct and indirect extracellular electron transfer by a hydrogenophilic methanogenic culture. *Bioresource Technology* 101(9) (2010) 3085-3090, doi:<https://doi.org/10.1016/j.biortech.2009.12.077>.
151. Rader, G.K., Logan, B.E.: Multi-electrode continuous flow microbial electrolysis cell for biogas production from acetate. *International Journal of Hydrogen Energy* 35(17) (2010) 8848-8854, doi:<https://doi.org/10.1016/j.ijhydene.2010.06.033>.
152. Malyan, S.K., Kumar, S.S., Singh, L., Singh, R., Ashok Jadhav, D., Kumar, V.: Chapter 9 - Bioelectrochemical systems for removal and recovery of heavy metals. In: Singh, L., Mahapatra, D.M., Thakur, S. (eds.) *Bioremediation, Nutrients, and Other Valuable Product Recovery*. pp. 185-203. Elsevier, (2021)

153. Ho, N.A.D., Babel, S.: Chapter 8 - Functional components of a bioelectrochemical system for removal and recovery of metals: A review. In: Singh, L., Mahapatra, D.M., Thakur, S. (eds.) *Bioremediation, Nutrients, and Other Valuable Product Recovery*. pp. 153-183. Elsevier, (2021)
154. Heijne, A.T., Liu, F., Weijden, R.v.d., Weijma, J., Buisman, C.J.N., Hamelers, H.V.M.: Copper Recovery Combined with Electricity Production in a Microbial Fuel Cell. *Environmental Science & Technology* 44(11) (2010) 4376-4381, doi:[10.1021/es100526g](https://doi.org/10.1021/es100526g).
155. Wang, Z., Lim, B., Choi, C.: Removal of Hg²⁺ as an electron acceptor coupled with power generation using a microbial fuel cell. *Bioresource Technology* 102(10) (2011) 6304-6307, doi:<https://doi.org/10.1016/j.biortech.2011.02.027>.
156. Modin, O., Wang, X., Wu, X., Rauch, S., Fedje, K.K.: Bioelectrochemical recovery of Cu, Pb, Cd, and Zn from dilute solutions. *Journal of Hazardous Materials* 235-236 (2012) 291-297, doi:<https://doi.org/10.1016/j.jhazmat.2012.07.058>.
157. Zhang, F., Li, J., He, Z.: A new method for nutrients removal and recovery from wastewater using a bioelectrochemical system. *Bioresource Technology* 166 (2014) 630-634, doi:<https://doi.org/10.1016/j.biortech.2014.05.105>.
158. Jadhav, D.A., Ghangrekar, M.M.: Effective ammonium removal by anaerobic oxidation in microbial fuel cells. *Environmental Technology* 36(6) (2015) 767-775, doi:10.1080/09593330.2014.960481.
159. Chen, X., Liang, P., Zhang, X., Huang, X.: Bioelectrochemical systems-driven directional ion transport enables low-energy water desalination, pollutant removal, and resource recovery. *Bioresource Technology* 215 (2016) 274-284, doi:<https://doi.org/10.1016/j.biortech.2016.02.107>.
160. Kuntke, P., Śmiech, K.M., Bruning, H., Zeeman, G., Saakes, M., Sleutels, T.H.J.A., Hamelers, H.V.M., Buisman, C.J.N.: Ammonium recovery and energy production from urine by a microbial fuel cell. *Water Research* 46(8) (2012) 2627-2636, doi:<https://doi.org/10.1016/j.watres.2012.02.025>.
161. Ieropoulos, I., Greenman, J., Melhuish, C.: Urine utilisation by microbial fuel cells; energy fuel for the future. *Physical Chemistry Chemical Physics* 14(1) (2012) 94-98, doi:10.1039/C1CP23213D.
162. Haddadi, S., Elbeshbishy, E., Lee, H.-S.: Implication of diffusion and significance of anodic pH in nitrogen-recovering microbial electrochemical cells. *Bioresource Technology* 142 (2013) 562-569, doi:<https://doi.org/10.1016/j.biortech.2013.05.075>.
163. Blázquez, E., Gabriel, D., Baeza, J.A., Guisasola, A.: Treatment of high-strength sulfate wastewater using an autotrophic biocathode in view of elemental sulfur recovery. *Water Research* 105 (2016) 395-405, doi:<https://doi.org/10.1016/j.watres.2016.09.014>.
164. Coma, M., Puig, S., Pous, N., Balaguer, M.D., Colprim, J.: Biocatalysed sulphate removal in a BES cathode. *Bioresource Technology* 130 (2013) 218-223, doi:<https://doi.org/10.1016/j.biortech.2012.12.050>.
165. Premier, G.C., Kim, J.R., Massanet-Nicolau, J., Kyazze, G., Esteves, S.R.R., Penumathsa, B.K.V., Rodríguez, J., Maddy, J., Dinsdale, R.M., Guwy, A.J.: Integration of biohydrogen, biomethane and bioelectrochemical systems. *Renewable Energy* 49 (2013) 188-192, doi:<https://doi.org/10.1016/j.renene.2012.01.035>.
166. Donovan, C., Dewan, A., Peng, H., Heo, D., Beyenal, H.: Power management system for a 2.5W remote sensor powered by a sediment microbial fuel cell. *Journal of Power Sources* 196(3) (2011) 1171-1177, doi:<https://doi.org/10.1016/j.jpowsour.2010.08.099>.

167. Williams, K.H., Nevin, K.P., Franks, A., Englert, A., Long, P.E., Lovley, D.R.: Electrode-Based Approach for Monitoring In Situ Microbial Activity During Subsurface Bioremediation. *Environmental Science & Technology* 44(1) (2010) 47-54, doi:[10.1021/es9017464](https://doi.org/10.1021/es9017464).
168. Mohanakrishna, G., Venkata Mohan, S., Sarma, P.N.: Bio-electrochemical treatment of distillery wastewater in microbial fuel cell facilitating decolorization and desalination along with power generation. *Journal of Hazardous Materials* 177(1) (2010) 487-494, doi:<https://doi.org/10.1016/j.jhazmat.2009.12.059>.
169. Mohan, S.V., Raghavulu, S.V., Peri, D., Sarma, P.N.: Integrated function of microbial fuel cell (MFC) as bio-electrochemical treatment system associated with bioelectricity generation under higher substrate load. *Biosensors and Bioelectronics* 24(7) (2009) 2021-2027, doi:<https://doi.org/10.1016/j.bios.2008.10.011>.
170. Güven, G., Perendeci, A., Tanyolaç, A.: Electrochemical treatment of deproteinated whey wastewater and optimization of treatment conditions with response surface methodology. *Journal of Hazardous Materials* 157(1) (2008) 69-78, doi:<https://doi.org/10.1016/j.jhazmat.2007.12.082>.
171. Israilides, C.J., Vlyssides, A.G., Mourafeti, V.N., Karvouni, G.: Olive oil wastewater treatment with the use of an electrolysis system. *Bioresource Technology* 61(2) (1997) 163-170, doi:[https://doi.org/10.1016/S0960-8524\(97\)00023-0](https://doi.org/10.1016/S0960-8524(97)00023-0).
172. Wilberforce, T., Sayed, E.T., Abdelkareem, M.A., Elsaid, K., Olabi, A.G.: Value added products from wastewater using bioelectrochemical systems: Current trends and perspectives. *Journal of Water Process Engineering* 39 (2021) 101737, doi:<https://doi.org/10.1016/j.jwpe.2020.101737>.
173. Kaur, R., Marwaha, A., Chhabra, V.A., Kim, K.-H., Tripathi, S.K.: Recent developments on functional nanomaterial-based electrodes for microbial fuel cells. *Renewable and Sustainable Energy Reviews* 119 (2020) 109551, doi:<https://doi.org/10.1016/j.rser.2019.109551>.
174. Wei, L., Han, H., Shen, J.: Effects of cathodic electron acceptors and potassium ferricyanide concentrations on the performance of microbial fuel cell. *International Journal of Hydrogen Energy* 37(17) (2012) 12980-12986, doi:<https://doi.org/10.1016/j.ijhydene.2012.05.068>.
175. You, S., Zhao, Q., Zhang, J., Jiang, J., Zhao, S.: A microbial fuel cell using permanganate as the cathodic electron acceptor. *Journal of Power Sources* 162(2) (2006) 1409-1415, doi:<https://doi.org/10.1016/j.jpowsour.2006.07.063>.
176. Jadhav, D.A., Ghadge, A.N., Mondal, D., Ghangrekar, M.M.: Comparison of oxygen and hypochlorite as cathodic electron acceptor in microbial fuel cells. *Bioresource Technology* 154 (2014) 330-335, doi:<https://doi.org/10.1016/j.biortech.2013.12.069>.
177. Mani, P., V T, F., Bowman, K., T S, C., Keshavarz, T., Kyazze, G.: Development of an electroactive aerobic biocathode for microbial fuel cell applications. *Environmental Microbiology Reports* 12(5) (2020) 607-612, doi:<https://doi.org/10.1111/1758-2229.12871>.
178. Zhao, Y., Dong, Z., Wang, Y., Li, J., An, X., Yang, D.: Process kinetics for the electrocatalytic hydrogen evolution reaction on carbon-based Ni/NiO nanocomposite in a single-chamber microbial electrolysis cell. *International Journal of Hydrogen Energy* 44(54) (2019) 28841-28847, doi:<https://doi.org/10.1016/j.ijhydene.2019.05.018>.
179. Su, M., Wei, L., Qiu, Z., Wang, G., Shen, J.: Hydrogen production in single chamber microbial electrolysis cells with stainless steel fiber felt cathodes. *Journal of Power Sources* 301 (2016) 29-34, doi:<https://doi.org/10.1016/j.jpowsour.2015.09.108>.

180. Wang, L., Trujillo, S., Liu, H.: Selective inhibition of methanogenesis by acetylene in single chamber microbial electrolysis cells. *Bioresource Technology* 274 (2019) 557-560, doi:<https://doi.org/10.1016/j.biortech.2018.12.039>.
181. Sevda, S., Yuan, H., He, Z., Abu-Reesh, I.M.: Microbial desalination cells as a versatile technology: Functions, optimization and prospective. *Desalination* 371 (2015) 9-17, doi:<https://doi.org/10.1016/j.desal.2015.05.021>.
182. Jacobson, K.S., Drew, D.M., He, Z.: Efficient salt removal in a continuously operated upflow microbial desalination cell with an air cathode. *Bioresource Technology* 102(1) (2011) 376-380, doi:<https://doi.org/10.1016/j.biortech.2010.06.030>.
183. Cao, X., Huang, X., Liang, P., Xiao, K., Zhou, Y., Zhang, X., Logan, B.E.: A New Method for Water Desalination Using Microbial Desalination Cells. *Environmental Science & Technology* 43(18) (2009) 7148-7152, doi:10.1021/es901950j.
184. Luo, H., Xu, P., Roane, T.M., Jenkins, P.E., Ren, Z.: Microbial desalination cells for improved performance in wastewater treatment, electricity production, and desalination. *Bioresource Technology* 105 (2012) 60-66, doi:<https://doi.org/10.1016/j.biortech.2011.11.098>.
185. Katuri, K.P., Ali, M., Saikaly, P.E.: The role of microbial electrolysis cell in urban wastewater treatment: integration options, challenges, and prospects. *Current Opinion in Biotechnology* 57 (2019) 101-110, doi:<https://doi.org/10.1016/j.copbio.2019.03.007>.
186. Philipp, L.-A., Edel, M., Gescher, J.: Chapter One - Genetic engineering for enhanced productivity in bioelectrochemical systems. In: Gadd, G.M., Sariaslani, S. (eds.) *Advances in Applied Microbiology*, vol. 111. pp. 1-31. Academic Press, (2020)
187. Noori Md, T., Ganta, A., Tiwari Bikash, R.: Recent Advances in the Design and Architecture of Bioelectrochemical Systems to Treat Wastewater and to Produce Choice-Based Byproducts. *Journal of Hazardous, Toxic, and Radioactive Waste* 24(3) (2020) 04020023, doi:10.1061/(ASCE)HZ.2153-5515.0000510.
188. Zhou, L., Yan, X., Yan, Y., Li, T., An, J., Liao, C., Li, N., Wang, X.: Electrode potential regulates phenol degradation pathways in oxygen-diffused microbial electrochemical system. *Chemical Engineering Journal* 381 (2020) 122663, doi:<https://doi.org/10.1016/j.cej.2019.122663>.
189. Perry, S.C., Pangotra, D., Vieira, L., Csepei, L.-I., Sieber, V., Wang, L., Ponce de León, C., Walsh, F.C.: Electrochemical synthesis of hydrogen peroxide from water and oxygen. *Nature Reviews Chemistry* 3(7) (2019) 442-458, doi:10.1038/s41570-019-0110-6.
190. Zhao, Q., An, J., Wang, X., Li, N.: In-situ hydrogen peroxide synthesis with environmental applications in bioelectrochemical systems: A state-of-the-art review. *International Journal of Hydrogen Energy* 46(4) (2021) 3204-3219, doi:<https://doi.org/10.1016/j.ijhydene.2020.05.227>.
191. Pasternak, G., Askitosari, T.D., Rosenbaum, M.A.: Biosurfactants and Synthetic Surfactants in Bioelectrochemical Systems: A Mini-Review. 11(358) (2020), doi:10.3389/fmicb.2020.00358.
192. Mao, D., Lu, L., Revil, A., Zuo, Y., Hinton, J., Ren, Z.J.: Geophysical Monitoring of Hydrocarbon-Contaminated Soils Remediated with a Bioelectrochemical System. *Environmental Science & Technology* 50(15) (2016) 8205-8213, doi:10.1021/acs.est.6b00535.
193. Lu, L., Yazdi, H., Jin, S., Zuo, Y., Fallgren, P.H., Ren, Z.J.: Enhanced bioremediation of hydrocarbon-contaminated soil using pilot-scale bioelectrochemical systems. *Journal of Hazardous Materials* 274 (2014) 8-15, doi:<https://doi.org/10.1016/j.jhazmat.2014.03.060>.

194. Du, W., Wan, Y., Zhong, N., Fei, J., Zhang, Z., Chen, L., Hao, J.: Status quo of soil petroleum contamination and evolution of bioremediation. *Petroleum Science* 8(4) (2011) 502-514, doi:10.1007/s12182-011-0168-3.
195. Pankratov, D., Pankratova, G., Gorton, L.: Current Trends in Development of Photosynthetic Bioelectrochemical Systems for Light Energy Conversion. In: *Novel Catalyst Materials for Bioelectrochemical Systems: Fundamentals and Applications*, vol. 1342. ACS Symposium Series, vol. 1342, pp. 123-146. American Chemical Society, (2020)
196. McCormick, A.J., Bombelli, P., Bradley, R.W., Thorne, R., Wenzel, T., Howe, C.J.: Biophotovoltaics: oxygenic photosynthetic organisms in the world of bioelectrochemical systems. *Energy & Environmental Science* 8(4) (2015) 1092-1109, doi:10.1039/C4EE03875D.
197. Apollon, W., Luna-Maldonado, A.I., Kamaraj, S.-K., Vidales-Contreras, J.A., Rodríguez-Fuentes, H., Gómez-Leyva, J.F., Aranda-Ruiz, J.: Progress and recent trends in photosynthetic assisted microbial fuel cells: A review. *Biomass and Bioenergy* 148 (2021) 106028, doi:<https://doi.org/10.1016/j.biombioe.2021.106028>.
198. Pinck, S., Ostormujof, L.M., Teychené, S., Erable, B.: Microfluidic Microbial Bioelectrochemical Systems: An Integrated Investigation Platform for a More Fundamental Understanding of Electroactive Bacterial Biofilms. *Microorganisms* 8(11) (2020), doi:10.3390/microorganisms8111841.
199. Noreña-Caro, D., Benton, M.G.: Cyanobacteria as photoautotrophic biofactories of high-value chemicals. *Journal of CO2 Utilization* 28 (2018) 335-366, doi:<https://doi.org/10.1016/j.jcou.2018.10.008>.
200. Liebal, U.W., Blank, L.M., Ebert, B.E.: CO2 to succinic acid – Estimating the potential of biocatalytic routes. *Metabolic Engineering Communications* 7 (2018) e00075, doi:<https://doi.org/10.1016/j.mec.2018.e00075>.
201. Takors, R., Kopf, M., Mampel, J., Bluemke, W., Blombach, B., Eikmanns, B., Bengelsdorf, F.R., Weuster-Botz, D., Dürre, P.: Using gas mixtures of CO, CO2 and H2 as microbial substrates: the do's and don'ts of successful technology transfer from laboratory to production scale. *Microbial Biotechnology* 11(4) (2018) 606-625, doi:10.1111/1751-7915.13270.
202. Gutiérrez-Guerra, N., Valverde, J.L., Romero, A., Serrano-Ruiz, J.C., de Lucas-Consuegra, A.: Electrocatalytic conversion of CO2 to added-value chemicals in a high-temperature proton-exchange membrane reactor. *Electrochemistry Communications* 81 (2017) 128-131, doi:<https://doi.org/10.1016/j.elecom.2017.06.018>.
203. Alper, E., Yuksel Orhan, O.: CO2 utilization: Developments in conversion processes. *Petroleum* 3(1) (2017) 109-126, doi:<https://doi.org/10.1016/j.petlm.2016.11.003>.
204. Maeda, C., Miyazaki, Y., Ema, T.: Recent progress in catalytic conversions of carbon dioxide. *Catalysis Science & Technology* 4(6) (2014) 1482-1497, doi:10.1039/C3CY00993A.
205. Aida, T., Inoue, S.: Activation of carbon dioxide with aluminum porphyrin and reaction with epoxide. Studies on (tetraphenylporphinato)aluminum alkoxide having a long oxyalkylene chain as the alkoxide group. *Journal of the American Chemical Society* 105(5) (1983) 1304-1309, doi:10.1021/ja00343a038.
206. Kruper, W.J., Dellar, D.D.: Catalytic Formation of Cyclic Carbonates from Epoxides and CO2 with Chromium Metalloporphyrinates. *The Journal of Organic Chemistry* 60(3) (1995) 725-727, doi:10.1021/jo00108a042.
207. Paddock, R.L., Hiyama, Y., McKay, J.M., Nguyen, S.T.: Co(III) porphyrin/DMAP: an efficient catalyst system for the synthesis of cyclic carbonates from CO2 and epoxides. *Tetrahedron Letters* 45(9) (2004) 2023-2026, doi:<https://doi.org/10.1016/j.tetlet.2003.10.101>.

208. Jin, L., Jing, H., Chang, T., Bu, X., Wang, L., Liu, Z.: Metal porphyrin/phenyltrimethylammonium tribromide: High efficient catalysts for coupling reaction of CO₂ and epoxides. *Journal of Molecular Catalysis A: Chemical* 261(2) (2007) 262-266, doi:<https://doi.org/10.1016/j.molcata.2006.06.011>.
209. Bai, D., Wang, Q., Song, Y., Li, B., Jing, H.: Synthesis of cyclic carbonate from epoxide and CO₂ catalyzed by magnetic nanoparticle-supported porphyrin. *Catalysis Communications* 12(7) (2011) 684-688, doi:<https://doi.org/10.1016/j.catcom.2010.12.024>.
210. Ahmadi, F., Tangestaninejad, S., Moghadam, M., Mirkhani, V., Mohammadpoor-Baltork, I., Khosropour, A.R.: Highly efficient chemical fixation of carbon dioxide catalyzed by high-valent tetraphenylporphyrinatotin(IV) triflate. *Inorganic Chemistry Communications* 14(9) (2011) 1489-1493, doi:<https://doi.org/10.1016/j.inoche.2011.05.053>.
211. Fujihara, T., Nogi, K., Xu, T., Terao, J., Tsuji, Y.: Nickel-Catalyzed Carboxylation of Aryl and Vinyl Chlorides Employing Carbon Dioxide. *Journal of the American Chemical Society* 134(22) (2012) 9106-9109, doi:10.1021/ja303514b.
212. Gu, L., Zhang, Y.: Unexpected CO₂ Splitting Reactions To Form CO with N-Heterocyclic Carbenes as Organocatalysts and Aromatic Aldehydes as Oxygen Acceptors. *Journal of the American Chemical Society* 132(3) (2010) 914-915, doi:10.1021/ja909038t.
213. Shoener, B.D., Bradley, I.M., Cusick, R.D., Guest, J.S.: Energy positive domestic wastewater treatment: the roles of anaerobic and phototrophic technologies. *Environmental Science: Processes & Impacts* 16(6) (2014) 1204-1222, doi:10.1039/C3EM00711A.
214. Foley, J., de Haas, D., Hartley, K., Lant, P.: Comprehensive life cycle inventories of alternative wastewater treatment systems. *Water Research* 44(5) (2010) 1654-1666, doi:<https://doi.org/10.1016/j.watres.2009.11.031>.
215. Fuchs, V.J., Mihelcic, J.R., Gierke, J.S.: Life cycle assessment of vertical and horizontal flow constructed wetlands for wastewater treatment considering nitrogen and carbon greenhouse gas emissions. *Water Research* 45(5) (2011) 2073-2081, doi:<https://doi.org/10.1016/j.watres.2010.12.021>.

Recent advances in bio-electrochemical system analysis in biorefineries

Siwal, Samarjeet Singh

2021-07-02

Attribution-NonCommercial-NoDerivatives 4.0 International

Siwal SS, Zhang Q, Saini AK, et al., (2021) Recent advances in bio-electrochemical system analysis in biorefineries. *Journal of Environmental Chemical Engineering*, Volume 9, Issue 5, October 2021, Article number 105982

<https://doi.org/10.1016/j.jece.2021.105982>

Downloaded from CERES Research Repository, Cranfield University