

Orientation-dependent energy-storage performance and electrocaloric effect in PLZST antiferroelectric thick films

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Abstract

The enhancement of the energy-storage performance and electrocaloric effect (ECE) was achieved via orientation control. The 1.5-mm-(Pb_{0.97}La_{0.02})(Zr_{0.73}Sn_{0.22}Ti_{0.05})O₃ (PLZST) antiferroelectric (AFE) thick films with (111), (110), and (100) crystallographic orientations were successfully prepared via a sol-gel method. It was found that both the enhanced energy-storage density of 13.5 J/cm³ at 900 kV/cm and the corresponding temperature reduction of $\Delta T = 28.1$ °C at room temperature were obtained in the PLZST thick film with (111) orientation due to a high polarization. Moreover, these films with different orientations display a small leakage current density at the room temperature. These results suggest that both energy-storage properties and cooling performance in AFEs could be optimized by the proper orientation control.

1. Introduction

In order to reduce the dependency on fossil based energy reserves, the successful production, distribution, and storage of electricity will form a cornerstone for the development and growth of society and technology in the coming century. The generation of electricity from renewable and non-conventional resources is being investigated and is attracting much attention. High energy-storage density dielectric capacitors as a means for storage of electric energy form an indispensable part of all modern electronic and electrical devices [1–4].

In general, small remnant polarization, the recoverable energy-storage density W and the energy-storage efficiency η are two crucial parameters, which can be calculated according to the equations as below [5]:

$$W = \int_{P_r}^{P_{\max}} E dP, \quad (1)$$

$$W_{\text{loss}} = \int E dP, \quad (2)$$

$$\eta = \frac{W}{W + W_{\text{loss}}}, \quad (3)$$

where E is the applied electric field, P is the polarization, P_r is remanent polarization, P_{\max} is the maximum polarization, and W_{loss} is the energy loss density. Obviously, the high energy-storage density depends on smaller P_r , larger P_{\max} and good electric field endurance. Antiferroelectrics (AFE) play a key role in the dielectric energy-storage capacitors due to the high maximum polarization and large electric field induced-strain. In recent years, PbZrO₃-based AFE materials, such as $\text{Pb}_{0.97}\text{La}_{0.02}(\text{Zr}_{0.97}\text{Ti}_{0.03})\text{O}_3$, $(\text{Pb}_{0.97}\text{La}_{0.02})(\text{Zr}_{0.95-x}\text{Sn}_x\text{Ti}_{0.05})\text{O}_3$, $\text{Pb}_{0.8}\text{Ba}_{0.2}\text{ZrO}_3$, Sr-doped PbZrO_3 , Eu-doped PbZrO_3 , and Nb-doped $\text{PbZr}_{0.4}\text{Ti}_{0.6}\text{O}_3$, were widely investigated in the high energy-storage performance widely investigated in the high energy-storage density capacitors storage performance [6–10].

Like dielectric capacitors, a highly efficient and environmentally friendly solid-state cooling technology also has attracted much attention for the micro-electromechanical systems [11,12]. It offers an alternative to the traditional vapor-compression cooling method accompanied hazardous gases, which is widely used in household and industrial applications. There are several technologies being developed by using of various solid-state cooling technology, such as electrocaloric, magnetocaloric and mechano-caloric effects. Electrocaloric effect (ECE) refers to a change in temperature under adiabatic application of external electric fields [13–15]. Alternatively, it can be described as an isothermal change in entropy. To achieve a large electrocaloric effect, a phase transition is desirable, where the entropy change (ΔS) is maximized. For example, the giant ECE observed in the $0.9\text{PbMg}_{1/3}\text{Nb}_{2/3}\text{O}_3\text{-}0.1\text{PbTiO}_3$ film, $\text{PbZr}_{0.95}\text{Ti}_{0.05}\text{O}_3$ film, and Pb

(Zr_{0.455}Sn_{0.455}Ti_{0.09})O₃ bulk ceramic are associated with ferroelectric (FE)-paraelectric (PE), AFE-PE, and AFE-FE phase transitions, respectively [16–18]. Compared with FE (or AFE)-PE phase transition, the FE-AFE phase transition could be easily realized in a wide temperature range near 25 °C. Moreover, a considerable pyroelectric effect is also formed during FE-AFE transition, which is even larger than that caused by FE (or AFE)-PE phase transition. Therefore, good ECE should also be obtained during FE-AFE phase transition. Quite recently, a giant ECE (DT = 53.8 °C and DS = 63.9 J K⁻¹ kg⁻¹) with an electrocaloric coefficient of 0.060 K cm/kV was also achieved at 5 °C in our work on 2-μm-Pb_{0.97}La_{0.02}(Zr_{0.75}Sn_{0.18}Ti_{0.07})O₃ AFE thick film, which is comparable with the best results in FE materials [19]. These results indicate that ECE originated from FE-AFE phase transition is a promising route for the practical cooling applications.

Energy-storage performance and ECE of the dielectric materials are strongly dependent on their polarization character, which could be tuned by the phase structure, grain size, stress, interface layer, growth orientation, and so on. Orientation is a powerful tool, through which, for instance, enhanced the energy-storage performance, tuned the ferroelectric-antiferroelectric phase transition, and improved the ECE [20–22]. Based on these results, in this work, we report that a typical lead-based AFE thick films of (Pb_{0.97}La_{0.02})(Zr_{0.73}Sn_{0.22}Ti_{0.05})O₃ exhibits an orientation-dependent energy storage performance and electrocaloric effect.

2. Experimental procedure

The composition of (Pb_{0.97}La_{0.02})(Zr_{0.73}Sn_{0.22}Ti_{0.05})O₃ (PLZST) was selected to study, which is located in the morphotropic phase boundary (MPB). The PLZST precursor solution was deposited on Pt (111), LaNiO₃(110)/Pt(111) and LaNiO₃(10 0)/Pt(111) substrates via a sol-gel method, respectively. The detailed process could be found in reference [23]. The only difference is that the PLZST films deposited on Pt(111) was annealed at 700 °C for 25 min and the film deposited on LaNiO₃(110)/Pt(111) and LaNiO₃(10 0)/Pt(111) were annealed at 700 °C for 30 min in order to reduce discrepancy of grain size. The final thickness of all the PLZST AFE thick film was about 1.5 μm.

Here, LaNiO₃ thin films were prepared on the Pt(111) by sol-gel method similar to Ref. [24]. Lanthanum nitrate [La(NO₃)₃] and nickel acetate [Ni(CH₃COO)₂·4H₂O] were used as the start

materials, and acetic acid and water were used as the solvents. In order to avoid cracking during heating, formamide was also added to the system. The concentration of the precursor solution was adjusted to 0.2 M. The LaNiO₃ precursor solution was spin coated on the Pt(111) substrates at 3000 rpm for 20 s. For LaNiO₃(110) film, each wet film was heat treated at 450 °C for 10 min. This procedure was repeated several times. The process was repeated several times. Finally, the LaNiO₃ films were annealed at 700 °C for 60 min to obtain a well-crystallized structure. The whole thermal treatment was completed in a tube furnace [24]. For LaNiO₃(100) film, each wet film was fired at 160 °C for 300 s, then pyrolyzed at 400 °C for 360 s and finally annealed at higher temperatures for 240 s. The process was repeated several times. Finally, the LaNiO₃ films were annealed at 700 °C for 45 min to obtain desired thickness LaNiO₃(100) films. The whole thermal treatment was completed in an RTP furnace [25]. The thickness of both LaNiO₃(110) film and LaNiO₃(100) film are about 400 nm. In order to obtain the same grain size of both LaNiO₃ (110) and LaNiO₃ (100) films, the annealing time of LaNiO₃ (100) film was shortened by 15 min than that in our previous work [23]. For convenience, the PLZST AFE thick films deposited on Pt(111), LaNiO₃(110)/Pt(111) and LaNiO₃(10 0)/Pt(111) substrates are abbreviated as PLZST(111), PLZST(110), and PLZST(100) films, respectively.

The microstructure of the PLZST AFE thick films was analyzed by X-ray diffraction (XRD Bruker D8 Advanced Diffractometer, German) and field-emission scanning electron microscopy (FE-SEM ZEISS Supra 55, German), respectively. For the electrical measurements, gold pads of 0.20 mm in diameter were coated on the films surface as top electrodes by using a DC sputtering method. A computer-controlled Agilent E4980A LCR analyzer was employed for measuring the frequency, and temperature-dependent dielectric properties with the AC drive amplitude of 50 mV. The polarization-electric field hysteresis loops (P–E) at 1 kHz and the leakage current characteristic of the films were measured by a Ferroelectric tester (Radiant Technologies, Inc., Albuquerque, NM).

3. Results and discussion

Fig. 1 shows the XRD patterns of PLZST AFE thick films grown on Pt(111), LaNiO₃(110)/Pt(111) and LaNiO₃(10 0)/Pt(111) substrates after annealing at 700 °C. For convenience, the lattice indexes of the

diffraction peaks are labeled as pseudocubic structure. All the thick films have crystallized into a pure perovskite phase without any other phases. Apparently, the PLZST film on the Pt(111) substrate shows well (111)-preferred orientation, while PLZST films on the LaNiO₃(110)/Pt(111) and LaNiO₃(10 0)/Pt(111) substrates display (110) and (100)-preferred orientations, respectively. In order to estimate the orientation degree of the PLZST films on three substrates, orientation factor $\alpha(111)$, $\alpha(110)$, and $\alpha(10 0)$ are given according to the following well known formulas [26]:

$$\alpha_{(111)} = \frac{I_{(111)}}{I_{(111)} + I_{(110)} + I_{(100)}}, \quad (4)$$

$$\alpha_{(110)} = \frac{I_{(110)}}{I_{(111)} + I_{(110)} + I_{(100)}}, \quad (5)$$

$$\alpha_{(100)} = \frac{I_{(100)}}{I_{(111)} + I_{(110)} + I_{(100)}}. \quad (6)$$

$$\alpha_{(111)} + \alpha_{(110)} + \alpha_{(100)} = 1$$

The calculated values of $\alpha(111)$, $\alpha(110)$, and $\alpha(100)$ are 0.95, 0.86, and 0.90 for PLZST(111), PLZST(110), and PLZST(100) films, respectively. For the PLZST(111) thick film, a transient intermetallic phase Pt₃Pb with preferred-(111) orientation formed at the interface between the PLZST film and Pt(111) substrates during the heat treatment, which function as a seed layer and decrease the activation energy for crystallization of the film [27]. For both PLZST(110) and PLZST (100) films, the introduced LaNiO₃ layer prevented the formation of intermetallic phase Pt₃Pb between the PLZST film and Pt(111) substrates [28]. Meanwhile, the (110) and (100)-preferred LaNiO₃ buffer layer facilitate the growth of PLZST films along the (110) and (100) orientations, respectively, which is due to the same pseudocubic perovskite structure and the small lattice mismatch between AFE thick films and LaNiO₃ bottom electrodes. The similar works were also reported in other lead-based FE and AFE films deposited on LaNiO₃ bottom electrodes [29,30].

The FE-SEM images of surface and cross-sectional morphologies of all the films are given in Fig. 2(a–f). Evidently, these thick films show a dense, uniform and compact structure, and no micro-cracks or other micro-structural defects are found in the films, which are attributed to the two-step heat-treatment procedure for the

thick-film preparation. The average grain size is about 473, 467 and 468 nm for AFE thick films deposited on Pt(111), LaNiO₃(110)/ Pt(111) and LaNiO₃(100)/Pt(111) substrates, respectively, which are calculated by using the Nano Measurer software. The cross-sectional morphologies of these thick films present a columnar-like structure, indicating that the nucleation and growth of the crystals may initiate from the bottom electrodes [31]. The thickness of these films is the same and about 1.5 μm .

Fig. 3 displays the frequency dependence of dielectric constant and dielectric loss of the PLZST(111), PLZST(110) and PLZST(100) AFE thick films measured from 1 kHz to 1 MHz at room temperature. In the measurement frequency range, dielectric constant of three samples are decreased slightly because polarization process of some framework, such as space charges, needs longer time and thus contributes less to the overall polarization under higher frequencies [32]. All the samples share the low loss tangent of below 0.013, which is attributed to their uniform microstructure. It can be seen that the orientation of the films have a clear effect on their dielectric constant. The dielectric constant at 100 kHz is 709, 620, and 512 for the PLZST(111), PLZST (110) and PLZST(100) AFE thick films, respectively. This result is consistent to the dielectric constant at room temperature in PbZrO₃ thin films [33].

Fig. 4 presents the temperature-dependent dielectric constant and dielectric loss of the PLZST(111), PLZST(110) and PLZST(100) AFE thick films measured at 100 kHz and on heating rate of 3 $^{\circ}\text{C}/\text{min}$. Obviously, the dielectric constant initially increase and then decrease with the further increase of temperature. The dielectric peaks corresponding to the transition from AFE to PE phase is observed at 184 $^{\circ}\text{C}$, 195, and 190 $^{\circ}\text{C}$ for the PLZST(111), PLZST(110) and PLZST(100) films, respectively, which correspond to the so-called Curie temperature (T_c). The difference of the Curie temperature of the films should be caused by the difference in the substrate constraint and annealing procedure [34].

Room temperature P-E loops of all the samples are shown in Fig. 5 (a–c), which were measured at 1 kHz. Clearly, all the thick films exhibit double loops, demonstrating their AFE character. A small remnant polarization below 5.0 pLC/cm^2 is detected in these samples, which should be caused by the interface layer, space charge, residual ferroelectric state, unstable AFE regions, and so on. Under the measurement condition, the corresponding maximum polarization for the PLZST(111), PLZST(110) and PLZST(100) films is 77.3, 66.2 and 59.1 pLC/cm^2 , respectively. The large P_{max} implies that a high recyclable energy-

storage density and a large ECE could possibly be obtained in these films. The schematic diagram of the primitive cells in AFE and FE phases with different orientations is shown in Fig. 5 (d–f). The polar direction of the antiparallel dipoles is along the [110] direction of the tetragonal primitive cell. When AFE phase changed into the FE phase under the applied electric field, the tetragonal primitive cell becomes rhombohedral with the polar directions along the [111] direction. Therefore, the PLZST(111) film possesses highest P_{\max} , followed by the PLZST(110), and then the PLZST(100). This result is consistent to that in PbZrO₃ antiferroelectric films, which can be explained by that the oriented film with the higher P_{\max} is nearer to polar vector of FE phase hence leading to lower transition field [21,29,35].

The energy-storage performance of the PLZST thick films with different crystallographic orientations are presented in Fig. 6(a), which were measured from 200 kV/cm to 900 kV/cm at 1 kHz and at room temperature. As desired, for all the films, W values increase nearly linearly with increasing electric field. In the measurement range, the discoverable energy-storage density of the films is improved by controlling the orientations, which attribute to the high P_{\max} . For example, the W value is 13.5, 12.9, and 11.6 J/cm³ for the PLZST(111), PLZST(110) and PLZST(100) AFE thick films at 900 kV/cm, respectively. In practical applications, larger energy-storage efficiency η is also desired. With the increase of the measured electric field, the η values are slightly declined and similar for these thick films. For example, the η value of PLZST(111) thick film is changed from 73% to 63% as the electric field increases from 200 to 900 kV/cm. To investigate the temperature-dependent stability of the capacitors, the energy-storage density and the energy-storage efficiency measured at 1 kHz in the temperature ranging from 20 to 150 °C are presented in Fig. 6(b). In order to avoid electric breakdown at higher temperature, the measurements of P-E loops for all the samples are carried out at a lower electric field of $E = 900$ kV/cm. Evidently, all the curves only slightly vary in the measurement range, indicating good temperature stability of the energy-storage performance.

Assuming the Maxwell relation $\left(\frac{\partial P}{\partial T}\right)_E = \left(\frac{\partial S}{\partial E}\right)_T$, the reversible changes of ΔT and ΔS in a material at applied electric-field E can be expressed as [36]:

$$\Delta S = -\frac{1}{\rho} \int_{E_1}^{E_2} \left(\frac{\partial P}{\partial T} \right)_E dE, \quad (7)$$

$$\Delta T = -\frac{1}{C\rho} \int_{E_1}^{E_2} T \left(\frac{\partial P}{\partial T} \right)_E dE, \quad (8)$$

where ρ is density, C is themolar heat capacity, T is operating temperature, P is maximum polarisation at applied electric field E , E_1 and E_2 are the initial and final applied electric field, and $\left(\frac{\partial P}{\partial T} \right)_E$ is the pyroelectric effect at selected electric field. According to Eq. (8), the reversible adiabatic temperature changes ΔT of the PLZST(111), PLZST(110) and PLZST(100) AFE thick films at 900 kV/cm are plotted in Fig. 7(a). Here $E_1 = 0$ and $E_2 = E$; thus ΔE is equal to E . The specific heat capacity $C = 330 \text{ J K}^{-1} \text{ kg}^{-1}$ and the theoretical density $\rho = 8.3 \text{ g cm}^{-3}$ are selected for these thick films as reported before [37]. Obviously, three samples display the same changing tendency in the operating temperature-dependent ΔT curves. As expected, the large ΔT of these films are received in a wide temperature range of room temperature, which is caused electric field-induced phase transition of FE-AFE. For example, the values of ΔT at 21 °C is 28.1, 23.1, and 21.2 °C for the PLZST(111), PLZST(110) and PLZST (100) films, respectively. With the operating temperature increasing, the ΔT firstly decrease, indicating a reduction of entropy change between AFE and FE phase transition. With the further increase of temperature, a peak of $\Delta T = 13.9, 11.8, \text{ and } 10.7 \text{ °C}$ is detected at 150, 165, and 169 °C of the PLZST(111), PLZST and PLZST(100) films at $E = 900 \text{ kV/cm}$, respectively, which is believed to be induced by the AFE-PE phase transition. The PLZST film displays the largest ΔT values, the PLZST(100) film has the lowest values, which coincides with the result in PbTiO₃/SrTiO₃ multilayers with different orientations [20]. The corresponding adiabatic ΔS of the PLZST(111), PLZST(110) and PLZST(100) films are given in Fig. 7(b), which were measured at 900 kV/cm. Obviously, the changing tendency of ΔS is in agreement with ΔT , as a function of temperature. The maximum values of ΔS are 31.6, 25.9, and 23.8 J K⁻¹ kg⁻¹ of the PLZST(111), PLZST(110) and PLZST (100) films at room temperature at 900 kV/cm, respectively. For the cooling applications, apart from large ΔT and ΔS , higher coefficient of performance COP ($COP = \Delta S \times T_j = \Delta W_j$) and electrocaloric coefficient (the ratio of $\Delta T/\Delta E$) are also desired [38,17]. The maximum COP and $\Delta T/\Delta E$ of 5.7, 5.0, 4.9 and 0.031,

0.026, 0.024 is achieved at room temperature and at 900 kV/cm in the PLZST(111), PLZST(110) and PLZST(100) films, respectively, as shown in Table 1.

Fig. 8(a) illustrates the current-time characteristics of the PLZST (111), PLZST(110) and PLZST(100) AFE thick films, as a function of external electric field, which were measured at room temperature and 600 kV/cm. The decay in dielectric relaxation current density obeys the Curie-von Schweidler law as follows [39]:

$$J = J_s + J_0 \times t^{-n}; \quad (9)$$

where J_s is the steady-state current density, J_0 a fitting constant, t the relaxation time in second, and n the slope of the log–log plot. It is believed that three possible mechanisms are associated with the Curie-von Schweidler law: space charge trapping, relaxation time distribution and electrical charge hopping [40]. By fitting the leakage current density data into Eq. (9), the steady-state leakage current density J_s is 2.14×10^{-6} , 4.37×10^{-6} , and 5.41×10^{-6} A/cm² for the PLZST(111), PLZST(110) and PLZST(100) films, respectively, as shown in Fig. 8(b). The smaller leakage current density is in agreement with our previous reports on the lead-based AFE films [4], which yield negligible Joule heating ($\sim 10^{-3}$ K) and do not affect P–E results because currents of hundreds of mA are required to switch the measured polarizations at 1 kHz. As a typical example, the current density-time characteristics of the PLZST(111) thick films are presented in Fig. 8(c), which was measured at 25, 75, 150, and 200 °C and at 600 kV/cm. The obtained steady-state leakage current density is 1.36×10^{-6} , 3.47×10^{-6} , 1.09×10^{-5} and 2.29×10^{-5} A/cm² at 25, 75, 150, and 200 °C, respectively. It could be obtained from these results that the leakage current density of the samples is increased as temperature increasing. The data could be fitted into the Arrhenius relationship [41]:

$$J_s = C \exp\left(-\frac{E_a}{kT}\right) \quad (10)$$

where C is a fitting constant, E_a the activation energy, and k the Boltzmann constant. According to the fitting curve showed in Fig. 8(d), the obtained activation energy is about 0.24 eV, which is close to the reported result in Pb_{0.92}La_{0.08}(Zr_{0.52}Ti_{0.48})O₃ film [42].

4. Conclusion

The energy-storage performance and cooling behavior were studied in $(\text{Pb}_{0.97}\text{La}_{0.02})(\text{Zr}_{0.73}\text{Sn}_{0.22}\text{Ti}_{0.05})\text{O}_3$ AFE thick films via orientation control. Because of its larger polarization in the thick films with (111) orientation, the best results of both energy-storage performance and ECE are simultaneously realized. A large recoverable energy-storage density of 13.5 J/cm^3 and corresponding efficiency of 73% were obtained at 900 kV/cm in the PLZST AFE thick film with (111) orientation. A giant ECE value of $\text{DT} = 28.1 \text{ }^\circ\text{C}$ was also achieved at $21 \text{ }^\circ\text{C}$ in the PLZST(111) AFE thick films during the FE-AFE switching, and the corresponding DS was $31.6 \text{ J K}^{-1} \text{ kg}^{-1}$. Moreover, a smaller leakage current density was detected in these AFE thick films, which is favored to their applications. In summary, these results indicate that AFEs are promising candidates for both energy-storage capacitors and room-temperature cooling devices, and the performance could be enhanced by the proper orientation control.

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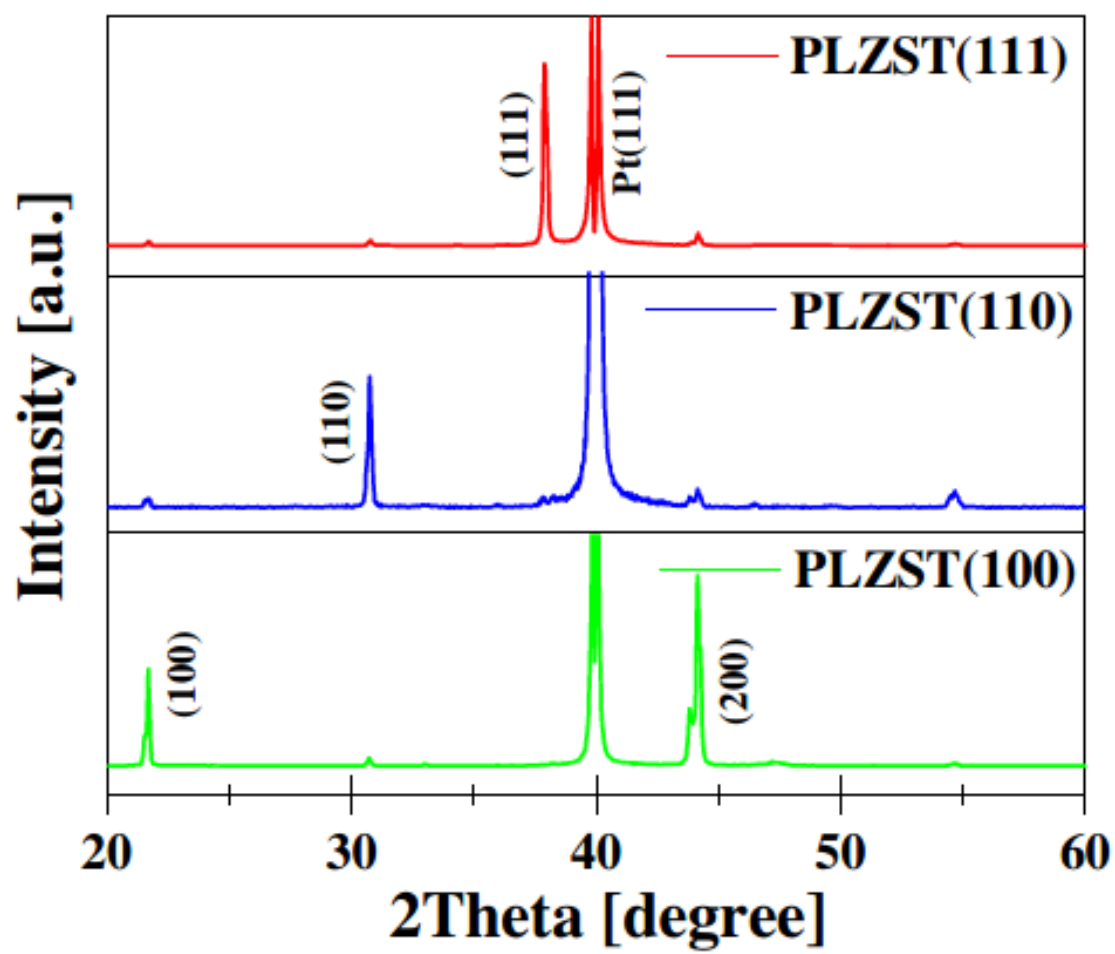


Fig. 1. XRD patterns of the PLZST(111), PLZST(110) and PLZST(100) AFE thick films.

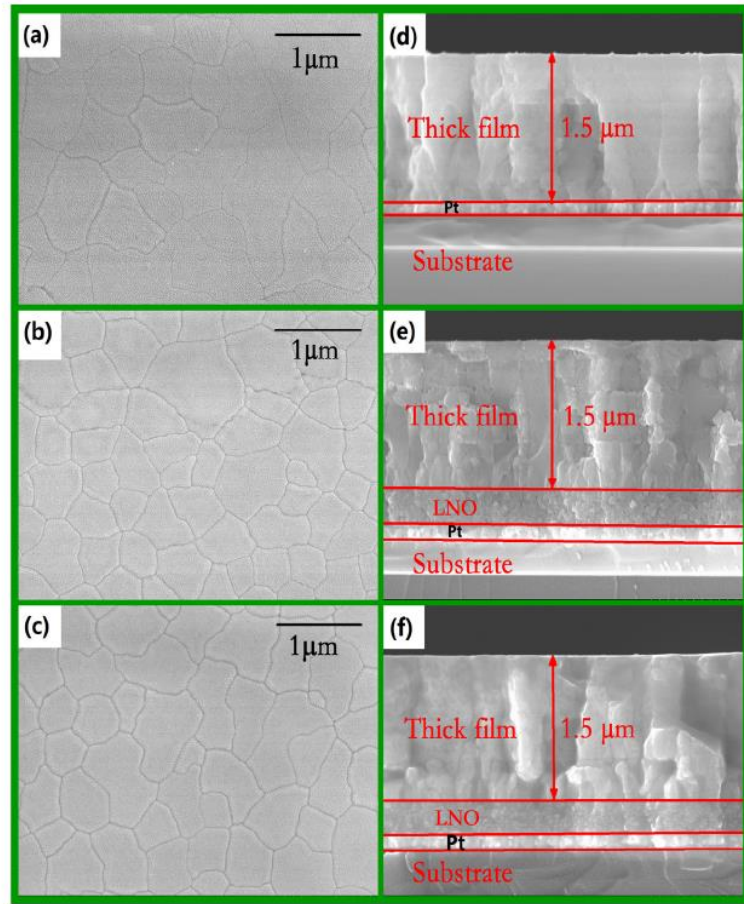


Fig. 2. Surface FE-SEM morphology images of the PLZST AFE thick films (a) PLZST(111), (b) PLZST(110), (c) PLZST(100); cross-sectional images of the films (d) PLZST(111), (e) PLZST(110), (f) PLZST(100).

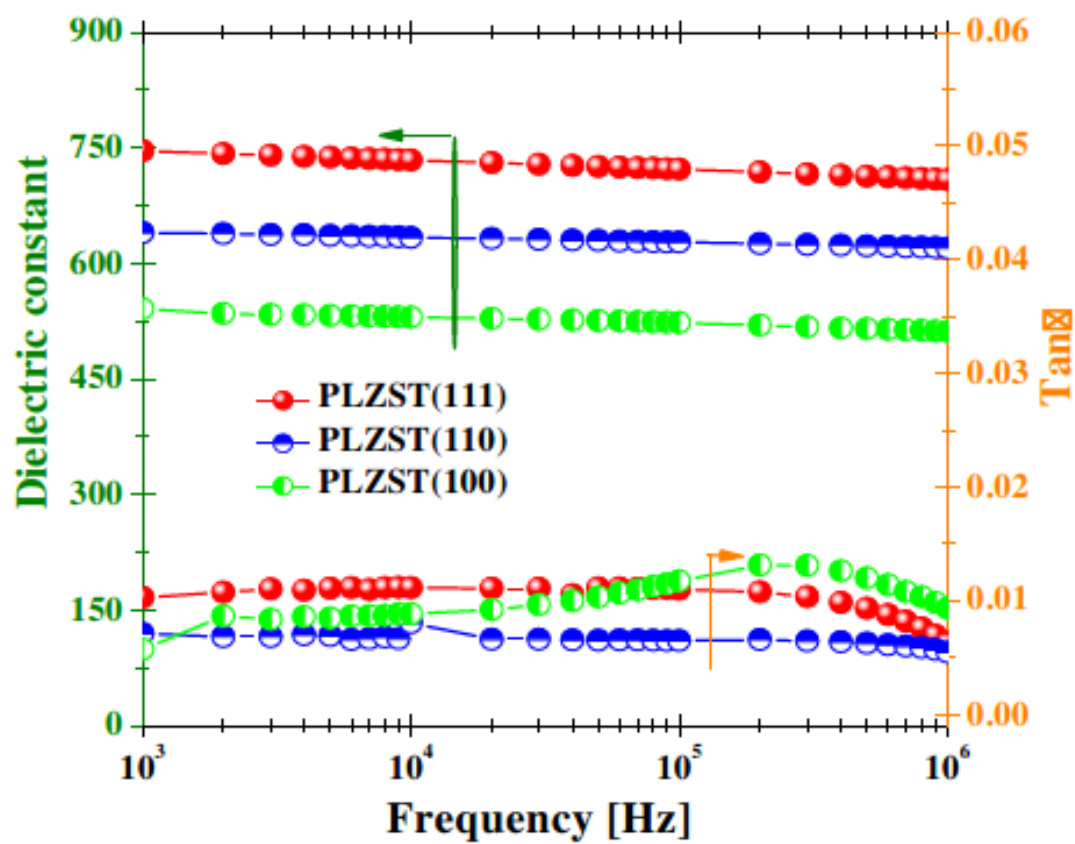


Fig. 3. Room temperature frequency-dependent dielectric constant and dielectric loss of the PLZST(111), PLZST(110) and PLZST(100) AFE thick films.

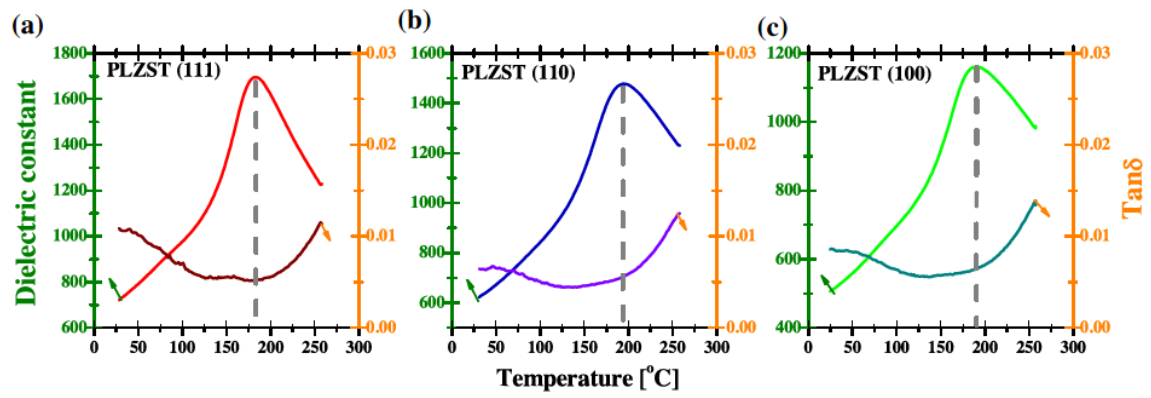


Fig. 4. Temperature dependence of dielectric constant and dielectric loss of the PLZST(111), PLZST(110) and PLZST(100) AFE thick films at 100 kHz.

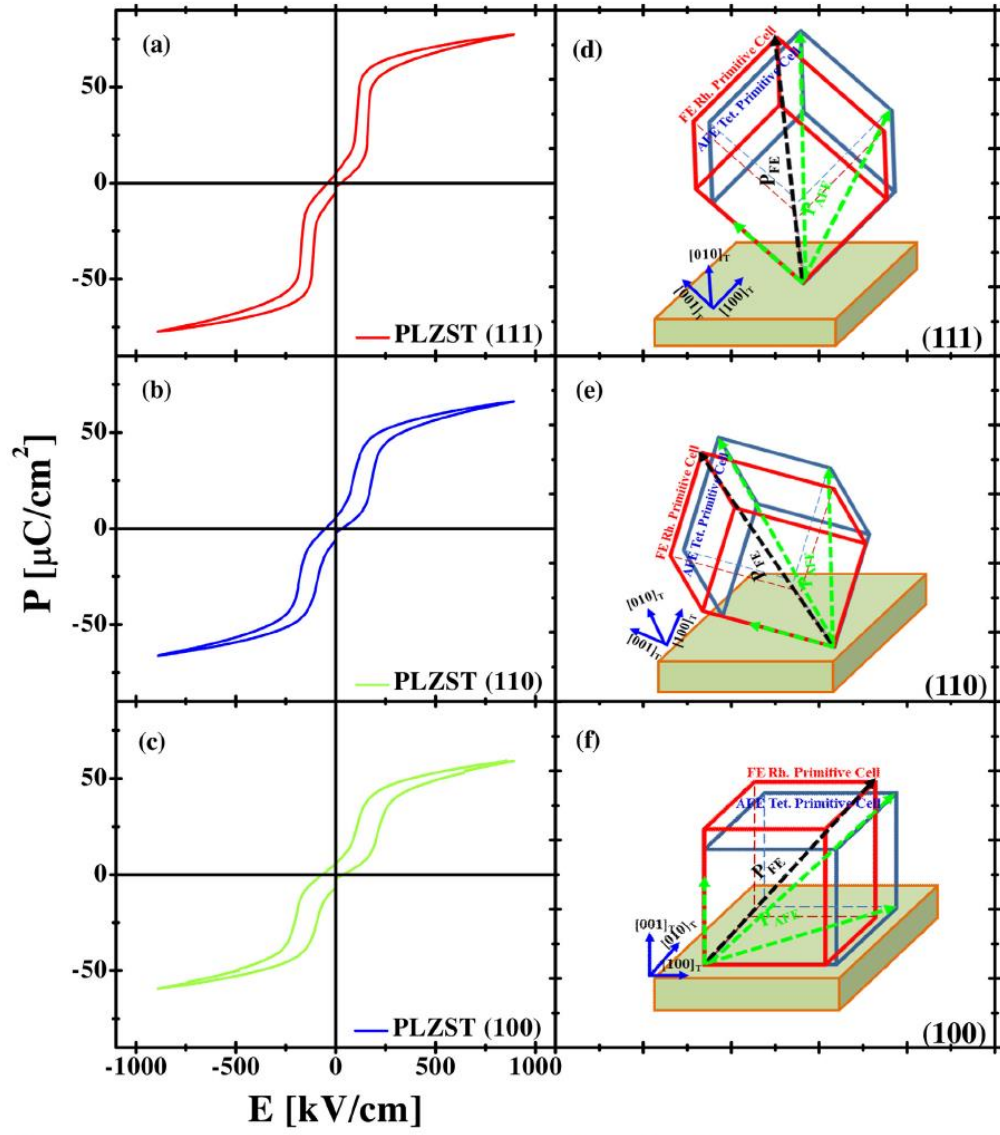


Fig. 5. (a-c) Room temperature P-E loops of the PLZST(111), PLZST(110) and PLZST(100) AFE thick films at 1 kHz. (d-f) the corresponding schematic diagram of the primitive cells.

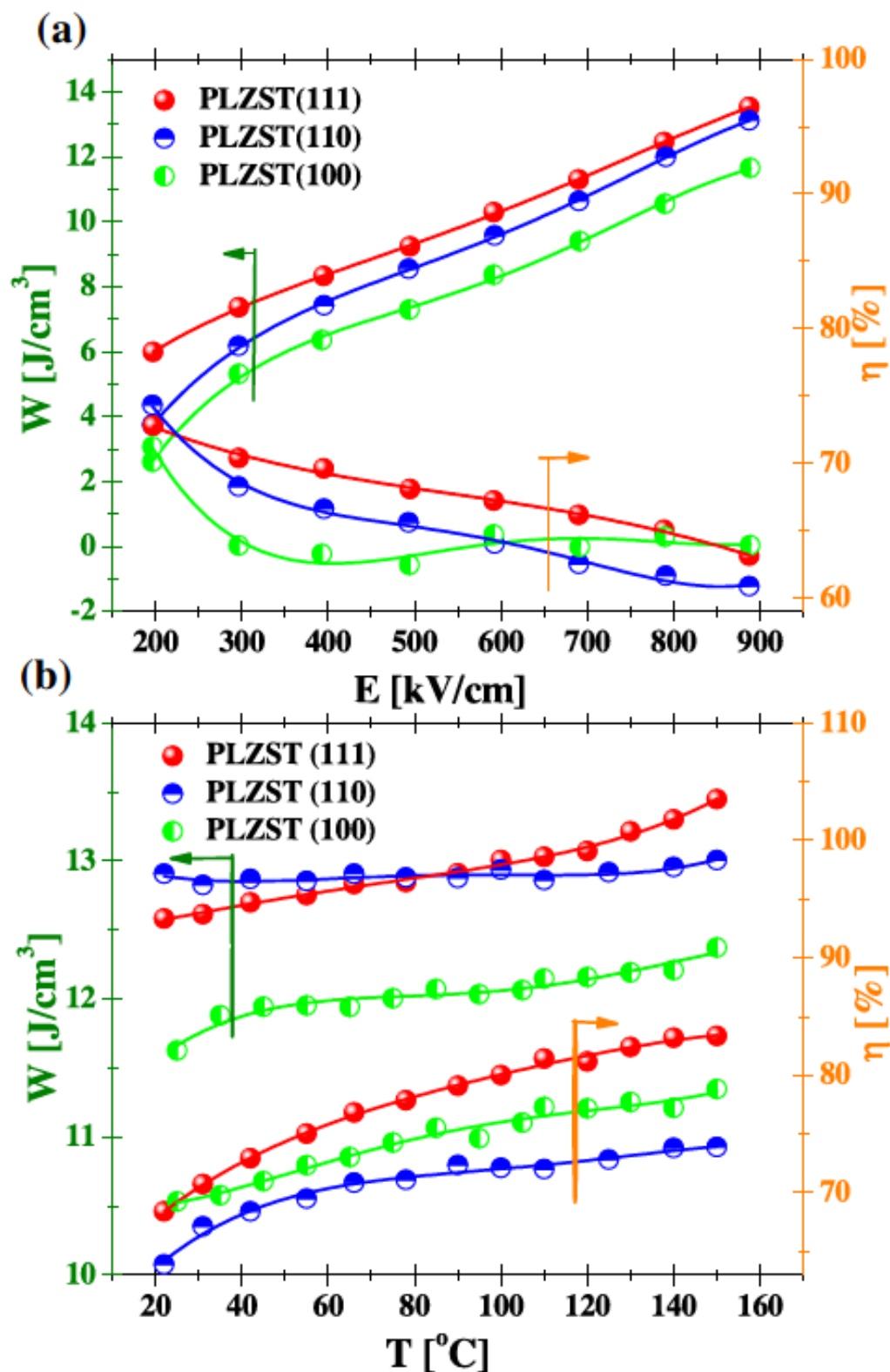


Fig. 6. (a) Electric-field dependence of recoverable energy-storage density and energy-storage efficiency of the PLZST(111), PLZST(110) and PLZST(100) AFE thick films. (b) The corresponding temperature dependence of energy-storage density and energy-storage efficiency measured at 900 kV/cm.

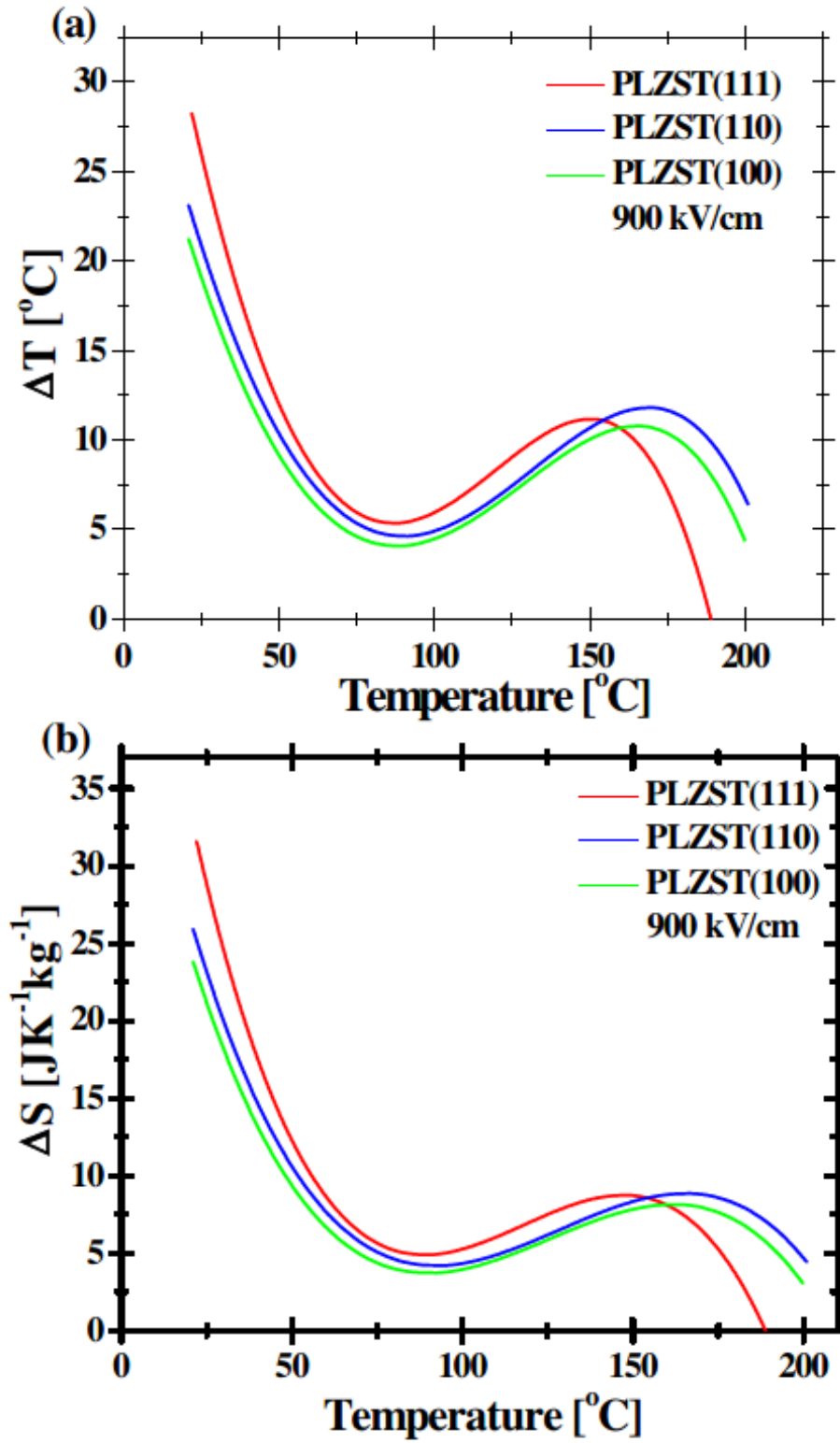


Fig. 7. (a) The adiabatic temperature changes ΔT as a function of temperature under 990 kV/cm of the PLZST(111), PLZST(110) and PLZST(100) AFE thick films, and (b) the corresponding adiabatic changes in entropy ΔS of the films.

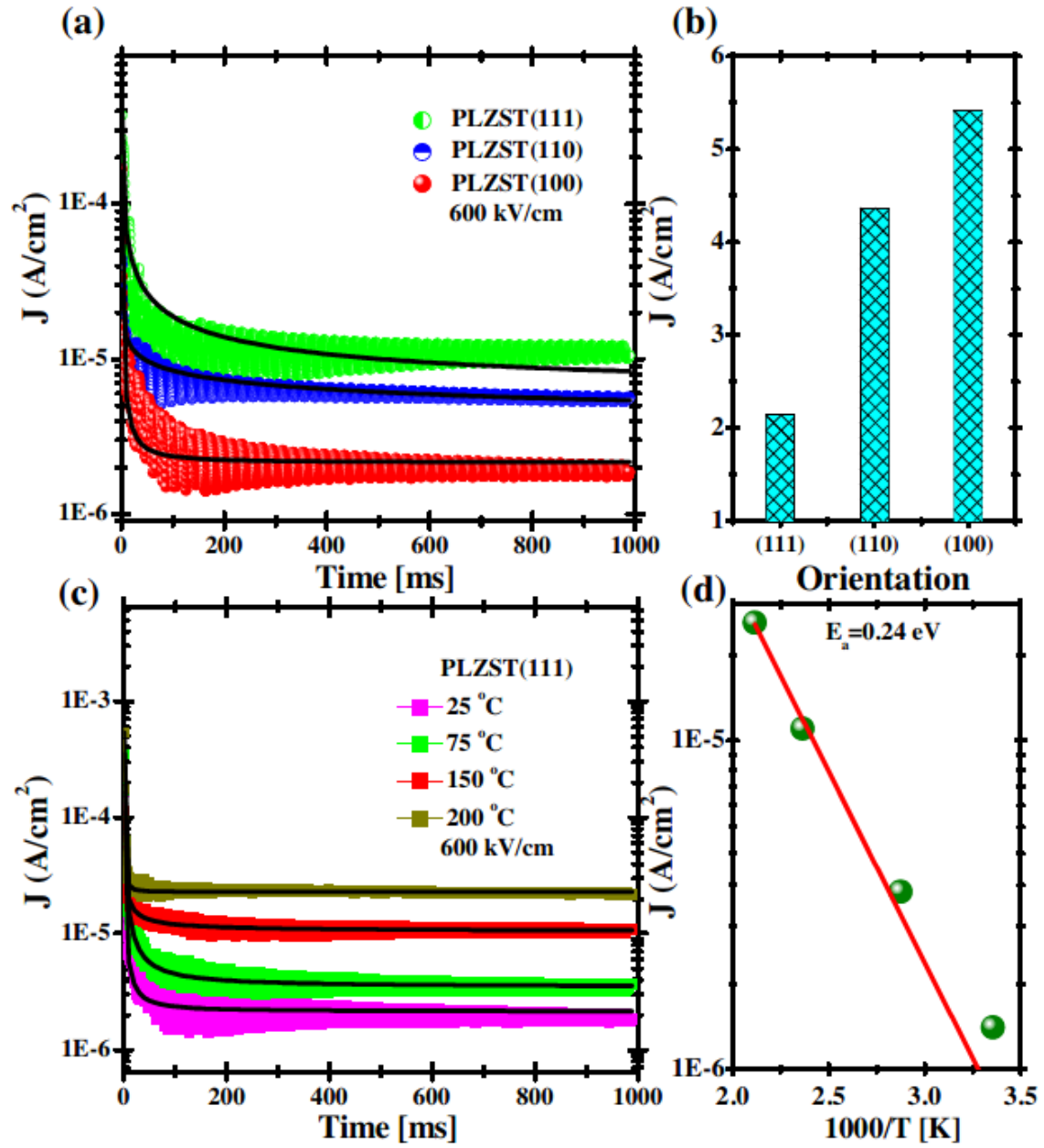


Fig. 8. (a) Dielectric relaxation current of the PLZST(111), PLZST(110) and PLZST(100) AFE thick films under 600 kV/cm and room temperature. (b) The corresponding steady-state leakage current of the films. (c) The temperature-dependent dielectric relaxation current and corresponding fitting curve of the PLZST(111) film at different temperature. (d) The corresponding steady-state current density and corresponding fitting curve as a function of reciprocal temperature.

Table 1

The parameters on ECE of PLZST thick films with different orientation at room temperature.

Orientation	COP	ξ_{\max} (K cm/kV)	ΔT (°C)	ΔS (J K ⁻¹ kg ⁻¹)
(111)	5.7	0.031	28.1	31.6
(110)	5.0	0.026	23.1	25.9
(100)	4.9	0.024	21.2	23.8

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