## Towards a Method of Exploring the Limits of Low Frequency Time-Temperature Superposition Models in Predicting the High Frequency Response of Epoxy-Amine Resins



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### CENTRE OF EXCELLENCE

#### Introduction

- Material responses over 0.1-2000 Hz are a cause for concern for a variety of industries
- Time-temperature superposition is one method of probing a materials response over a wide range of frequencies and temperatures from low frequency data
- There are three main challenges with using TTS as a method, the problems are outlined below and potential solutions identified

#### **Experimental**

Two epoxies were manufactured and Dynamic Mechanical Analysis (DMA) was performed to obtain both the shear damping properties and the glass transition temperatures of the two materials.

Time-temperature superposition performed done using a 0.05 % strain, a logarithmic frequency sweep over the range 1-100 Hz and a step temperature increase of 5 °C between 0-180 °C.

The master curves were generated using the procedure below.

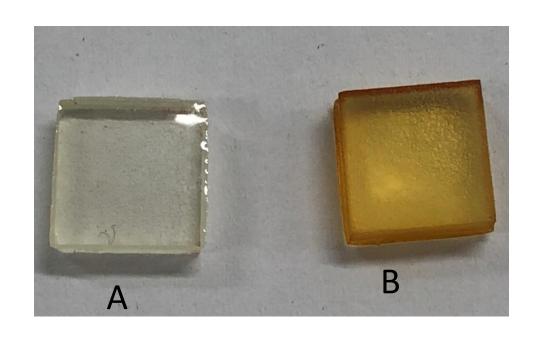


Figure 1: Epoxy resin samples made from bisphenol A diglycidyl ether and A: 3-diethylaminopropylamine. B: 5-Amino-1,3,3-trimethylcyclohexanemethylamine

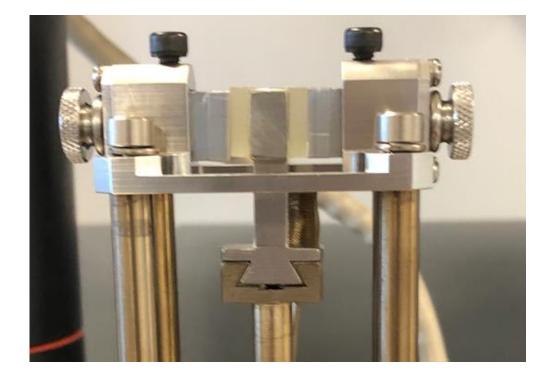


Figure 2: DMA shear clamp with epoxy sample in place.

Step 1: select refence temperature

Step 2: determine initial shifts

Step 3: refine shifts

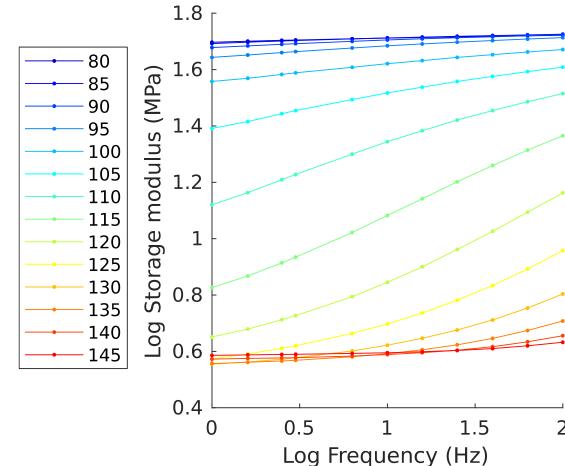
Step 4: calculate WLF and Arrhenius parameters

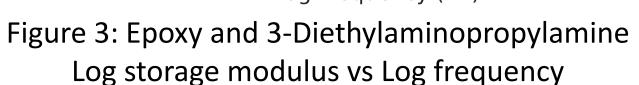
#### Step 1 and Challenge 1

The reference temperature is an arbitrarily selected temperature within the experimental data. It is very much a personal choice and will differ for all materials.

**Potential solution**: Often the reference temperature may be the glass transition for the material.

**Our solution**: was to use a reference temperature approximately within the middle of the isotherm plot (Figures 3 and 4). For the shear data this was found to be different to the glass transition temperatures of the materials.





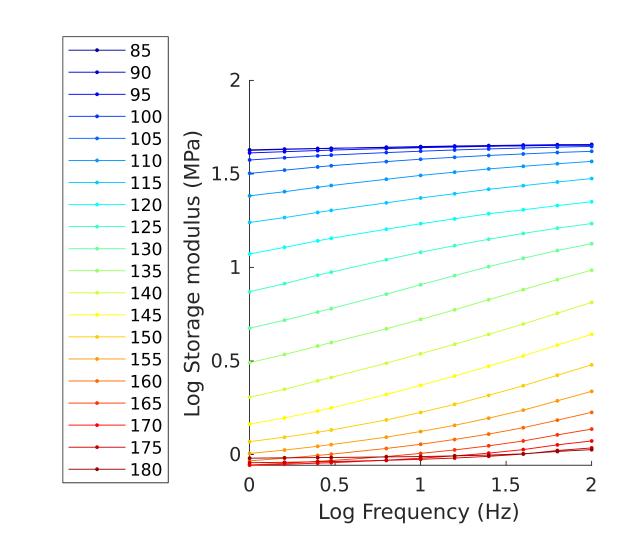


Figure 4: Epoxy and 5-Amino-1,3,3trimethylhexanemethylamine Log storage modulus vs Log frequency

#### Steps 2 and 3 and Challenge 2

After shifting the data around the reference temperature a master curve is created. For the Storage modulus this looks like a typical "s" shape (Figures 5 and 6). The "s" shape is dependent on the analyst. Here a lot of operator bias can be introduced.

Potential solution: Using software will remove some operator bias.

Our solution: To validate the master curves by performing experiments at higher frequencies in line with the ASTM standard E756-05.

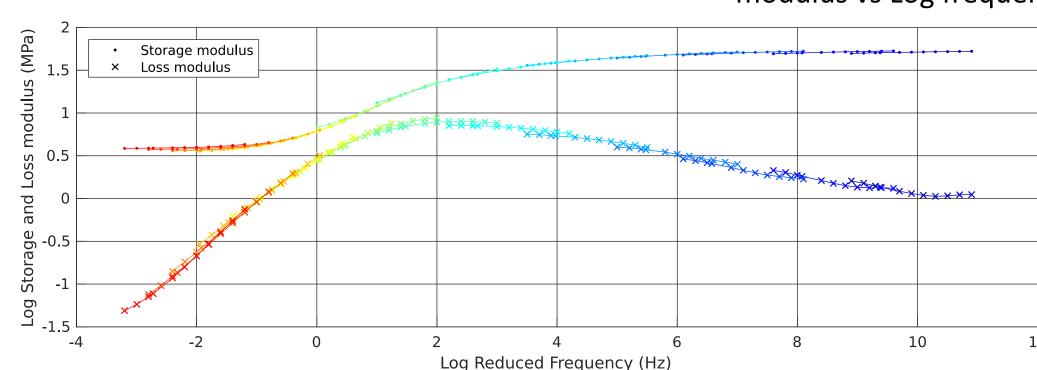


Figure 5: Epoxy and 3-Diethylaminopropylamine master curve for storage modulus

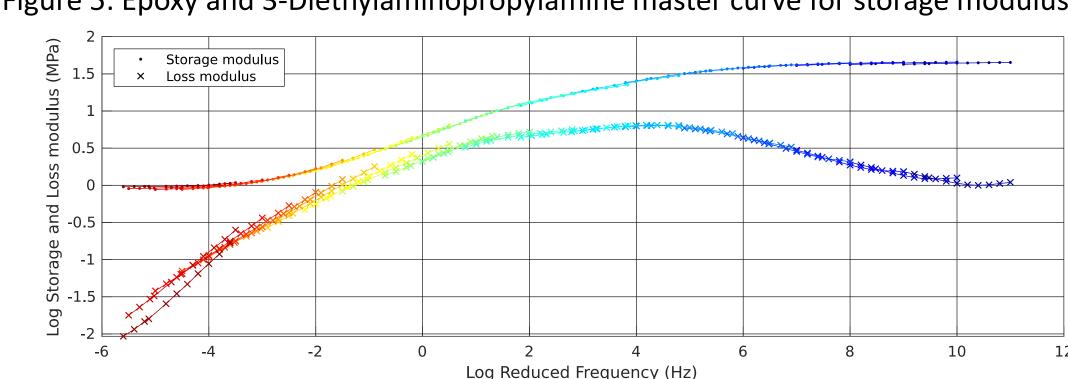


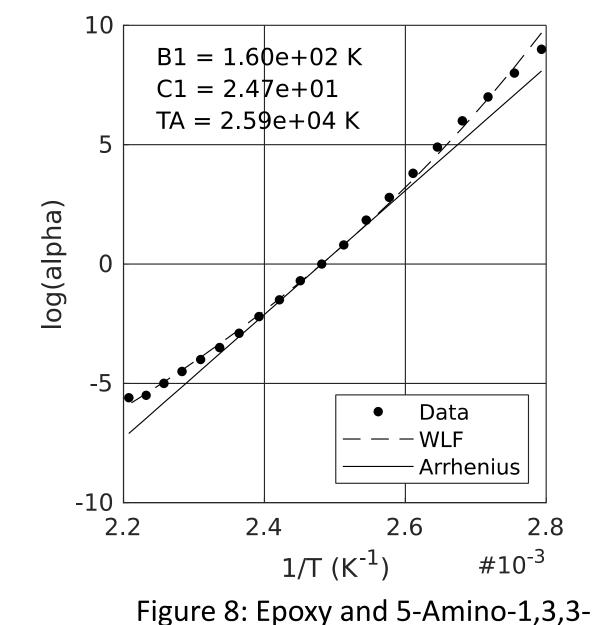
Figure 6: Epoxy and 5-Amino-1,3,3-trimethylhexanemethylamine master curve for storage modulus

# B1 = 7.36e+01 K C1 = 1.13e+01 TA = 2.93e+04 K 5 Data ----WLF -Arrhenius

Figure 7: Epoxy and 3-Diethylaminopropylamine shift factors

 $1/T (K^{-1})$ 

#10<sup>-3</sup>



trimethylhexanemethylamine shift factors

#### Step 4 and Challenge 3

To date there are multiple different shift equations that could be used ranging from the traditional Williams-Landen-Ferry and Arrhenius to newer ones of polynomial and double shift equations.

**Our solution**: Is to use the Williams-Landen-Ferry (WLF) and Arrhenius due to this being the most frequently used throughout literature.

 $\log[\alpha(T)] = T_A \left(\frac{1}{T} - \frac{1}{T_0}\right)$  Arrhenius shift equation

 $\log[lpha(T)] = -C_1 \, rac{(T-T_0)}{B_1 \, (T-T_0)}$  Williams-Landen-Ferry shift equation

Figures 7 and 8 show the trends of the data and how they best fit to the WLF and Arrhenius equations.

#### Conclusions

- Three solutions have been presented to overcome the challenges of TTS
- A good initial starting point for Challenge 1 is to select a reference in the middle of the isotherm plot
- Validating the master curves through high frequency experiments and working close to literature are adopted as approaches to overcoming Challenges 2 and 3



